MINERAGRAPHIC IDENTIFICATION OF PSILO-MELANE AND MANGANITE¹

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In the course of some recent mineragraphic work on the manganiferous iron ores of the Cuyuna Range, Minnesota, undertaken by the Mississippi Valley Experiment Station of the United States Bureau of Mines in cooperation with the Missouri School of Mines and Metallurgy, Rolla, Missouri, it was necessary to make quick identification of the various oxides of iron and manganese. Hematite and psilomelane were found to be the predominating minerals, although the occurrence of manganite and pyrolusite was important to the problem in hand. The mineragraphic characteristics of these four minerals are shown in Table 1.

Pyrolusite can be readily distinguished from the other minerals on the basis of its luster, color, surface characteristics and hardness. The pitted surface of the hematite and its creamy white color usually serve to make its identification positive. Psilomelane and manganite jointly can be distinguished from the hematite and pyrolusite, but difficulty was encountered in this investigation in visually distinguishing between the two. Psilomelane pseudomorphs after manganite were often confusing. Although the hardness and reactions with HNO₃ and HCl are apparently diagnostic, the results in practice were not distinctive enough to permit certain identification at all times. The action of other etchants was therefore studied. The results are shown in Table 2.

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TABLE I
MINERAGRAPHIC CHARACTERISTICS OF SEVERAL MINERALS

C	haracteristics	Hematite	Psilomelane	Manganite	Pyrolusite
Luster		Bright	Bright, but less than hematite	Bright, between hematite and psilomelane	Bright when well polished
Color		Creamy white	Gray	Gray	Brown when not well polished; gray with good polish
Surface		Much relief,con- siderable pitting		Smooth	Rough; some specimens very difficult to polish
	Hardness	5.5 to 6.5, brittle	5 to 6, brittle	4	2 to 2.5
	Structure	Pitted, irregular grains	Varied	Radiating needles	Varied, some times banded
Etch Behavior	HNO_3	Negative	Tarnishes; rubs nearly clean	Negative	Negative
	HCl	Negative	Tarnishes; rubs faint brown	Negative	Tarnishes; rubs
	KCN	Negative	Negative	Negative	20 per cent so- lution — nega- tive; dilute — rapid tarnish
	FeCl ₃	Negative	Negative		Tarnishes dark brown; rubs pale

Table II

Etching of Hematite, Psilomelane, Manganite and Pyrolsuite

72	Effect of Reagent on				
Reagent	Hematite	Psilomelane	Manganite	Pyrolusite	
H ₂ SO ₃ (6–8 per cent SO ₂)	Negative	Darkens, with sorbitic pat- tern		Darkens rapid- ly;deeplyetched	
Equal volumes of $\mathrm{HNO_3}$, $\mathrm{H_2O_2}$, $\mathrm{H_2O}$ used cold	do	Similar to above but much more rapid	do	do	
FeSO ₄ ·(NH ₄) ₂ SO ₄ (0.3N) 10 per cent H ₂ SO ₄ used hot	do	Darkens rapidly	do	do	
HF	do	do	Accentuates cleavage cracks	do	

Because of its moderate rate of attack and characteristic etch pattern on psilomelane, sulphurous acid is recommended as a distinguishing etching reagent to those desiring to differentiate rapidly between the manganese minerals.

Figs. 1 and 2 are photographs of the same field in a polished section of a briquet containing hematite, psilomelane, manganite, and pyrolusite. The specimen in Fig. 1 was unetched, but in Fig. 2 it was etched for 90 seconds in a solution of 8 per cent sulphurous acid.

Comparison between the two figures shows that the hematite and manganite are unattacked, the psilomelane is darkened with a spotted or sorbitic pattern, and the pyrolusite is blackened. The etch readily brings out the mineral components of the mixed grains. The true nature of the grain of psilomelane pseudomorphic after manganite [Ps(a)] is also shown.

Sulphurous acid is a very convenient etching reagent for differentiating between manganese minerals, especially in ores where pseudomorphism is encountered.

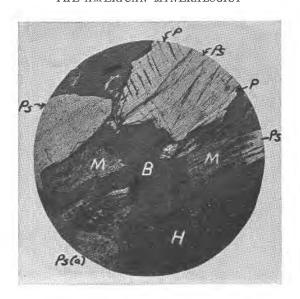


Fig. 1. Unetched. Magnification 50 x.



Fig. 2. Etched for 90 seconds in H₂SO₃.

Magnification 50 x.

Legend: B = bakelite matrix; P = pyrolusite;
Ps = psilomelane; M = manganite; H = hematite;
Ps(a) = psilomelane pseudomorphic after manganite.