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X-RAY CRYSTALLOGRAPHY OF WEGSCHEIDERITE ($\text{Na}_2\text{CO}_3 \cdot 3\text{NaHCO}_3$)¹D. E. APPLEMAN, *U. S. Geological Survey, Washington, D. C.*

The x-ray crystallography of the new mineral wegscheiderite, $\text{Na}_2\text{CO}_3 \cdot 3\text{NaHCO}_3$, has been determined by single-crystal and powder-diffraction techniques. The material used in the study came from the drill core of the Grierson Well No. 1, Green River formation (Eocene), Wyoming, and is described in the preceding paper by Fahey and Yorks. Figure 2 of that paper is a photograph of the actual sample from which the material was taken. In this sample the wegscheiderite occurs as bladed or fibrous aggregates with individual crystals as much as five cm long, without any well-developed crystal faces or cleavages. The specimen was chosen for x-ray study because it contained the most perfectly crystalline and least fibrous material available.

Small fragments, approximately $0.2 \times 0.2 \times 0.5$ mm, chipped from the specimen described above, were investigated by the Buerger precession technique, using $\text{MoK}\alpha$ and $\text{CuK}\alpha$ radiation. A study of the $hk0$, $hk1$, $h0l$ and $h1l$ reciprocal lattice nets and of the $[001]$ and $[010]$ cone-axis photographs yielded the unit-cell data given in Table 1. The optic axial angle $2V$ was measured on a universal stage in sodium light, using optic-axis interference figures.

X-ray powder-diffraction patterns were made of wegscheiderite from all the localities mentioned in the paper by Fahey and Yorks, as well as of synthetic Wegscheider's salt prepared by Yorks. All the patterns were identical. Indexed powder-diffraction data for wegscheiderite are given

TABLE 1. CRYSTALLOGRAPHIC DATA FOR WEGSCHEIDERITE
Space group: triclinic, $P\bar{1}$, (non-piezoelectric)

Direct Cell	Reciprocal Cell
$a = 10.04 \text{ \AA} \pm 0.03 \text{ \AA}$	$a^* = 0.10588 \text{ \AA}^{-1}$
$b = 15.56 \pm 0.04$	$b^* = 0.68002$
$c = 3.466 \pm 0.01$	$c^* = 0.29072$
$\alpha = 91^\circ 55' \pm 05'$	$\alpha^* = 86^\circ 00'$
$\beta = 95^\circ 49' \pm 05'$	$\beta^* = 83^\circ 12'$
$\gamma = 108^\circ 40' \pm 05'$	$\gamma^* = 71^\circ 01'$
$V = 509.1 \text{ \AA}^3 \pm 2.6 \text{ \AA}^3$	$V^* = 0.001964 \text{ \AA}^{-3}$
$a:b:c = 0.6452:1:0.2228$	

Cell contents: $2[\text{Na}_2\text{CO}_3 \cdot 3\text{NaHCO}_3]$

¹ Publication authorized by the Director, U. S. Geol. Survey.

TABLE 2. INDEXED X-RAY POWDER DIFFRACTION DATA FOR WEGSCHEIDERITE¹

<i>h</i>	<i>k</i>	<i>l</i>	<i>d</i> _{calc.} ²	<i>d</i> _{meas.}	<i>I</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>d</i> _{calc.} ²	<i>d</i> _{meas.}	<i>I</i>
0	1	0	14.71	14.7 ₈	13	2	3	0	2.954	2.95 ₄	100 <i>b</i>
-1	1	0	9.47	9.44	6	-2	0	1	2.952		
1	0	0	9.45								
0	2	0	7.35								
-1	2	0	7.01	6.99	3	0	5	0	2.941	2.91 ₄	42
1	1	0	6.98								
-1	3	0	5.08	5.05	2 <i>b</i>	0	5	0	2.942		
1	2	0	5.06								
-2	1	0	4.99	4.73	9	0	-3	1	2.913		
0	3	0	4.90								
-2	2	0	4.74	4.13	15	-3	4	0	2.903		
2	0	0	4.72								
-2	3	0	4.14	3.87 ₈	60	3	1	0	2.892		
2	1	0	4.12								
-1	4	0	3.87 ₈	3.68	25	-2	2	1	2.886		
1	3	0	3.86 ₇								
0	4	0	3.67 ₆	3.49	60	1	-3	1	2.860	2.85 ₈	6
-2	4	0	3.50 ₈								
2	2	0	3.49 ₁	3.45	2	-1	3	1	2.836	2.83 ₂	35
0	0	1	3.44 ₀								
0	-1	1	3.40 ₂	3.31	6	-2	-1	1	2.817	2.79 ₀	50
-1	0	1	3.36 ₃								
-1	1	1	3.31 ₂	3.30 ₇	15	0	3	1	2.727	2.72 ₈	30
-3	2	0	3.31 ₀								
-3	1	0	3.29 ₉	2.25	15	-1	-3	1	2.713		
0	1	1	3.29 ₉								
-1	-1	1	3.24 ₅	3.15 ₇	4	1	2	1	2.708		
0	-2	1	3.20 ₃								
1	-1	1	3.15 ₉	3.108	21	2	-1	1	2.707		
-3	3	0	3.15 ₇								
3	0	0	3.14 ₈	3.07 ₇	30	-2	3	1	2.706	2.66 ₀	42
1	0	1	3.11 ₆								
-1	2	1	3.11 ₄	3.04 ₂	9	2	0	1	2.690	2.64 ₂	60
-1	5	0	3.11 ₀								
1	4	0	3.10 ₂	3.00 ₄	5	-3	5	0	2.615	2.62 ₁	4
1	-2	1	3.06 ₃								
0	2	1	3.03 ₅	2.977	15	-2	-2	1	2.613		
-1	-2	1	3.00 ₄								
-2	1	1	2.977	2.221	15	1	-4	1	2.606		
-2	5	0	2.967								
						0	-4	1	2.604		
						3	2	0	3.604		
						2	-3	1	2.589	2.59 ₁	5
						-1	6	0	2.586		
						1	5	0	2.580		
						-1	4	1	2.542	2.54 ₅	21
						-2	6	0	2.539		
						2	4	0	2.529		
						-3	1	1	2.525	2.51 ₇	2
						-3	2	1 ³	2.504		
						-3	0	1	2.473	2.473	42 <i>b</i>
						1	3	1	2.447	2.446	30
						-3	3	1	2.416	2.397	4 <i>b</i>
						-2	-3	1	2.382		
						-4	4	0	2.368	2.372	4 <i>b</i>
						3	3	0	2.327	3.329	18 <i>b</i>
						-4	5	0	2.229	2.224	15
						4	1	0	2.221	2.224	15

¹CrK α radiation, camera diameter 114.6 mm, upper limit of *d* measurable=15.5 Å, Wilson technique.

² Calculated from cell parameters of Table 1.

³ All calculated *d* spacings listed down to *d*=2.5 Å; only values corresponding to observed lines listed below *d*=2.5 Å.

TABLE 2—Continued

<i>h</i>	<i>k</i>	<i>l</i>	$d_{\text{calc.}}^2$	$d_{\text{meas.}}$	I	<i>h</i>	<i>k</i>	<i>l</i>	$d_{\text{calc.}}^2$	$d_{\text{meas.}}$	I
-3	2	1	2.211	2.214	60	1	5	1	1.977	1.973	4
3	0	1	2.197	2.193	2	-2	-5	1	1.948	1.945	5
2	5	0	2.196			-1	8	0	1.926	1.925	6
-4	1	1	2.116			2.113	5	3	-6	1	1.914
1	-6	1	2.109	-4	7			0	1.908		
-4	3	1	2.092	4	-1			1	1.903	1.902	7
3	1	1	2.091	2.089	30	4	3	0	1.899		
-3	7	0	2.089			1.881	3 <i>b</i>				
3	4	0	2.080			2.084	30	1.848	3 <i>b</i>		
0	-6	1	2.065	2.063	5	1.809	4				
-4	0	1	2.064			1.801	4				
4	2	0	2.061			1.786	5				
-5	3	0	1.994	2.000	15	1.770	6 <i>b</i>				
-5	2	0	1.992	1.986	4 <i>b</i>	1.757	5				
-4	-1	1	1.979			1.742	7				
						1.628	42				

in Table 2. A film taken with CrK α radiation was selected for publication, since it gave the best resolution of the many doublets and triplets which occur on the diffraction pattern; however, films taken with FeK α and CuK α radiations and diffractometer patterns taken with CuK α radiation were also measured and gave similar values. The results are in general agreement with the pattern given in the ASTM X-ray Powder Data File under Sodium Carbonate Hydrate, 5Na₂O·8CO₂·3H₂O, Wegscheider's Salt (Card No. 2-0696).

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A RAPID CHANGE ROTATING SAMPLE HOLDER FOR
X-RAY DIFFRACTION OF POWDERS

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Deviations in the *x*-ray diffraction intensity from a flat powdered sample are due principally to inhomogeneities from orientation of crystallites and segregation of crystalline mixtures (Klug and Alexander, 1954; Parrish, 1956; de Wolff *et al.*, 1959). Rotation of the powder sample about an axis perpendicular to its face reduces these deviations by a