1   **Tungsten mineralization during evolution of a magmatic–hydrothermal system:**

2   **mineralogical evidence from the Xihuashan rare-metal granite in South China**

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4   *Revision 2*

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18   **ABSTRACT**

19   Tungsten deposits are usually associated with granitic intrusions that record a long

20   and complex evolution of the magmatic–hydrothermal system. However, the genetic

21   link between magmatic–hydrothermal evolution and tungsten mineralization remains

22   unclear. The Xihuashan tungsten deposit in South China, an important vein-type

23   wolframite deposit, is closely associated with greisen and multiphase intrusive activity

24   that produced biotite granite, two-mica granite, and muscovite granite. From the biotite

25   granite to the two-mica granite to the muscovite granite, micas vary from siderophyllite

26   to lithian siderophyllite, with decreasing K/Rb and Nb/Ta ratios and increasing Rb and

27   Cs contents. The zoned micas in the muscovite granite and greisen display

28   fluorine-depleted rims, reflecting subsolidus replacement by external aqueous fluids.

29   The presence of siderite indicates a Fe, Mn and CO_2-rich fluid under reducing

30   conditions. The micas in the greisen have higher F contents and lower Fe^{3+}/Fe^{2+} ratios

31   than those in the muscovite granite, suggesting that the fluids contributing to greisen

32   formation had a relatively high fluorine content and were reduced. The increase of CO_2

33   in the fluid enhanced its ability to unlock W from melts/rocks into fluids. The reducing

34   environment also facilitated the tungsten mineralization. During greisenization, the pH

35   value of the fluid increased, which destabilized the polymeric tungstates to form WO_4^{2-}.

36   The mixture of W-rich solution and Fe, Mn-rich external fluid eventually precipitated as

37   vein-type wolframite in favorable locations. An empirical equation ($Li_2O = 0.0748 \times F^2$

38   $+ 0.0893 \times F$) was introduced for estimating the Li_2O contents of hydrothermal micas

39   using the F contents determined by EPMA.

40   **Key words:** rare-metal granite; tungsten mineralization; magmatic–hydrothermal

41   evolution; mica; siderite; South China

INTRODUCTION

42   Most tungsten deposits are closely associated with granitic intrusions, and the styles

43 of mineralization may be diverse, including quartz veins, stockworks, breccia, skarn,

44 greisen, pegmatite, and porphyry. Wolframite-bearing quartz veins, regarded as an

45 important type of hydrothermal deposit, commonly occur in differentiated granitic

46 plutons that show varying degrees of hydrothermal alteration (e.g., Gleeson et al. 2001;

47 Schaltegger et al. 2005; Vigneresse 2006). The relationship between magmatic–

48 hydrothermal activities and tungsten mineralization is therefore a critical issue in

49 understanding the formation of tungsten deposits. For individual deposits, the role of the

50 intrusion as the main source of fluids and metals is usually unclear due to the multiple

51 overprint/reopening of the veins and to the lack of geochronological data constraining

52 the timing of magmatic and hydrothermal events. It has been proposed that the

53 ore-forming elements may be concentrated during multiple stages of magmatic activity

54 and then extracted by magmatic fluids (e.g., Che et al. 2013; Harlaux et al. 2018a), or be

55 leached from granitoids and/or metamorphic rocks and transported by external

56 (metamorphic or meteoric) fluids (e.g., Linnen and Williams–Jones et al. 1995; Vindel

57 et al. 1995; Vallance et al. 2001). Most previous studies on the relationship between

58 magmatic–hydrothermal episodes and tungsten mineralization have been based on the

59 geochronology of deposits and their host granites (e.g., Wang et al. 2011; Li et al. 2013),

60 experiments on partition coefficients between minerals and fluids/melts (e.g., Linnen

61 and Cuney 2005; Che et al. 2013), analyses of fluid inclusions (e.g., Wei et al. 2012),

62 and analyses of mineral geochemistry and stable isotopes (Legros et al. 2019). Although

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64   recent studies have successfully dated wolframite from tungsten deposits (e.g., Harlaux

65   et al. 2018b; Deng et al. 2019), limitations still exist, owing to the lack of suitable

66   techniques for precisely dating both hydrothermal deposits and associated granitoids,

67   the intrinsic differences between modeled and natural granitic systems, and

68   uncertainties on whether fluid inclusions are representative of ore-forming fluids.

69   The textures and chemical compositions of ore-bearing oxide minerals,

70   rock-forming minerals, and even accessory minerals may be used to track magmatic–

71   hydrothermal processes in granitic systems and related metal mineralization (e.g.,

72   Roda–Robles et al. 2007; Van Lichtervelde et al. 2008; Yang et al. 2013; Li et al. 2015;

73   Yin et al. 2019). Micas are ubiquitous minerals in peraluminous granitic systems, and

74   these primary magmatic or secondary minerals (due to fluid interactions) have the

75   potential to record the processes of fractional crystallization and hydrothermal evolution.

76   As the major rock-forming mineral that hosts tungsten (e.g., 1.9−198 ppm in micas from

77   peraluminous granites; Simons et al. 2017; El Korh et al. 2020), micas provide an

78   effective proxy for tracing the progressive enrichment and ore-forming processes during

79   magmatic–hydrothermal evolution (Legros et al. 2016, 2018; Yin et al. 2019). In

80   addition to CO_2-rich fluid inclusions, carbonates (including synchysite, siderite, and

81   calcite) may occur frequently in rocks associated with tungsten deposits, suggesting that

82   the fluids involved in W-mineralization are enriched in CO_2 (Castorina and Masi 2008;

83   Wei et al. 2012).

84   The widespread rare-metal deposits in South China are closely associated with

85   Mesozoic granitic rocks, and an example is the Nanling W−Sn polymetallic

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86   mineralization region (Fig. 1a) (Yin et al. 1995; Wang et al. 2003; Li et al. 2015). The

87   Xihuashan tungsten deposit located in the Jiangxi Province is a typical vein-type

88   wolframite deposit with minor Sn–Be mineralization (Nie and Wang 2007), and it is one

89   of the major tungsten deposits in the world (Sinclair et al. 2014). The Xihuashan deposit

90   is spatially and temporally associated with Jurassic granitic intrusions (Wang et al. 2011;

91   Li et al. 2013; Yang et al. 2018), and it shows genetic links with the magmatic and

92   hydrothermal evolution of the granitic intrusions. This deposit therefore provides an

93   opportunity to investigate the mechanism of W enrichment and mineralization in

94   rare-metal granites during the evolution of the magmatic–hydrothermal system.

95   Previous studies on the Xihuashan deposit were focused mainly on the geochronology

96   of the deposits and related granites (Wang et al. 2011; Guo et al. 2012; Hu et al. 2012;

97   Li et al. 2013), fluid inclusions and wolframite formation (Wei et al. 2012; Huang et al.

98   2013), and magmatic fractionation or hydrothermal alteration of the granites (Le Bel et

99   al. 1983; Maruéjol et al. 1990; Wang et al. 2003; Yang et al. 2013). This paper presents

100   the results of whole-rock major and trace element analyses of the Xihuashan rare-metal

101   granite as well as the chemical compositions of the micas and siderite, to elucidate the

102   relationships between the tungsten mineralization and the magmatic–hydrothermal

103   evolution of highly fractionated granitic melts, and to reveal the relative contributions of

104   magmatic and hydrothermal processes to the formation of tungsten ore deposits. The

105   results show the critical role of greisenizing fluids and magma fractionation in the

106   tungsten mineralization, and provide new insights into the metallogenesis of tungsten in

107   rare-metal granites.

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GEOLOGICAL SETTING AND PETROGRAPHY

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110   The South China Block (SCB) comprises the Yangtze Block in the northwest and

111 the Cathaysia Block in the southeast (Fig. 1a). The two blocks were amalgamated

112 during the early Neoproterozoic (e.g., Li et al. 2009; Zhao et al. 2011). During the

113 Phanerozoic, the Cathaysia Block was subjected to intensive and widespread multiphase

114 orogenesis and magmatism (Li et al. 2012). The igneous rocks of Jurassic and

115 Cretaceous age are predominantly granites and rhyolites with subordinate mafic and

116 rare intermediate lithologies (Zhou et al. 2006; Li et al. 2012). The Jurassic rocks occur

117 over large parts of the Cathaysia Block, while the Cretaceous rocks cover the coastal

118 area of SE China and the middle to lower reaches of the Yangtze River in the Yangtze

119 Block (Fig. 1a). The majority of highly evolved and peraluminous granitic rocks were

120 formed during the Jurassic with a few during the Cretaceous (Legros et al. 2020). These

121 rocks in the Nanling area are associated with W–Sn–Mo–Bi–Be–Nb–Ta mineralization,

122 and the area is well known as the Nanling tungsten–tin polymetallic mineralization

123 region (NPMR, Fig. 1a) (e.g., Wang et al. 1982; Pei and Hong 1995).

124   The Xihuashan mining district is located in the southwestern part of the Jiangxi

125 Province, South China (Fig. 1). It contains more than 700 economically viable ore veins,

126 including W–Sn–Be mineralization (Nie and Wang 2007). There are six main tungsten

127 mines in the district, namely Xihuashan, Dangping, Shenlongkou, Luokeng,

128 Xialuogushan, and Niuzishi (Fig. 1b). The Xihuashan Mine hosts the largest vein-type

129 wolframite deposit in the district, with the total estimate of WO_3 being 81,300 tons and

130   an average ore grade of 1.08 % WO_3 (Wei et al. 2012). The deposit is spatially and

131   temporally related to a multiphase granitic intrusion that crops out over an area of about

132   20 km^2 (Guo et al. 2012). Previous geochronological studies indicated a close temporal

133   relationship between the granitic magmatism and the W mineralization. For example,

134   the crystallization ages of the major intrusions are 156–161 Ma (Guo et al. 2012; Li et al.

135   2013; Tang et al. 2020), and the Re–Os ages of molybdenites in the ore are identical

136   (157 ± 3 Ma; Wang et al. 2011). The Xihuashan granitic complex was emplaced in

137   upper Cambrian low-grade metamorphosed flysch-like sedimentary rocks, which

138   underwent contact metamorphism (Le Bel et al. 1984). Wei et al. (2012) suggested the

139   ore-forming fluids to be a mixture of magmatic and hydrothermal fluids derived from

140   the granitic melts and meteoric water. With the knowledge obtained from geological

141   mapping and exploratory drilling, the Xihuashan granitic complex may be divided into

142   four successive intrusions (G1–G4; Fig. 1b) (Maruéjol et al. 1990; Wang et al. 2003).

143   The G1 (coarse-grained biotite granite) occurs as the outermost part of the complex. The

144   G2 (medium-grained biotite granite) hosts tungsten mineralization at the Xihuashan

145   Mine. The G3 (medium- to fine-grained biotite granite) hosts tungsten mineralization at

146   the Dangping Mine, whereas the G4 (fine-grained biotite granite) occurs as a barren

147   border of G3.

148   The collected samples of G2 (medium-grained biotite granite) in the Xihuashan

149   Mine include biotite granite, two-mica (lithian siderophyllite–muscovite) granite,

150   muscovite granite, and greisen (Fig. 2a–d). The biotite granite crops out over an area of

151   about 5.2 km^2. The greisen, two-mica granite and muscovite granite are only exposed in

152   the sub-surface and were investigated using drilled samples and exposure in the mine

153   tunnels. Liu (2005) and Huang et al. (2006) showed that the greisen was usually

154   associated with tungsten ore veins cutting the fine-grained two-mica granite, the

155   muscovite granite, or the country rocks. There is a transition between the two-mica

156   granite and the muscovite granite, both of which intruded the biotite granite (Liu 2005;

157   Huang et al. 2006). The main accessory minerals of the Xihuashan intrusive rocks are

158   fluorite, apatite, zircon, thorite, xenotime, and rutile (Supplementary Table 1).

159   The *biotite granite* consists of quartz (32–38%), K-feldspar (19–20%), plagioclase

160   (32–34%), and siderophyllite (8–13%) (Fig. 2a). Plagioclase is subhedral or euhedral

161   and shows compositional zoning. The siderophyllite grains are coarse and up to 2 mm in

162   diameter. The plagioclase and siderophyllite alway host inclusions of ilmenite and

163   apatite. Rutile is common as a secondary mineral associated with chloritization of the

164   micas. The fluorite grains are anhedral and contain high yttrium contents (Y_2O_3 = 8–15

165   wt%). The accessory minerals enriched in REEs include apatite, monazite, allanite, and

166   zircon.

167   The *two-mica (lithian siderophyllite–muscovite) granite* is composed mainly of

168   quartz (30–35%), K-feldspar (14–15%), plagioclase (43–49%), and micas (6–8%) (Fig.

169   2b). The micas are dominantly lithian siderophyllite (~6%) with lesser amounts of

170   muscovite (~2%). The primary mica crystals are coarse and up to 2 mm in size, while

171   most secondary micas are fine-grained sericite within large plagioclase and K-feldspar

172   phenocrysts. The lithian siderophyllite always contains inclusions of ilmenite and

173   apatite, and the rims always show porosity due to dissolution. Xenotime is the principal

174   phosphate mineral in the two-mica granite, and other common accessory minerals are

175   zircon, garnet, and fluorite.

176   The *muscovite granite* consists of quartz (39–45%), K-feldspar (19–22%),

177   plagioclase (32–35%), and muscovite (~4%) (Fig. 2c). The accessory minerals include

178   spessartine, zircon, fluorite, xenotime, and siderite. The siderite crystals are anhedral

179   and elongate, and they occur mostly along the cleavages of the micas.

180   The *greisen* is composed of quartz (49–55%), lithian siderophyllite (30–35%), and

181   muscovite (9–11%) (Fig. 2d). The principal accessory minerals include zircon, xenotime,

182   fluorite, siderite, niobotantalates, and wolframite. Xenotime and zircon are the

183   predominant REE-rich minerals, while niobotantalates and wolframite are the

184   tungsten-bearing minerals.

185

186   **ANALYTICAL METHODS**

187   The whole rock geochemical analyses were carried out at the State Key Laboratory

188   of Isotope Geochemistry (SKLaBIG), Guangzhou Institute of Geochemistry (GIG),

189   Chinese Academy of Sciences (CAS). To eliminate potential contamination, samples

190   were cut into slabs and the central parts (> 200 g) were selected for whole-rock analyses.

191   The rocks were crushed into pieces with size of < 0.5 cm in diameter, cleaned with

192   deionized water in ultrasonic bath, and then pulverized in a corundum mill. Major

193   element oxides were determined using a Rigaku RIX 2000 X-ray fluorescence (XRF)

194   spectrometer. Samples were prepared as glass discs by a Rigaku desktop fusion machine,

195   mixing 0.50 g of rock powder (dried at 110 °C) with 4.0 g of $Li_2B_4O_7$ for 15 min at

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196   1100 °C in 95% Pt - 5% Au crucibles. Calibration lines used for quantification were

197   produced by bivariate regression of data from 36 reference materials. Matrix corrections

198   were also incorporated into calibration using the empirical Traill-Lachance procedure.

199   For whole rock analyses, total Fe is reported as Fe_2O_3 (Supplementary Table 2). The

200   analytical uncertainty was between 1 % and 5 %. The additional quantitative analysis of

201   F was conducted by ALS Chemex (Guangzhou) Co Ltd., China, using the methods of

202   KOH fusion and ion selective electrode, or Na_2O_2 fusion, citric acid leach and ion

203   selective electrode. Analytical precision of F was lower than 10%.

204   Trace elements were determined using inductively coupled plasma-mass

205   spectrometry (ICP-MS). The rock powder was digested in a mixture of HF and HNO_3 in

206   high-pressure Teflon bombs. An internal standard solution containing single element Rh

207   was used to monitor signal drift. The detailed procedure was described in Li et al.

208   (2006). The US Geological Survey and Chinese National Standards, including SARM-4,

209   W-2, BHVO-2, AGV-2, GSR-1, GSR-2 and GSR-3, were used for calibration.

210   Analytical precision of REE and other incompatible elements was in the range of 1 to

211   5 %.

212   The siderites were examined using HORIBA Xplora laser Raman

213   microspectroscopy at the Key Laboratory of Mineralogy and Metallogeny (KLMM),

214   GIG-CAS. An Ar ion laser operated at 44 mW was used to produce an excitation line

215   with wavelength of 532 nm. The scanning range of spectra was set between 100 and

216   4000 cm^{-1} with an accumulation time of 10 s for each scan. The spectral resolution was

217   0.65 cm^{-1}. The Raman shift of a monocrystalline silicon piece was measured to be 520.7

218   cm^{-1} before analyses.

219   Back-scattered-electron (BSE) images were obtained by a Zeiss Supra 55

220   field-emission scanning electron microscope (FE-SEM) at the SKLaBIG-GIG-CAS.

221   The *in-situ* mineral composition analyses were performed using the JEOL JXA8100

222   electron probe micro-analyzer (EPMA) at the SKLaBIG-GIG-CAS, with the operating

223   conditions of 15 kV accelerating voltage, 20 nA beam current and 1 μm beam size

224   (siderite) or 5 μm beam size (micas). A variable peak counting time of 7–40 s was

225   designed for the intensity of the characteristic X-ray lines and desired precision

226   (Supplementary Table 3). The raw data were corrected using ZAF correction. For micas,

227   the detection limits for most elements are lower than 251 ppm, and those for F and Rb

228   are 568 ppm and 631 ppm, respectively (Supplementary Table 3). Relative uncertainties

229   were estimated to be ca. 7% for F and 10–15 % (1σ) for Cl. Chemical formulae of micas

230   were calculated from mineral analyses based on 24 anions (O, F, OH), and Fe^{3+} was

231   estimated following the method by Lin and Peng (1994) described in Supplementary

232   Table 4. The CO_2 content in siderite was calculated by stoichiometry.

233   *In-situ* trace element concentrations were measured in mica and siderite grains

234   using an Agilent 7500a quadrupole ICP-MS coupled with a RESOlution M-50 laser

235   ablation system at the KLMM-GIG-CAS. The operating conditions included a spot size

236   of 42 μm, repetition rate of 5 Hz, maximum energy of 90 mJ and laser fluence of 7

237   J/cm^2. Argon was used as a make-up gas and mixed with helium as the carrier gas via a

238   cyclone mixer prior to entering the ICP torch. The National Institute of Standards and

239   Technology (NIST) 610 glass and 612 glass were used as external standards for quality

240   control (Jochum et al. 2011). Every analysis of six samples was followed by NIST 610

241   and 612 standards for time-dependent calibration of sensitivity drift. The internal

242   standards are ^{27}Al and ^{57}Fe for mica and siderite, respectively. The off-line data

243   processing was performed using the commercial software ICPMSDataCal 6.7 (Liu et al.

244   2008). The dwell time of each spot is 70 s of analysis time (30 s background, 40 s

245   ablation). Limits of detection (LOD) were calculated using the 3σ criterion detailed by

246   Longerich et al. (1996). The relative standard deviation (RSD) varies from 1% to 10%.

247   More details on the standards are reported in Supplementary Table 5.

248

249   **WHOLE-ROCK COMPOSITIONS**

250   Eight samples (two of each rock type) were analyzed for their whole rock major

251   and trace element compositions (Supplementary Table 2). The granite samples are all

252   peraluminous (A/CNK = 1.02–1.10; Supplementary Table 2), with high SiO_2 (72.50–

253   82.54 wt%), F (0.08–0.67 wt%), and alkali contents (Na_2O + K_2O = 6.36–8.78 wt%) but

254   low contents of MgO (0.02–0.38 wt%), Fe_2O_3 (0.44–2.02 wt%), and TiO_2 (<0.18 wt%).

255   From the biotite granite to the two-mica granite to the muscovite granite, the TiO_2, MgO,

256   P_2O_5, CaO, and F contents decrease gradually with decreasing Fe_2O_3 contents (Fig. 3),

257   consistent with previous data reported by Guo et al. (2012). The two greisen samples

258   display higher contents of SiO_2 (79.81–81.42 wt%) but lower contents of Al_2O_3 (8.98–

259   9.74 wt%), MgO (0.06–0.09 wt%), MnO (0.15–0.19 wt%), TiO_2 (0.02–0.03 wt%), CaO

260   (0.21–1.64 wt%), and alkalis (Na_2O + K_2O = 3.37–3.38 wt%) than the granite samples

261   (Supplementary Table 2; Fig. 3). The greisen samples have high Fe_2O_3 (3.28–3.36 wt%)

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262   and F (1.05–1.22 wt%) contents (Fig. 3), due to abundant lithian siderophyllite,

263   wolframite, and fluorite.

264   The granite and greisen samples have a narrow range of REE concentrations

265   (80.4–129 ppm; Supplementary Table 2) and similar chondrite-normalized REE patterns

266   with strong negative Eu anomalies (Eu/Eu* = 0.01–0.31) and a weak lanthanide tetrad

267   effect, consistent with previous data reported by Guo et al. (2012) (Fig. 4a). The

268    $[La/Yb]_N$ and Eu/Eu* values decrease following the sequence of biotite granite,

269   two-mica granite, muscovite granite, and greisen. The muscovite granite and greisen are

270   depleted in LREEs and Eu, but enriched in HREEs, and this differs from the typical

271   "wing-shaped" REE patterns for highly evolved granites (Irber 1999; Li et al. 2015). In

272   the multi-element diagrams normalized to the mean composition of upper continental

273   crust, all the samples are enriched in Cs (11.1–73.5 ppm), Rb (312–1020 ppm), W

274   (4.82–65.3 ppm), Ta (3.35–27.2 ppm), Sn (5.65–198 ppm), and Li (37.8–987 ppm), but

275   depleted in Ba (3.31–266 ppm), Sr (1.31–90.6 ppm), P (9.29–204 ppm), and Ti (12.1–

276   1069 ppm) (Fig. 4b). Overall, the muscovite granite and greisen have much higher Li,

277   Cs, Rb, and Sn abundances but lower Ba, Sr, and P than the biotite granite and two-mica

278   granite.

279

280   **MINERAL COMPOSITIONS**

281   **Mica**

282   Most micas in the biotite granite and two-mica granite are homogeneous in their

283   chemical compositions, but those in the muscovite granite and greisen show

14

284 compositional zoning with distinct core and rim on BSE images or have a porous rim

285 due to dissolution (Fig. 5). A total of 267 EPMA analyses were performed on eight

286 samples, along with 76 LA–ICP–MS analyses on the same spots previously analyzed by

287 EPMA. Altogether, these analyses cover all the rock types in the pluton and different

288 areas of compositionally zoned mica grains (Supplementary Table 6).

289 The micas in the Xihuashan pluton show a wide range of elemental compositions.

290 Siderophyllite in the biotite granite has the highest contents of FeO^{Tot} (21.70–24.57 wt%,

291 2.757–3.253 apfu), TiO_2 (0.53–3.20 wt%, 0.061–0.373 apfu), and MgO (3.96–5.12 wt%,

292 0.925–1.165 apfu) of all the micas, but the lowest contents of SiO_2 (34.44–37.00 wt%,

293 5.469–5.667 apfu) and Al_2O_3 (16.34–20.77 wt%, 2.333–2.531 apfu) (Fig. 6;

294 Supplementary Table 6). In the two-mica granite, the lithian siderophyllite has higher

295 contents of FeO^{Tot} (19.40–23.17 wt%, 2.439–3.002 apfu) and TiO_2 (0.03–1.57 wt%,

296 0.004–0.183 apfu) than the muscovite (FeO^{Tot} = 5.00–8.85 wt%, 0.576–1.051 apfu;

297 TiO_2 = 0.01–0.40 wt%, 0.001–0.042 apfu). Micas in the muscovite granite and greisen

298 have lower contents of FeO^{Tot} (3.43–19.97 wt%, 0.388–2.504 apfu), TiO_2 (below the

299 detection limit to 0.51 wt%, bdl.–0.058 apfu), and MgO (0.04–0.70 wt%, 0.008–0.147

300 apfu), but higher contents of SiO_2 (37.77–47.99 wt%, 5.683–6.529 apfu) than those in

301 the biotite granite and two-mica granite (Fig. 6). Micas in the biotite granite, two-mica

302 granite, and muscovite granite have similar F contents, all of which are lower than the

303 value of the cores of greisen micas (Fig. 6). Micas in all samples have extremely low Cl

304 contents (mostly less than 0.06 wt%), which decrease roughly from siderophyllite,

305 lithian siderophyllite, and muscovite along with a negative correlation with SiO_2

306   contents (Fig. 6f).

307   Micas in the biotite granite have much higher V, Sc, Cr, Co, and Ni contents than

308   the micas in other rocks (Fig. 7a; Supplementary Table 6). Overall, the micas in the

309   Xihuashan pluton have high contents of Rb, Cs, Nb, and Ta (typically Rb > 1000 ppm,

310   Cs > 200 ppm, Nb > 20 ppm, Ta > 10 ppm) but the contents of most REEs are below

311   detection limits (Supplementary Table 6). The lithian siderophyllites in the two-mica

312   granite and the cores of micas in the muscovite granite and greisen have higher Rb and

313   Cs contents than siderophyllites in the biotite granite (Fig. 7b, c), while the Nb and Ta

314   contents of micas increase from the biotite granite to the two-mica granite, but decrease

315   slightly in the greisen (Fig. 7d, e). Micas in the muscovite granite and greisen have

316   lower Nb/Ta ratios (0.84–5.63) than siderophyllites in the biotite granite and lithian

317   siderophyllites in the two-mica granite (Nb/Ta = 1.77–17.15) (Fig. 7f). The Sn and W

318   contents of micas in the analyzed samples increase gradually from the biotite granite to

319   the two-mica granite and to the cores of micas in the muscovite granite.

320   The micas in the muscovite granite and greisen generally exhibit core–rim

321   zonation on BSE images (Fig. 5). The cores of zoned micas in the muscovite granite and

322   greisen commonly have lower SiO_2 (37.77–42.47 wt%, 5.683–6.333 apfu) and Al_2O_3

323   (23.33–26.91 wt%, 1.667–2.317 apfu) contents but higher FeO^{Tot} (11.32–19.97 wt%,

324   1.356–2.504 apfu), Li_2O (0.26–1.74 wt%, 0.141–1.023 apfu), F (0.73–4.26 wt%, 0.342–

325   1.981 apfu), Rb (4753–5860 ppm), Cs (377–1691 ppm), Ta (15.8–171 ppm), and Nb

326   (33.9–195 ppm) contents than the rims (Figs 5, 6e, 7b–e; Supplementary Table 6). The

327   cores of zoned micas in the muscovite granite and greisen have lower Nb/Ta ratios

328   (0.97–1.55 and 0.84–2.14, respectively) than the rims (1.29–5.63 and 2.01–3.40) (Fig.

329   7f). In addition, the mica cores in the muscovite granite have higher W (70–84 ppm) and

330   Sn (694–848 ppm) contents than the rims (14.9–63 ppm and 265–694 ppm,

331   respectively). The zoned micas in the greisen samples exhibit an increasing trend of W

332   and Sn from core to rim, with values higher than the rims of zoned micas in the

333   muscovite granite (Fig. 7g, h).

334

335   **Siderite**

336   All the siderites in the muscovite granite and greisen samples occur along the

337   cleavages or the crystal borders of zoned micas (Fig. 5), which were most likely formed

338   as secondary minerals due to the presence of CO_2-rich hydrothermal fluids (Buckley

339   and Woolley 1990). The siderites were identified using Raman spectroscopy and EPMA.

340   The Raman spectrum of the investigated mineral shows a strong intensity band at

341   1081.73 cm^{-1} and a low intensity band at 276.97 cm^{-1} (Fig. 8), corresponding to the

342   symmetric stretching and external vibration of the CO_3 group, respectively, which are

343   consistent with the characterization of carbonates reported by Buzgar and Apopei

344   (2009). Minor shifts in position may be due to the effect of natural impurities in the

345   studied samples. The siderites have high contents of FeO (30.14–56.01 wt%), MnO

346   (3.47–20.39 wt%), and CO_2 (35.29–37.60 wt%), but low contents of MgO (<0.22 wt%)

347   and alkalis ($Na_2O + K_2O < 1$ wt%) (Supplementary Table 7). The siderites in the

348   muscovite granite have higher REE contents (ΣREE = 62–109 ppm) than those in the

349   greisen (ΣREE = 6.92–34.4 ppm; Supplementary Table 8). They show chondrite

17

350   normalized LREE-depleted and HREE-enriched pattern with very low values of

351   $[La/Yb]_N$ (0.01–0.05) (Fig. 8c; Supplementary Table 8) and strongly negative Eu

352   anomalies (Eu/Eu* < 0.32; Fig. 8c; Supplementary Table 8). On the multi-element

353   diagram normalized to average upper continental crust, the siderites are enriched in Li,

354   Rb, Cs, Ta, W, and Sn, but depleted in Ba, Zr, and Ti, similar to the whole rock

355   composition (Fig. 8d).

356

357   **DISCUSSION**

358   **Estimation of lithium contents in micas**

359   The Li_2O content of mica may be measured directly by LA–ICP–MS or estimated

360   by EPMA from empirical equations based on the crystallochemical relationship between

361   Li and other major elements in the mica. Considering the different beam sizes of the

362   LA–ICP–MS (e.g., 42 μm) and EPMA (e.g., 1 μm) methods, the Li_2O content of mica

363   estimated from the EPMA results using empirical equations may provide a better

364   constraint on the compositional variation at the micrometer scale.

365   The micas in the Xihuashan pluton display two trends on the diagram of SiO_2

366   versus Li_2O with contents measured by LA–ICP–MS (Fig. 9a). Siderophyllites in the

367   biotite granite and lithian siderophyllites in the two-mica granite have relatively low

368   SiO_2 contents that show a positive correlation with Li_2O (Fig. 9a), which is consistent

369   with the regression function between Li_2O and SiO_2 (Fig. 9a) for trioctahedral micas

370   (Tischendorf et al. 1997):

371          $$Li_2O = 0.289 \times SiO_2 - 9.658 \qquad\qquad (1)$$

18

372   In contrast, muscovites in the two-mica granite and micas in the muscovite granite and

373   greisen have relatively higher SiO_2 contents that show a negative correlation with Li_2O

374   contents (Fig. 9a). In particular, muscovites in the two-mica granite and the cores of

375   micas in the muscovite granite show good correlation with the regression function

376   between Li_2O and F (Fig. 9b) for dioctahedral micas (Tischendorf et al. 1997):

377        $$Li_2O = 0.3935 \times F^{1.326} \qquad\qquad\qquad (2)$$

378   while micas in the greisen samples clearly deviate from this function (Fig. 9b). The Li

379   contents of hydrothermal micas (e.g., micas in the greisen) appear to be overestimated

380   using Eq. (2) for dioctahedral micas (Fig. 9b). While the equations in Tischendorf et al.

381   (1997) were designed for magmatic systems, additional adjustment is required for

382   application to hydrothermal systems (Legros et al. 2016). In this paper, a new empirical

383   equation for the Li_2O and F contents of hydrothermal micas is introduced (Fig. 9c):

384        $$Li_2O = 0.0748 \times F^2 + 0.0893 \times F \qquad\qquad (3)$$

385   This equation may be used for estimating the Li_2O content of hydrothermal mica from

386   the F content determined by EPMA. The Li contents calculated with this equation are in

387   agreement with the values measured by LA–ICP–MS ($R^2 = 0.82$) (Fig. 9d).

388   The Li_2O contents determined using EPMA indicate that the biotites in the biotite

389   granite and two-mica granite are siderophyllite and protolithionite (lithian

390   siderophyllite), respectively, which is consistent with the trend of magmatic evolution

391   (Fig. 10). Lithian siderophyllites in the greisen have higher Li_2O contents (1.57–1.74

392   wt%) than those in the two-mica granite (0.88–1.33 wt%). They are protolithionites

393   following the classification proposed by Tischendorf et al. (1997) (Fig. 10a).

394   Alternatively, they belong to the transitional series between dioctahedral and

395   trioctahedral micas based on Foster (1960) (Fig. 10b), also termed the ferrous aluminum

396   micas in Monier and Robert (1986). Muscovites in the two-mica granite, muscovite

397   granite, and greisen all belong to the transitional series, but in their octahedral sites they

398   have lower contents of Li_2O and divalent metal cations (Fe^{2+}, Mn^{2+}, Mg^{2+}) than the

399   lithian siderophyllites in the greisen (Fig. 10b).

400

401   **Variation of halogens in the melts and fluids**

402   The micas display a wide range of F contents (0.41–4.26 wt%). Overall, the

403   siderophyllites in the biotite granite have the lowest F contents (0.84–1.95 wt%), while

404   micas in the greisen have the highest (up to 4.26 wt%). The cores of zoned micas in the

405   muscovite granite and greisen typically have higher F contents (0.73–4.26 wt%) than

406   the rims (0.41–3.46 wt%) (Supplementary Table 6). In addition, the micas have much

407   lower Cl contents (up to 0.06 wt%) than F (0.41–4.26 wt%; Supplementary Table 6).

408   Indeed, the substitution of Cl into the OH site of mica is much less favorable than the

409   substitution of F (Munoz 1992), due to the significantly larger ionic radius of Cl^- (1.81

410   Å; cf. $F^- = 1.31$ Å and OH = 1.38 Å).

411   The exchange of F, Cl, and OH between micas and associated fluids may be used

412   to obtain the halogen fugacity of the fluids (Munoz 1984) as well as the halogen

413   concentrations in the melt (Zhang et al. 2012). Since the measured Cl contents of the

414   Xihuashan micas are extremely low (mostly less than 0.06 wt%, with large errors of

415   10%–15%), the exchange of Cl and OH between micas and associated fluids is not

20

416   further discussed.

417   The equations used for estimating the F fugacity of a fluid [log(fH$_2$O/fHF)] and

418   the F concentration in a melt (W_F^m) are listed as follows (Munoz 1984; Zhang et al.

419   2012):

420         $$\log\left(\frac{fH_2O}{fHF}\right) = \frac{1000}{T} \times \left(2.37 + 1.1X_{Mg}\right) + 0.43 - \log(\frac{X_F}{X_{OH}})^{mica} \qquad (4)$$

421

422         $$W_F^m = \frac{1}{1+\frac{fH_2O}{fHF}} \times \frac{19}{18.02} \times \frac{10^6}{0.000943\times T - 0.574} \qquad (5)$$

423   where X$_F$ and X$_{OH}$ are the mole fractions of F and OH in the hydroxyl site of mica,

424   respectively, X_{Mg} is the mole fraction of Mg in the octahedral site of mica, and T is the

425   crystallization temperature of mica in degrees Kelvin. Due to the difficulties in

426   calculating the crystallization temperatures of micas directly, the homogenization

427   temperatures of fluid inclusions (Mu et al. 1982; Zhang et al. 1982; Wei et al. 2011) and

428   melt inclusions (Chang and Huang 2002) in the Xihuashan tungsten deposit were

429   adopted to estimate the F fugacity of the fluids and the F content of the melt. The melt

430   inclusions in beryl from the wolframite–quartz veins in the Xihuashan tungsten deposit

431   gave a homogenization temperature of 720 °C (Huang et al. 2006), consistent with the

432   mean value (703 ± 28 °C; range of 643 to 890 °C) of the temperature given by the

433   Ti-in-zircon thermometer for the Xihuashan granites (Yang et al. 2018), which can be

434   regarded as the crystallization temperature of the Xihuashan granitic magma. In addition,

435   there were two generations of fluid in the Xihuashan granites, based on stable oxygen

436   isotope data from fluid inclusions (Mu et al. 1982; Zhang et al. 1982; Wei et al. 2011).

437   The primary ore-fluid had a temperature of 370 ± 30 °C (fluid of magmatic origin, δ^{18}O

21

438   = 10‰–13‰) and the late fluid had a temperature of 220 ± 10 °C (meteoric water, $\delta^{18}O$

439   = −8.7‰ to +3.1‰). It is assumed that the primary ore-fluid was related to the

440   formation of the cores of zoned micas in the greisen and that the late fluid was related to

441   the rims of zoned micas in the muscovite granite and greisen, and this will be discussed

442   later. The results of calculations using Eq (4) and Eq (5) are given in Supplementary

443   Table 6 and shown on Fig. 11. Overall, the log(fH$_2$O/fHF) values of the fluid decrease

444   from the greisen (4.15–4.65, mean = 4.33) to the biotite granite (3.95–4.42, mean = 4.14)

445   to the two-mica granite (3.38–3.82, mean = 3.65), and to the muscovite granite (3.26–

446   3.91, mean = 3.56) (Fig. 11a). In general, F is preferentially concentrated in a residual

447   melt when the volatiles of a hydrous magma are exsolved (Zhu and Sverjensky 1992).

448   The loss of volatiles from a hydrous magma will enhance the F content in the residual

449   melt, producing a lower OH/F value in the melt but a higher OH/F ratio in the

450   associated fluid. This implies that the F content (W_F^m values) of the melt increases

451   gradually from the biotite granite to the two-mica granite, and to the muscovite granite

452   (Fig. 11c).

453   With regard to the zoned micas in the muscovite granite and greisen, the fluids

454   related to the rims always have higher log(fH$_2$O/fHF) values than those related to the

455   cores (Fig. 11a). In addition, the rims of micas in the muscovite granite and greisen all

456   have higher Fe^{3+}/Fe^{2+} ratios than the cores (Fig. 11b). As the Fe^{3+}/Fe^{2+} ratio shows a

457   positive correlation with temperature (Wones and Eugster 1965), the higher Fe^{3+}/Fe^{2+}

458   ratios of the rims of zoned micas cannot be attributed to lower temperatures relative to

459   the magmatic cores, but an increasing oxygen fugacity and H$_2$O content.

22

460

Magmatic evolution of the Xihuashan intrusive rocks

461

462   The Xihuashan granites have extremely variable whole-rock and mineral

463   compositions, indicating that the rocks might have undergone varying degrees of

464   fractional crystallization.

465   The micas show a wide range of compositions (Fig. 10) from siderophyllite to

466   lithian siderophyllite, which corresponds to the evolution of the magma from the biotite

467   granite to the two-mica granite. As a peraluminous melt system evolves, Rb and Cs are

468   compatible in biotite while Y, Th, and REEs are incompatible (Bea et al. 1994). The

469   micas in our samples have low Y, Th, and REE contents but extremely high Rb and Cs

470   contents (Supplementary Table 6), suggesting that fractionation of the micas was

471   insignificant during the magmatic evolution. The increase of Rb and Cs contents with

472   decreasing K/Rb ratios from the biotite granite to the two-mica granite (Fig. 12a, b)

473   further excludes the possibility of significant mica fractionation during the magmatic

474   evolution of the Xihuashan intrusive rocks. With regard to the feldspars, Rb and Cs are

475   compatible in K-feldspar but incompatible in plagioclase (Bea et al. 1994), and an

476   increase of Rb and Cs in micas from the biotite granite to the two-mica granite (Fig. 7b,

477   c) could be attributed to a remarkable fractionation of plagioclase. It is also consistent

478   with the fact that our samples are notably enriched in Li, Rb, and Cs compared with

479   upper continental crust (UCC) (Fig. 4b). In addition, xenotime rather than apatite is the

480   principal phosphate mineral in Ca-poor peraluminous granitic melts (Wang et al. 2003).

481   In the Xihuashan pluton, the two-mica granite and muscovite granite samples contain

482   abundant xenotime with apatite being absent or rare, which indicates an essentially

483   Ca-poor peraluminous melt. The Ca-poor melt would have been the product of the

484   fractional crystallization of Ca-rich minerals, including apatite and plagioclase.

485   However, apatite fractionation, if it existed, could not have been primarily responsible

486   for producing the Ca-poor peraluminous melt, because it would have required a large

487   amount of phosphorus to exhaust the Ca in the system. Therefore, the presence of

488   abundant xenotime in the two-mica granite and muscovite granite may provide evidence

489   for plagioclase fractionation during the magmatic evolution, which is consistent with

490   decreasing MgO, TiO_2, Fe_2O_3 and CaO (Fig. 3) and gradually pronounced Eu negative

491   anomalies (Fig. 4a) from biotite granite to two-mica granite to muscovite granite.

492   Collectively, the variations in composition of mica in the Xihuashan pluton are

493   consistent with a trend of magmatic evolution from the biotite granite to the two-mica

494   granite to the muscovite granite. This evolutionary process was controlled mainly by the

495   fractional crystallization of plagioclase.

496

The hydrothermal evolution recorded in zoned micas and siderite

498   Micas in highly evolved granites and pegmatites always show compositional

499   zoning as a result of the evolution of the magma or sub-solidus fluid interactions

500   (Roda–Robles et al. 2007; Van Lichtervelde et al. 2008; Li et al. 2015). Volatile

501   components (e.g., H_2O, F, B, and P) and lithophile rare metal elements (e.g., Li, Rb, Cs,

502   Ta, Nb, Sn, and W) are incompatible in major rock-forming minerals during the

503   evolution of peraluminous melt systems. Accordingly, the Li_2O, F, Rb, Cs, Nb, and Ta

24

504 contents of mica may increase with the progressive evolution of the magma (Roda–

505 Robles et al. 2007), leading to differences in brightness in the zoning textures revealed

506 by the BSE images (Fig. 5). Since fluorine is concentrated in the melt during the

507 differentiation of a granitic melt when there is lack of fractionation of fluorine-enriched

508 minerals such as apatite and mica (Li et al. 2015), the supercritical fluids exsolved from

509 a highly fractionated Sn–W granitic magma are usually enriched in F (Thomas et al.

510 2005). The addition of F into the melt may increase the abundance of non-bridging

511 oxygens (NBOs), which increases the proportion of WO_4^{2-} tetrahedra in silicate melt.

512 Indeed, with our samples, the $\log(f\mathrm{H_2O}/f\mathrm{HF})$ values decrease and the W_F^m values

513 increase from the biotite granite to the two-mica granite and the muscovite granite (Fig.

514 11), consistent with an evolution of magma without involvement of any notable

515 fractionation of apatite or mica. However, the rims of the zoned micas in the muscovite

516 granite and greisen have lower Li_2O, Rb, Cs, Nb, Ta, and F contents but higher Si and

517 Al contents than the cores (Fig. 9; Supplementary Table 6). Moreover, the rims of the

518 zoned micas belong to the transition series (Fig. 10), similar to the ferrous aluminum

519 micas described by Monier and Robert (1986), which were the products of metasomatic

520 alteration. Altogether, the characteristics of the compositional zoning and dissolution

521 textures (Fig. 5) of the micas in our samples indicate that they were produced by fluid–

522 rock interactions rather than melt differentiation. Furthermore, the micas would have

523 been enriched in F when they interacted with supercritical fluids exsolved from a highly

524 differentiated magma. The rims of the zoned micas show significant depletion in F

525 (Supplementary Table 6), corresponding to higher $\log(f\mathrm{H_2O}/f\mathrm{HF})$ values than the mica

25

526   cores, and this suggests a process of subsolidus replacement by a dominantly external

527   aqueous fluid.

528   The siderites have variable REE contents (ΣREE = 6–110 ppm; Supplementary

529   Table 8) and negative Eu anomalies, which would have been caused by the fluids

530   having variable REE contents or pH values (Michard 1989; Möller et al. 2004). In

531   general, the total REE content of hydrothermal solutions increases with decreasing pH

532   values (Michard 1989; Möller et al. 2004). In acidic solutions, LREEs are likely to be

533   enriched in aqueous phases, while HREEs are preferentially adsorbed on mineral

534   surfaces (Schwinn and Markl 2005). Thus, siderites precipitated from fluids with

535   variable pH values would have variable LREE/HREE ratios. The siderites in our

536   samples of muscovite granite and greisen have similar $(La/Yb)_N$ ratios. Instead, different

537   sources of fluid with variable REE contents are more likely to have been responsible for

538   the wide range of REE contents that characterizes the siderites, and this is consistent

539   with the log(fH$_2$O/fHF) values of the rims of zoned micas in the greisen being generally

540   lower than those in the muscovite granite (Fig. 11).

541   The stability of siderite depends on the CO_2 fugacity and the oxygen fugacity

542   (Fusswinkel et al. 2013). Siderite is stable under conditions of low oxygen fugacity (i.e.,

543   fO_2 = −38 to −54) and high CO_2 fugacity (i.e., fCO_2 = −1 to +2) (Fusswinkel et al. 2013).

544   It is replaced by magnetite in the reaction "6 Siderite + O$_2$ = 2 Magnetite + 6 CO$_2$"

545   when the oxygen fugacity increases or the activity of CO_2 decreases (Hovis et al. 1991).

546   The lack of magnetite in the muscovite granite and greisen (Table 1) may be attributed

547   to a relatively reducing environment where the siderite crystals were stable. Therefore,

26

548   the presence of siderite in the Xihuashan granite indicates a reducing environment

549   during the fluid–rock interactions, and the oxygen fugacity (fO$_2$) would have been close

550   to the value constrained by the "C + O$_2$ = CO$_2$" buffer (Yang et al. 2013).

551   The fluids associated with the tungsten mineralization would have been of

552   external origin or exsolved from the melt (e.g., Wang et al. 2003; Wei et al. 2012;

553   Harlaux et al. 2018). It is noteworthy that the cores of zoned micas in the greisen and

554   muscovite granite show much lower log(fH$_2$O/fHF) values and Fe^{3+}/Fe^{2+} ratios than the

555   rims (Fig. 11), and this indicates that the fluids associated with the formation of the

556   greisen and muscovite granite had a higher fluorine fugacity and were even more

557   reducing than the fluids associated with the fluid–rock interactions. Such fluids might

558   have been supercritical fluids exsolved from the highly fractionated magma, and this is

559   consistent with the observation that the log(fH$_2$O/fHF) values associated with the cores

560   of micas in the greisen are close to those associated with magmatic micas in the granite

561   samples (Fig. 11a, b). In addition, based on the fluid inclusion studies, Wei et al. (2012)

562   proposed that a mixture of fluids from two different sources could have been

563   responsible for the deposition of wolframite in the Xihuashan deposit. Fluids with a low

564   oxygen fugacity and enriched in fluorine may be derived from fluid oversaturation of a

565   granitic melt (Maruejol et al. 1990; Wang et al. 2003), which may contribute to the

566   formation of the cores of zoned micas in the greisen and muscovite granite. The

567   fluid-rock interactions, however, involved mainly meteoric fluids that were associated

568   with alteration of early Li-rich micas into late muscovite (Legros et al. 2016). These

569   fluids were also enriched in H$_2$O and CO$_2$, as recorded in the rims of zoned mica and the

27

570   siderite in the greisen and muscovite granite. In particular, the rims of zoned micas in

571   the greisen samples have lower $\log(fH_2O/fHF)$ values than those in the muscovite

572   granite samples, which further indicates that the fluids associated with the greisen

573   contained larger amounts of magma-derived fluids.

574

575   **Tungsten enrichment during the evolution of the magmatic–hydrothermal system**

576   Tungsten deposits are usually associated with fractionated granitic magmas and

577   hydrothermal processes (e.g., Xie et al. 2016; Breiter et al. 2017). In peraluminous

578   granitic systems, tungsten commonly behaves as an incompatible element for most

579   rock-forming minerals and accumulates in melts during fractional crystallization

580   (Hulsbosch et al. 2016). The WO_3 content of columbite-group minerals increases from 5

581   to 15 wt% during magma evolution (Xie et al. 2018), which confirms that tungsten can

582   be concentrated in melts. In addition, mica may be used as an indicator of ore-forming

583   potential for tungsten granites (Yin et al. 2019). In the Xihuashan granite, the tungsten

584   content of micas increases from the biotite granite (0.36–12.8 ppm) to the muscovite

585   granite (14.9–84.1 ppm) while the K/Rb ratio of micas decreases (Fig. 12c), consistent

586   with the trend of magma evolution. However, magmatic wolframite can crystallize

587   directly from a highly evolved granite melt under restricted conditions, as for example

588   from a peraluminous melt with high concentrations (6 wt%) of fluxing compounds (B,

589   Li, F, and P) and with crystallization temperatures lower than 550 °C at 800 bars (Che et

590   al. 2013). In most cases, the required conditions for saturation of magmatic wolframite

591   cannot be reached in granitic melts due to the low levels of fluxing compounds. It is rare

592   to have a wolframite of purely magmatic origin, and hydrothermal fluids typically play

593   a significant role in W mineralization. In the Xihuashan granite, the lack of wolframite

594   in the muscovite granite suggests that the tungsten concentration in the melts did not

595   become saturated relative to wolframite as a result of magmatic fractionation. Indeed,

596   the whole-rock content of W in the Xihuashan granite (4.82–65.3 ppm) is much lower

597   than the solubility of wolframite in peraluminous granitic melts (~1000 ppm WO_3 at

598   750–800 °C; Štemprok 1990). Therefore, although tungsten may be concentrated in a

599   melt via magmatic fractionation, in the magmas of the Xihuashan pluton the W contents

600   were always much lower than the oversaturation level during the evolution of the

601   magma.

602   Hydrothermal activities may play an important role in W mineralization (e.g.,

603   Štemprok 1990; Li et al. 2018). The wolframite crystals in the greisen are intergrown

604   with zoned mica (Fig. 5d), and they would have formed during hydrothermal activity.

605   However, wolframite is absent from the muscovite granite, even though the abundant

606   zoned micas with dissolution textures (Fig 5a, b) provide evidence of fluid alteration.

607   The rims of zoned micas in the muscovite granite have lower W contents than the cores

608   (Fig. 7f), which was probably due to hydrothermal alteration. The hydrothermal fluids

609   liberated W from the micas and resulted in formation of a W-depleted rim, whereas W

610   was concentrated in the hydrothermal fluid. The elevated CO_2 concentration may

611   increase the acidity of fluid and the solubility of tungstates, which facilitates

612   concentrating W in the fluids (Liu et al. 2017; Wang et al. 2020). The rims of zoned

613   micas in the greisen have higher W contents than those in the muscovite granite (Fig.

29

614   7f), which was probably due to different CO_2 and W concentrations in the hydrothermal

615   fluids. The W-rich rims of zoned mica in the greisen clearly show the involvement of

616   W-rich fluids in the formation of the tungsten deposit (Figs 5d, 7h). Furthermore, a fluid

617   rich in F, Li, and rare metals can induce greisenization in the upper part of the granitic

618   cupola, causing dissolution of feldspars and crystallization of new quartz, zinnwaldite,

619   fluorite, and topaz (Breiter et al. 2017). Greisenization consumes H^+ and increases the

620   pH value of fluids following the reaction:

$$2Na[AlSi_3O_8] + K[AlSi_3O_8] + 2H^+ = KAl_2[AlSi_3O_{10}](OH)_2 + 6SiO_2 + 2Na^+$$
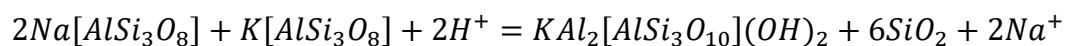
621   　　Albite　　　　K-feldspar　　　　　　　Muscovite　　　　　Quartz

622   　　An increase in fluid pH during greisenization will destabilize the polymeric

623   tungstates to form WO_4^{2-} and other monomeric tungstates (Wang et al. 2020), which

624   may interact with metal cations (i.e., Fe, Mn or Ca) to form wolframite or scheelite.

625   Thus, greisenization promoted the generation of WO_4^{2-} in the fluids, and potentially

626   contributed to the formation of tungsten deposit (Audéta et al. 2008; Harlaux et al.

627   2017; Legros et al. 2019).

628

629   　　　　　　　　　IMPLICATIONS FOR TUNGSTEN MINERALIZATION

630   　　The hydrothermal fluid related to the W deposits would have different sources,

631   including an early magmatic fluid and a later metamorphic or meteoric fluid (Wei et al.

632   2012; Hulsbosch et al. 2016). However, the signature of the early magmatic fluid is

633   usually overprinted by the metamorphic or meteoric fluid (Wei et al. 2012; Hulsbosch et

634   al. 2016). Thus, attempts to determine whether a magmatic or metamorphic fluid was

635 responsible for W mineralization associated with a granite typically provide

636 inconclusive results. In the Xihuashan tungsten granite, the composition and zoning

637 texture of micas may provide some valid constraints on the nature of fluids and the

638 mechanism for precipitation of wolframite. The interaction between external aqueous

639 fluids and pre-existing micas has played an important role in leaching the ore-forming

640 elements (i.e., W and Fe) from the micas into the fluids, which is proved by remarkably

641 dropped concentrations of W and Fe from the cores to the rims of zoned micas in the

642 muscovite granite (Figs 6c, 7h). The W-rich rims of the zoned mica in the greisen,

643 together with the high whole-rock F contents (up to 1.22 wt%) of greisen samples,

644 demonstrate that the fluid related to the greisenization was enriched in F and W. The

645 F-rich fluids involved in the greisenization were most likely derived from the

646 oversaturation of a rare-metal granitic melt (Maruéjol et al. 1990; Launay et al. 2018;

647 Legros et al. 2018). Therefore, the rims of zoned mica in the greisen are enriched in W

648 due to the effect of W-rich fluids, whereas the rims of zoned mica in the muscovite

649 granite are related to the dominant external fluids that had much lower W contents. This

650 difference between the zoned micas in the greisen and muscovite granite indicates that

651 the process of greisenization was critical for extracting and transporting the W from the

652 melts/rocks into the fluids (Launay et al. 2018).

653 The processes identified in the Xihuashan tungsten granite may have been active

654 in many wolframite-bearing quartz-vein deposits elsewhere. Some famous metallogenic

655 provinces or plutons, such as the French Massif Central, the Erzgebirge, the northern

656 Canadian Cordillera, Iberian Massif and the Nanling and Dahutang mineralization

657   regions in South China, are closely related to magmatic and hydrothermal processes

658   (e.g., Legros et al. 2016, 2018; Breiter et al. 2017a, b; Legro et al. 2016, 2018; Xie et al.

659   2018; Yin et al. 2019). Furthermore, the micas in these tungsten-bearing granites are

660   commonly zoned and enriched in fluxing compounds (e.g., Breiter et al. 2017; Harlaux

661   et al. 2018a; Xie et al. 2018; Yin et al. 2019), which might also record the processes of

662   tungsten mineralization related to fluids during their evolution through the magmatic–

663   hydrothermal transition. Collectively, the external and magmatic fluids all facilitated the

664   tungsten mineralization. The overall scenario of the ore-forming processes for the

665   Xihuashan tungsten deposit is summarized as follows (Fig. 13):

666   1) The W and F were gradually concentrated in the magmatic fluids via magmatic

667   fractionation.

668   2) The W and F-rich magmatic fluids pooled in the granitic cupola and induced

669   greisenization. This process destabilized the polymeric tungstates to form WO_4^{2-} in the

670   fluids.

671   3) The pooled fluid flow might have been overpressured, eventually being

672   injected into the vein system.

673   4) Tungsten and the fluids would have migrated along faults and fractures with

674   tungsten being precipitated as vein-type wolframite at favorable sites when the W and

675   F-rich fluid was mixed with an external meteoric fluid.

676   Therefore, tungsten mineralization is closely related to the mixture of magmatic

677   and external fluids. Furthermore, the presence of siderite in the greisen and muscovite

32

678   granite indicates that the fluids are rich in Fe and Mn, which are ore-forming elements

679   generating metal complexes with WO_4^{2-} during the precipitation of wolframite.

680   　　The input of Fe and Mn in the fluids would be attributed to the reactions between

681   external CO_2-bearing fluid and host-rock (Lecumberri-Sanchez et al., 2017; Wang et al.

682   2020). In addition, the fluid boiling process might also take the role of triggering the

683   precipitation of wolframite, as the transport of volatile elements (e.g., CO_2, H^+) into

684   vapor phase changes the chemical composition of the fluid (Korges et al., 2018).

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1005

1006   **FIGURE CAPTIONS**

1007   **Fig. 1**   (a) The distribution of granites and volcanic rocks of different ages in SE China

1008   (modified from Li et al. 2010), showing the locations of the Nanling W–Sn

1009   polymetallic mineralization region (NPMR) and the Xihuashan tungsten

1010   deposit. (b) Simplified geological map of the Xihuashan pluton (modified from

1011   Guo et al. 2012), showing four successive intrusions in the Xihuashan mining

1012   district (G1–G4, which correspond to coarse-grained biotite granite,

1013   medium-grained biotite granite, medium- to fine-grained biotite granite, and

1014   fine-grained biotite granite, respectively. After Wang et al. 2003).

1015   **Fig. 2**   Hand specimen photographs (left) and photomicrographs (right) showing the

1016   petrography of the Xihuashan granites and greisen (in cross-polarized light)

1017   and the different micas encountered in this study. (a) Biotite granite. Biotite

1018   (siderophyllite) and associated apatite and ilmenite inclusions. (b) Two-mica

1019   granite. Biotite (lithian siderophyllite), muscovite and associated apatite and

1020   ilmenite inclusions; the rim of the biotite was eroded, as shown by the irregular

1021   embayed form. (c) Muscovite granite. Muscovite is present between the

1022   feldspar (both plagioclase and K-feldspar) and quartz. (d) Greisen. Biotite

1023   (lithian siderophyllite) and Li-mica occur between quartz grains. Mineral

1024   abbreviations: K-feldspar (Kfs), plagioclase (Pl), quartz (Qtz), biotite (Bt),

1025   muscovite (Ms), apatite (Ap), ilmenite (Ilm), zircon (Zrn), xenotime (Xtm).

1026   **Fig. 3**   The diagrams of (a) TAS, (b) A/NK vs. A/CNK, and (c–h) TiO_2, MgO, P_2O_5, F,

1027 K_2O, and CaO contents versus Fe_2O_3 for the Xihuashan pluton. The total Fe in

48

1028   the whole rock analyses is reported as Fe_2O_3. Plots I–IV from Guo et al. (2012)

1029   represent the data of different intrusions.

1030   **Fig. 4**   Chondrite-normalized REE patterns and mean upper crust-normalized

1031   multi-element diagrams for rocks in the Xihuashan pluton. The chondrite and

1032   mean upper crust values are from Taylor and McLennan (1985) and Rudnick

1033   and Gao (2003), respectively. Data in shaded field are from Guo et al. (2012).

1034   **Fig. 5**   (a-d) Back-scattered-electron (BSE) images of micas in different intrusions of

1035   the Xihuashan pluton; (e) and (f) representative BSE images of wolframite in

1036   greisen.

1037   **Fig. 6**   Diagrams of TiO_2, Al_2O_3, FeO^{Tot}, MgO, F, and Cl contents versus SiO_2 for

1038   micas in the rocks of the Xihuashan pluton. BG = biotite granite, TMG =

1039   two-mica granite, and MG = muscovite granite.

1040   **Fig. 7**   Diagrams of V, Cs, Rb, Nb, Ta, W, Sn, and Nb/Ta versus FeO^{Tot} + MnO + MgO

1041   contents for micas in the rocks of the Xihuashan pluton. The dashed lines

1042   connect the cores and corresponding rims of zoned mica. BG = biotite granite,

1043   TMG = two-mica granite, and MG = muscovite granite.

1044   **Fig. 8**   (a) Representative BSE image, (b) Raman spectra, (c) chondrite-normalized

1045   REE patterns, and (d) mean upper crust-normalized multi-element diagrams for

1046   siderite in the Xihuashan muscovite granite and greisen. Chondrite and mean

1047   upper crust values are from Taylor and McLennan (1985) and Rudnick and

1048   Gao (2003), respectively.

1049   **Fig. 9**   (a, b) Variation of measured Li_2O versus SiO_2 and F for micas in the rocks of

49

1050   the Xihuashan pluton. The regression equations of (1) $Li_2O = 0.289 \times SiO_2 -$

1051   9.658 and (2) $Li_2O = 0.3935 \times F^{1.326}$ from Tischendorf et al. (1997) can be

1052   applied to trioctahedral and dioctahedral micas, respectively. (c) Relationships

1053   between measured Li_2O and F in hydrothermal micas, which define a

1054   regression equation of (3) $Li_2O = 0.0748 \times F^2 + 0.0893 \times F$ that could be

1055   applied to the hydrothermal micas in our samples. (d) Comparison between

1056   measured and calculated Li_2O for micas in the rocks of the Xihuashan pluton.

1057   Some data for the hydrothermal micas were taken from previous studies (Van

1058   Lichtervelde 2008; Akoh and Ogunleye 2014; Xie et al. 2015; Yin et al. 2019;

1059   Ma et al. 2020). BG = biotite granite, TMG = two-mica granite, and MG =

1060   muscovite granite.

1061   **Fig. 10**   (a) Variations in the compositions of mica from the Xihuashan pluton on the

1062   diagram of $[Mg - Li]$ vs. $[Fe^{Tot} + Mn + Ti - Al^{VI}]$ (according to Tischendorf et

1063   al. 1997). (b) Relationships between Li and octahedral cations (based on Foster

1064   1960). The arrows indicate the trends in mica compositions due to magmatic

1065   evolution. The shadow area represents the field of the transitional series,

1066   known as the ferrous aluminous micas (Monier and Robert 1986). D.T. and T.T.

1067   denote the compositional trends of dioctahedral and trioctahedral micas,

1068   respectively. BG = biotite granite, TMG = two-mica granite, and MG =

1069   muscovite granite.

1070   **Fig. 11**   Plots of (a) SiO_2 versus $\log(fH_2O/fHF)$, (b) Fe^{3+}/Fe^{2+} versus $\log(fH_2O/fHF)$,

1071   and (c) SiO_2 versus W_F^m for micas in the Xihuashan pluton. BG = biotite

1072   granite, TMG = two-mica granite, and MG = muscovite granite.

1073 **Fig. 12** Plots of Rb, Cs, and W versus K/Rb for micas in the Xihuashan pluton. BG =

1074   biotite granite, TMG = two-mica granite, and MG = muscovite granite.

1075 **Fig. 13** Cartoon of tungsten mineralization in the Xihuashan pluton, illustrating the

1076   relationship between the evolution of the magmatic–hydrothermal system and

1077   the W mineralization. The spatial relationships between the bodies of granite

1078   and greisen are based on previous investigations (Liu 2005; Huang et al. 2006).

1079

Figure showing two spider diagrams. (a) Sample/Chondrite normalized diagram with elements La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu; labeled Guo et al. (2012). (b) Sample/Upper Crust normalized diagram with elements Cs, Rb, Ba, W, Th, Ta, Nb, K, La, Ce, Pb, Sr, P, Nd, Hf, Zr, Sm, Sn, Ti, Li, Dy, Tb, Y, Yb, Lu; labeled Guo et al. (2012). Legend: biotite granite, two-mica granite, muscovite granite, greisen.

(a) Biotite granite — ilmenite

(b) Two-mica granite — apatite, ilmenite

(c) Muscovite granite — siderite, siderite

(d) Greisen — siderite, wolframite

(e) Greisen — wolframite, Muscovite, Muscovite

(f) Greisen — Muscovite, wolframite

(a) V (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(b) Cs (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(c) Rb (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(d) Nb (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(e) Ta (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(f) Nb/Ta — FeO^{Tot}+MnO+MgO (wt%)
(g) Sn (ppm) — FeO^{Tot}+MnO+MgO (wt%)
(h) W (ppm) — FeO^{Tot}+MnO+MgO (wt%)

siderophyllite in BG
lithian siderophyllite in TMG
muscovite in TMG
core of mica in MG
rim of mica in MG
core of mica in greisen
rim of mica in greisen

(a)

muscovite

siderite

quartz

lithian
siderophyllite

(b)

1081.73

276.97

(a) $Li_2O = 0.289 \times SiO_2 - 9.658$

(b) $Li_2O = 0.3935 \times F^{1.326}$

(c) $Li_2O = 0.0748 \times F^2 + 0.0893 \times F$ ($R^2 = 0.95$, n=76)

(d) $R^2 = 0.82$, n=52

Legend:
- siderophyllite in BG
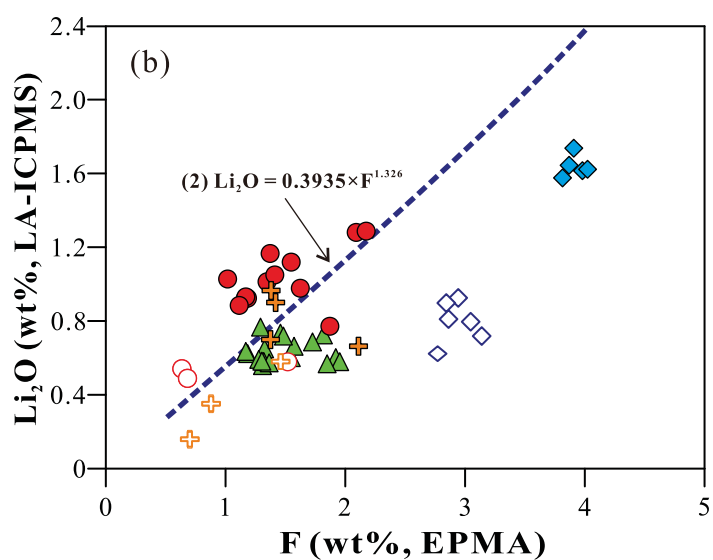- lithian siderophyllite in TMG
- muscovite in TMG
- core of mica in MG
- rim of mica in MG
- core of mica in greisen
- rim of mica in greisen
- literature data

(a) Fe$_{tot}$+Mn+Ti-Al^{VI} versus Mg-Li classification diagram showing mica compositions, with fields for siderophyllite, proto-lithionite, zinnwaldite, lepidolite, Li-muscovite, Fe-biotite, Li-phengite, phengite, and muscovite.

(b) Li vs Octahedral R^{3+} (Al, Fe^{3+})+Ti^{4+} vs Octahedral R^{2+} (Fe^{2+}, Mn, Mg) ternary classification diagram with fields for polylithionite, lepidolite, zinnwaldite, siderophyllite, protolithionite, muscovite, lithian muscovite, mixed forms, and transition series.

Legend:
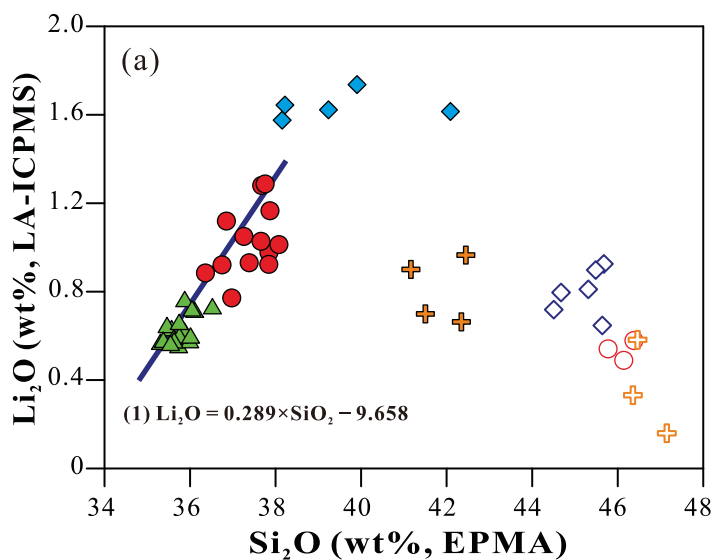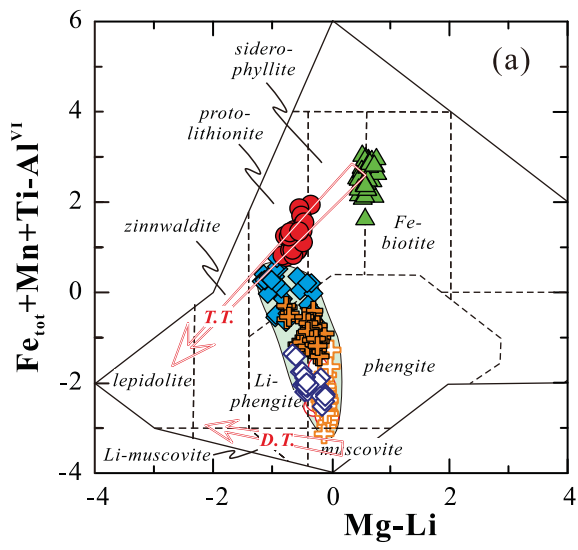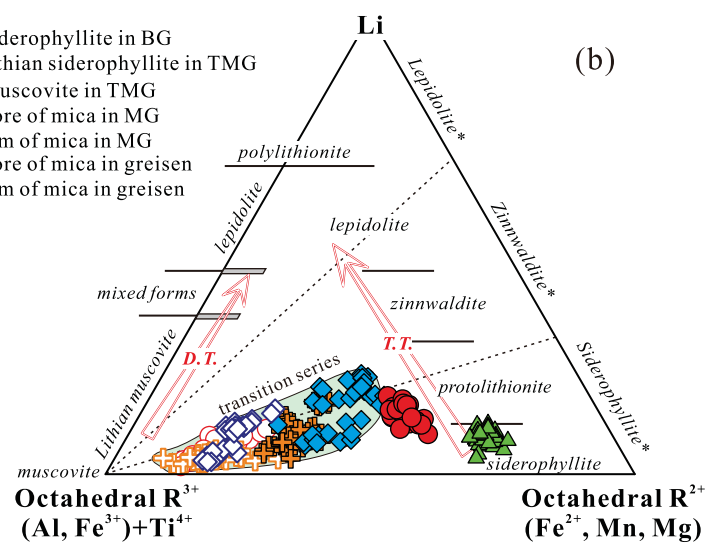- siderophyllite in BG
- lithian siderophyllite in TMG
- muscovite in TMG
- core of mica in MG
- rim of mica in MG
- core of mica in greisen
- rim of mica in greisen

Legend:
- siderophyllite in BG
- lithian siderophyllite in TMG
- muscovite in TMG
- core of mica in MG
- rim of mica in MG
- core of mica in greisen
- rim of mica in greisen