

1 **Revision 3**

2 **Identifying serpentine minerals by their chemical compositions with machine learning**

3 **Word Count: 8454**

4

5 Shichao Ji^{1,2}, Fang Huang^{3*}, Shaoze Wang⁴, Priyantana Gupta⁵, William Seyfried, Jr⁶, Hejia
6 Zhang⁷, Xu Chu⁸, Wentao Cao⁹, J ZhangZhou¹⁰

7

8 ¹ CAS Key Laboratory of Mineralogy and Metallogeny/Guangdong provincial Key Laboratory of
9 Mineral Physics and Materials, Guangzhou Institute of Geochemistry, Chinese Academy of
10 Sciences (CAS), Guangzhou 510640, China

11 ² CAS Center for Excellence in Deep Earth Science, Guangzhou 510640, China

12 ³ CSIRO Mineral Resources, Kensington, WA 6151, Australia

13 ⁴ Surface Science Lab, Physics and Astronomy Department, Ohio University, Athens, OH 74501,
14 USA

15 ⁵ Indian Institute of Technology, Kharagpur, West Bengal 721302, India

16 ⁶ Department of Earth and Environmental Sciences, University of Minnesota, Minneapolis, MN
17 55455, USA

18 ⁷ School of the Environment, Yale University, New Haven, CT 06511, USA

19 ⁸ Department of Earth Sciences, University of Toronto, Toronto, ON M5S 3B1, Canada

20 ⁹ Department of Geology and Environmental Sciences, State University of New York at
21 Fredonia, Fredonia, NY 14063, USA

ABSTRACT

25

26 The three main serpentine minerals, chrysotile, lizardite, and antigorite, form in various geological
27 settings, and have different chemical compositions and rheological properties. The accurate
28 identification of serpentine minerals is thus of fundamental importance to understanding global
29 geochemical cycles and the tectonic evolution of serpentine-bearing rocks. However, it is
30 challenging to distinguish specific serpentine species solely based on geochemical data obtained
31 by traditional analytical techniques. Here, we apply machine learning approaches to classify
32 serpentine minerals based on their chemical compositions alone. Using the Extreme Gradient
33 Boosting (XGBoost) algorithm, we trained a classifier model (overall accuracy of 87.2%) that is
34 capable of distinguishing between low-temperature (chrysotile and lizardite) and high-temperature
35 (antigorite) serpentines mainly based on their SiO₂, NiO, and Al₂O₃ contents. We also utilized a
36 *k*-means model to demonstrate that the tectonic environment in which serpentine minerals form
37 correlates with their chemical compositions. Our results obtained by combining these classification
38 and clustering models imply the increase of Al₂O₃ and SiO₂ contents and the decrease of NiO
39 content during the transformation from low- to high-temperature serpentine (i.e., lizardite and
40 chrysotile to antigorite) under greenschist–blueschist conditions. These correlations can be used
41 to constrain mass transfer and the surrounding environments during the subduction of hydrated
42 oceanic crust.

43

44 **Keywords:** Serpentine, machine learning, XGBoost, classifications, *k*-means, clustering

45

46

INTRODUCTION

47 The hydration of upper mantle rocks and metamorphism are the two predominate processes
48 producing serpentine minerals (John et al. 2011; Plümper et al. 2017; Tamblyn et al. 2019; Zhang
49 et al. 2019; Ulrich et al. 2020), resulting in the wide occurrence of serpentines in various geological
50 settings (Rüpke et al. 2004; Reynard 2013; Canales et al. 2017). Lizardite and chrysotile are
51 produced during near-surface metasomatism or alteration, corresponding to greenschist facies or
52 lower-grade metamorphism (Auzende et al. 2006; Koutsovitis 2017; Boskabadi et al. 2020),
53 whereas antigorite dominates in blueschist- and eclogite-facies ultramafic rocks, which usually
54 represent subduction contexts at depths exceeding 75 km (Hattori and Guillot 2007; Zhang et al.
55 2009, 2019; Tamblyn et al. 2019). Antigorite is also less hydrous and contains more minor fluid-
56 mobile elements than lizardite and chrysotile (Deschamps et al. 2010; Debret et al. 2019).
57 Antigorite can also endure greater shear strain than lizardite or chrysotile, indicating the different
58 rheological properties of the serpentine minerals (Hirauchi and Katayama 2013; Plissart et al.
59 2019). Therefore, distinguishing serpentine minerals from one another is of fundamental
60 importance for constraining geothermal gradients and mass transfers, as well as for understanding
61 the mechanism of strain localization, which may induce exhumation, deformation, and seismicity
62 in subduction zones (Boudier et al. 2010; Deschamps et al. 2010; Reynard 2013; Schwartz et al.
63 2013; Rouméjon et al. 2019; Wheat et al. 2020).

64 Serpentine minerals are hydrous, Mg-rich, 1:1 (i.e., one tetrahedral silicate sheet, mainly
65 Si^{4+} occupying the tetrahedral sites, and one octahedral sheet, mainly Mg^{2+} occupying the
66 octahedral sites, stack alternately) trioctahedral clay minerals with the ideal composition
67 $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$. Because of their chemical compositional differences, the tetrahedral sheets are
68 slightly smaller than the octahedral sheets (Whittaker and Wicks 1970), and different sheet

69 arrangements accommodate this mismatch, producing the different structures of the serpentine
70 minerals, with lizardite, chrysotile, and antigorite being the three most abundant serpentine
71 minerals (Auzende et al. 2002; Boudier et al. 2010; Deschamps et al. 2013). Common substitutions
72 occurring in natural samples are Fe^{3+} , Ti^{4+} , Mn^{2+} , or Cr^{3+} for Mg^{2+} in the octahedral sites and Al^{3+}
73 for Si^{4+} in the tetrahedral sites (Evans 2004).

74 Modern techniques used to discriminate between serpentine minerals present specific
75 advantages and disadvantages. Serpentine minerals can be distinguished based on structural
76 differences (Mellini 1986; Banfield et al. 1995; Schwartz et al. 2013; Rouméjon et al. 2019),
77 generally by transmission electron microscopy (TEM), requiring important sample preparation.
78 Serpentine minerals can also be identified from their Raman spectra (Rinaudo et al. 2003;
79 Petriglieri et al. 2015). Auzende et al. (2004) and Groppo et al. (2006) found that antigorite can be
80 identified from the frequencies at which Si-O-Si and the inner and outer OH stretchings occur, and
81 the translation modes of OH-Mg-OH linkages (vibrating at $500\text{--}550\text{ cm}^{-1}$) can be used to
82 distinguish between chrysotile and lizardite. To identify serpentine based on these vibrational
83 differences, the serpentine species being analyzed should be well-crystallized and sufficiently
84 compositionally distinct. However, serpentine minerals may share the same chemical
85 compositions via substitutions (e.g., the substitution of Ni, Fe, and Al for Mg in octahedral sheets
86 and Al for Si in tetrahedral sheets; O'Hanley and Dyar 1998), and poorly crystallized serpentine
87 minerals are common. The associated shifts of the vibrational frequencies of these bonds can result
88 in different serpentine minerals sharing similar Raman spectra (Wang et al. 2015). Therefore, it
89 may be difficult to reliably distinguish poorly crystallized serpentine minerals based only on their
90 chemical compositions and spectroscopic features using traditional geological methodologies.

91 Recent studies suggest that machine learning (ML) approaches offer new opportunities for
92 classifying minerals based on their chemical compositions in high-dimensional space (Petrelli and
93 Perugini 2016; O'Sullivan et al. 2020; Valetich et al. 2021; Wang et al. 2021). Because they are
94 driven purely by data, ML models can unravel complexities in large datasets through a learning
95 process unimpeded by *a priori* defined conceptual models (e.g., the definition of an equilibrium
96 state; Hazen et al. 2019; Caricchi et al. 2020; Petrelli et al. 2020).

97 Here, we tested the potential of ML approaches as an alternative tool for classifying
98 serpentine minerals. We first compiled and filtered a dataset comprising eight elemental
99 concentrations in serpentine minerals. Then, we trained the Extreme Gradient Boosting (XGBoost)
100 classification and *k*-means clustering models, employing strategies to avoid overfitting. XGBoost
101 was applied to classify the serpentine minerals at first. Based on the results of these models, we
102 discussed the classification performance and feature importance, which can be used to explore the
103 features dominating the chemical differences between serpentines crystallized in natural systems.
104 Then, the *k*-means algorithm was used to cluster each class. We tested the clustering results to
105 identify the tectonic environments in which serpentine minerals form and to constrain the typical
106 variations in SiO₂, NiO, and Al₂O₃ contents during the replacement of chrysotile/lizardite by
107 antigorite in subduction zones.

108

109

DATA AND METHODS

110 Data Compilation

111 The dataset used in this study contains compositional data for serpentine-group minerals from
112 50 publications (see Supplementary Reference List), containing 1,375 chemical analyses of 566
113 serpentine phases. To explore the relationship between the serpentine minerals and their formation

114 conditions, we grouped the minerals as low-temperature, (L_Srp, 534 analyses of 264 lizardite and
115 chrysotile samples) vs, high-temperature serpentines (H_Srp, 841 analyses of 302 antigorite
116 samples; Supplementary Table S2). Lizardite has flat, stretched tetrahedral sheets with interlayer
117 H-bonds, favoring the coupled substitution of Fe, Ni, and Al for Mg in the octahedral sheets and
118 Al for Si in the tetrahedral sheets (Mellini 1986; Viti and Mellini 1997). Chrysotile forms multi-
119 walled nanotubes or nanoscrolls (Wicks and Whittaker 1975). Antigorite displays curved, wavy
120 layers similar to Roman roof tiles (e.g., Banfield et al. 1995; Ji et al. 2018): the octahedral sheets
121 are continuous and wavy, whereas the tetrahedral sheets undergo periodic reversals along the *a*-
122 axis to connect to the concave half-waves of adjacent octahedral sheets (Capitani and Mellini 2004).
123 These reversals bind the antigorite layers through strong, mainly covalent Si–O bonds (Evans et
124 al. 2013) (Fig. S1).

125 The samples in our dataset are mainly from three geological settings: ophiolites, modern
126 seafloor hydrothermal systems, and subduction zones (e.g., Sachan et al. 2007; Zhang et al. 2019;
127 Majumdar et al. 2020). The serpentine minerals within these samples have been accurately
128 characterized by X-ray diffraction, Raman spectroscopy, scanning electron microscopy, and/or
129 TEM (e.g., O'Hanley and Wicks 1995; Rouméjon et al. 2019; Shen et al. 2020), and their
130 compositions determined by electron microprobe analysis, X-ray fluorescence, and/or energy
131 dispersive spectrometry (e.g., Hirose et al. 2006; Wu et al. 2018). The concentrations of 12 oxides
132 (SiO₂, TiO₂, Al₂O₃, Cr₂O₃, FeO_{tot} [total iron expressed as FeO], MnO, NiO, MgO, CaO, Na₂O,
133 K₂O, and H₂O) are commonly reported for the major chemical compositions of serpentine minerals.
134 However, CaO, Na₂O, and K₂O contents are negligible because the large ionic radii of Ca²⁺, Na⁺,
135 and K⁺ make their substitutions for smaller cations (Mg²⁺ and Si⁴⁺) almost impossible. Thus, the
136 occurrence of CaO, Na₂O, and K₂O contents in these publications is most likely due to impure

137 materials (e.g., the presence of pyroxene), and we excluded these three oxides from our processed
138 dataset (Supplementary Table S2). We also excluded H₂O content because most studies did not
139 report loss on ignition for serpentine minerals.

140

141 **Data Processing**

142 To train the classification algorithm, we applied the XGBoost algorithm directly to the
143 unprocessed, labeled (L_Srp vs. H_Srp) dataset because it can handle missing data. However, we
144 did some processing to the original dataset before applying the *k*-means clustering (Fig. 1). (1)
145 Missing data imputation: missing data from the mineralogically labeled (L_Srp vs. H_Srp)
146 unprocessed dataset were imputed by the *k*-nearest neighbor (KNN) algorithm (Gheyas and Smith
147 2010). (2) Centered Log-Ratio (CLR) transformation: like all other compositional data, our dataset
148 suffers from data closure problems due to the multivariate nature of geochemical data (Aitchison
149 1982), which we chose to address by applying the CLR transformation (Aitchison 1982). Because
150 this transformation requires strictly positive values, we replaced any zero values with 0.0001. We
151 also reduced the complexity of the $1,375 \times 8$ data matrix by employing principal component
152 analysis (PCA). PCA measures interrelationships among multiple variables in numerous
153 dimensions using a covariance matrix (Ringnér 2008). We used the first three principal
154 components (explaining over 85% of the variance in the L_Srp and H_Srp datasets) to run the *k*-
155 means algorithm for visualization purposes. After applying the *k*-means algorithm, the resulting
156 datasets with predicted labels were named *L_Srp_km* and *H_Srp_km* (Supplementary Tables S3
157 and S4, respectively) (Fig. 1).

158

159 **Machine learning (ML) models**

160 Given a set of observations (here, serpentine chemical compositions), ML models classify
161 the dataset. We trained the classifier using the XGBoost, support vector machine (SVM), and
162 random forest (RF) algorithms and their results were similar (not shown). We chose the XGBoost
163 algorithm for our classification because it does not require imputation of missing values. Next, we
164 used both the *k*-means and Gaussian mixture algorithms to cluster the data; again, both algorithms
165 showed consistent results (not shown). Here, we showed and discussed only the clustering results
166 obtained using the well-known unsupervised learning approach of the *k*-means algorithm. As
167 mentioned above, we also applied PCA to the dataset before the *k*-means clustering to visualize
168 the results in three dimensions.

169

170 **Extreme Gradient Boosting (XGBoost).** XGBoost is a highly efficient and flexible end-
171 to-end boosting tree system (Chen and Guestrin 2016), improved by the gradient-boosting decision
172 tree. It uses a parallel computing strategy in the training process to maintain optimal calculation
173 speeds while adding a common term to the objective function. This prevents overfitting and
174 optimizes computing resources during the adjustment of the objective function value. Unlike using
175 feature vectors to calculate similarity, boosted trees are constructed to intelligently obtain feature
176 scores, revealing the importance of each feature to the training model. When a feature is used with
177 higher frequency to make critical decisions in boosted trees, that feature is given a higher score.
178 The principle of the algorithm is to continuously add trees to fit the residuals of the previous
179 prediction, summing the prediction scores of each tree to obtain the final score. More specific
180 information on the XGBoost algorithm is available in Chen and Guestrin (2016). We used the
181 open-source Python package (“xgboost”) to implement the algorithm.

182

183 **Model optimization with Scikit-learn.** Scikit-learn is a Python module integrating a wide
184 range of state-of-the-art machine learning algorithms for medium-scale supervised and
185 unsupervised problems (Pedregosa et al., 2011). Before training the model, the dataset was
186 randomly split into training (70%) and testing subsets (30%). The XGBoost algorithm uses
187 numerous hyperparameters that affect the performance of the final model. Here, we selected the
188 optimal hyperparameter combination by grid search, a typical parameter-tuning approach that
189 methodically builds and evaluates a model in a specific grid (Hsu et al. 2003). We also performed
190 tenfold cross-validation by building the prediction model with 90% of the training subset during
191 the grid search and subsequently using the model to predict the withheld 10%. This was performed
192 iteratively until all observations had been withheld and subsequently predicted. Tenfold cross-
193 validation is important for avoiding overfitting and evaluating the model's predictive capability
194 (Fig. S2). We performed the grid search over a 1,500-parameter space to identify the optimal
195 parameter combination and define the final classifier. Then, the classifier was evaluated based on
196 its performance when applied to the testing subset.

197

198 **Model evaluation with confusion matrix.** A confusion matrix (Supplementary Table S5)
199 contains information about the actual and predicted classifications, which is used to evaluate the
200 model performance. Based on the confusion matrix, four classification performance measures are
201 typically used, defined based on the total numbers of true positive (TP), false positive (FP), true
202 negative (TN), and false negative (FN) predictions: precision, recall, accuracy, and the traditional
203 F-measure or balanced F-score (F_1 score).

204 In this context, precision is a measure of accuracy, provided that a specific class has been
205 predicted:

$$206 \text{ Precision} = \text{TP}/(\text{TP} + \text{FP}) \quad (\text{Eq. 1})$$

207 Recall measures the ability of the model to select instances of a specific class from a dataset:

$$208 \text{ Recall} = \text{TP}/(\text{TP} + \text{FN}) \quad (\text{Eq.2})$$

209 Accuracy is a straightforward measurement of the proportion of the total number of predictions
210 that were correct:

$$211 \text{ Accuracy} = (\text{TP} + \text{TN})/(\text{TP} + \text{TN} + \text{FP} + \text{FN}) \quad (\text{Eq. 3})$$

212 The F_1 score is the harmonic mean of precision and recall:

$$213 F_1 \text{ score} = 2 * (\text{Precision} - \text{Recall}) / (\text{Precision} + \text{Recall}) = 2\text{TP}/(2\text{TP} + \text{TN} + \text{FP}) \quad (\text{Eq. 4})$$

214 The F_1 score is high when both precision and recall are high.

215

216 ***k*-means clustering.** The *k*-means clustering algorithm aims to partition a given dataset into
217 *k* clusters based on the distances between data points and cluster centroids (Zhou et al. 2018;
218 Ghezelbash et al. 2020). Because the naive *k*-means algorithm can be highly computationally
219 expensive and very sensitive to the initialization of *k* centroids, multiple variations have been
220 proposed to improve the algorithm's efficiency and accuracy. Here, we used the *k*-means++
221 algorithm, which adopts a smarter centroid initialization (Arthur and Vassilvitskii 2007) than the
222 naive *k*-means algorithm, which randomly chooses the initial *k* centroids. Generally, the
223 procedure is as follows:

- 224 1. The algorithm chooses the first centroid randomly from a given dataset.
- 225 2. The distance between each data point and the nearest centroid is calculated.

- 226 3. The data point farthest from the previously chosen centroid(s) is selected as the next new
227 centroid.
- 228 4. Steps 2 and 3 are iterated until k centroids are found.
- 229 5. Each data point is assigned to a cluster corresponding to the nearest centroid.
- 230 6. The k cluster centers are taken as k new centroids (cluster centers) when Step 5 is
231 completed.
- 232 7. Steps 5 and 6 are iterated until the positions of the centroids remain fixed between
233 subsequent iterations.

234

235

RESULTS

236 Chemical compositions of serpentine minerals

237 The two classes of serpentine-group minerals have different compositional patterns (Fig.
238 2). H_Srp have systematically higher SiO₂, whereas L_Srp spans a wide range of SiO₂ contents
239 (36.0–46.0 wt%). Based on their higher Al concentrations, Al substitution into crystal sites is
240 slightly more frequent in H_Srp than in L_Srp. Furthermore, although both H_Srp and L_Srp span
241 similar ranges of FeO_{tot} contents (0.0–12.5 wt%), >75% of H_Srp contain ≤5.0 wt% FeO_{tot}. H_Srp
242 tend to contain less MnO and NiO than L_Srp. H_Srp MgO contents are concentrated in the range
243 37.0–39.0 wt%, whereas those of most L_Srp are more broadly distributed over 37.0–41.0 wt%.
244 The distributions of TiO₂ (0–0.2 wt%) and Cr₂O₃ contents (0.0–1.3 wt%) are almost identical in
245 H_Srp and L_Srp.

246

247 **XGBoost model**

248 **Confusion matrix and performance evaluation.** The raw confusion matrix of true
249 (known) and predicted L_Srp and H_Srp labels after applying the XGBoost algorithm to the test
250 dataset (i.e., 30% of the unprocessed dataset) is shown in Figure 3a and normalized to the known
251 total number of analyses with each label in Figure 3b (i.e., the rows of the matrix sum to 1). The
252 classification results indicate an overall accuracy of 87.2% (Table 1), with 86.2% of L_Srp and
253 87.4% of H_Srp correctly classified (Fig. 3b). Of the 160 L_Srp examples in the test dataset, 22
254 were falsely predicted as H_Srp, and 31 of the 253 H_Srp examples were falsely classified as
255 L_Srp (Fig. 3a). These results indicate that the ML model provides reasonably accurate
256 classifications of serpentine minerals.

257

258 **Feature importance.** Here, ‘feature’ refers to an elemental concentration, and feature
259 importances reflect the relative influences of constituent elements on the model predictions. To
260 identify the main features controlling the performance of the XGBoost algorithm in classifying
261 serpentines, we calculated feature importances for the eight elemental concentrations considered.
262 SiO₂ content had the highest feature importance score (0.29), implying its importance in
263 classifying serpentine minerals (Fig. 3c). The classification was also fairly sensitive to NiO content,
264 with a feature importance score of 0.15. Therefore, the classification of serpentine minerals
265 performs well when SiO₂ and NiO are used to evaluate the models. Al₂O₃, MgO, FeO_{tot}, MnO, and
266 TiO₂ contents had similar but weaker influences on the classification than NiO content (Fig. 3c).

267 ***k*-means model**

268 In our dataset, fewer than 10 analyses were available for some locations, leading to non-
269 representative clustering. Therefore, we excluded these analyses before applying the *k*-means

270 clustering model. We trained separate k -means models for the L_Srp and H_Srp samples and used
271 their silhouette scores (measuring the mean distance between each sample to its cluster centroid)
272 to evaluate their performances. As shown in Figure S3, the clustering results for L_Srp and H_Srp
273 samples have maximum respective silhouette scores of 0.57 and 0.54 when $k = 4$, making this the
274 optimum number of clusters. In addition, four well-separated groups are observed in the PCA plots
275 of the first three PCs (Fig. 4). These results indicate that the datasets were well separated into four
276 clusters.

277 Of the L_Srp analyses, ~68.0% were classified into Cluster L_1 and are globally distributed
278 (Fig. 5a). Furthermore, analyses from most locations fall into multiple clusters, with the exception
279 of the landward slope of the Middle America Trench off Guatemala (Fig. 5a), for which all
280 analyses were classified as Cluster L_1. Cluster L_2 of L_Srp only spread in four locations,
281 including New Caledonia, NW Iran, the Eastern Desert of Egypt, and NE Corsica (Fig. 5a). These
282 k -means results imply that analyses from British Columbia Canada, Elba Island, New Caledonia,
283 and the North Qaidam orogen (Tibet) are characterized by only two meaningful classes, whereas
284 analyses from the other eight areas (i.e., excluding Guatemala) are characterized by three or four
285 clusters (Fig. 5a). We note that the classification of most of these analyses as Cluster L_1 indicates
286 that the major chemical compositions of chrysotile and lizardite do not strongly reflect their
287 sampling locations.

288 Of the H_Srp analyses, 87.9% were classified as either Cluster H_1 or H_2 (Fig. 5b; note that
289 Clusters H_1–4 are distinct from Clusters L_1–4). NW Iran was the only location without any
290 samples classified as Cluster H_2. Analyses from Cuba, the Hess Deep, and Japan were uniquely
291 classified as Cluster H_1. Only analyses from New Caledonia and Qinling China were categorized
292 into just two clusters: Clusters H_2 and H_4 and Clusters H_2 and H_3, respectively (Fig. 5b).

293 The remaining nine locations showed broadly similar clustering distributions (Fig. 5b), implying
294 that the compositional clustering of H_Srp (antigorite) is not largely due to the chemical
295 environment in which is formed.

296

297

DISCUSSION

298 **Classification of serpentine minerals**

299 Raman analyses are currently the most widely used traditional analytical method for
300 discriminating between serpentine minerals (Rinaudo et al. 2003). The presence of peaks at ~1050
301 cm^{-1} and 3680-3700 cm^{-1} , representing antisymmetric Si-O_b-Si and OH stretching, respectively,
302 usually indicates that a polymorph is an antigorite (Auzende et al. 2004; Petriglieri et al. 2015),
303 whereas chrysotile and lizardite are identified based on the presence of a peak attributed to
304 antisymmetric stretching modes of Si-O_{nb} groups at ~1100 cm^{-1} (Groppo et al. 2006). However,
305 these peaks may be absent in poorly crystallized serpentines. Valid peaks may also not occur if the
306 sample was first analyzed by electron microprobe analysis which could destroy its structure by the
307 electron beam. Another problem in most applications of Raman spectroscopy is the strong
308 fluorescence background (Wei et al., 2015). Therefore, the high quality of the sample and
309 appropriate methods to suppress the fluorescence background are required to collect useful Raman
310 results. In contrast, our quantitative ML model can distinguish between serpentine minerals based
311 only on their major chemical compositions. Our model can distinguish between L_Srp and H_Srp
312 with 87.2% accuracy (Table 1), which is far more reliable than traditional geochemical methods.

313 Our binary classification model sometimes misclassified serpentines (Figs. 3a, 3b), possibly
314 due to mineral replacements among the serpentine minerals. Antigorite is more stable than
315 chrysotile and lizardite at high temperatures (~250 to >500 °C; O'Hanley and Wicks 1995; Ghaderi
316 et al. 2015), and usually forms through prograde and retrograde metamorphic reactions during

317 subduction and obduction, respectively (Guillot et al. 2001; Boudier et al. 2010; Wu et al. 2018;
318 Zhang et al. 2019). Field observations and laboratory evidence suggest that chrysotile and lizardite
319 can precede replacement by antigorite during prograde metamorphism (Koutsovitis 2017; Plissart
320 et al. 2019), which is associated with the formation of brucite or a modest influx of SiO₂ (e.g.,
321 O'Hanley and Wicks 1995; Evans 2010; Rouméjon et al. 2019). These two mineral replacement
322 processes likely explain why antigorite contains more Si than lizardite and chrysotile (Fig. 2a),
323 and are probably why the classification between H_Srp (antigorite) with L_Srp (chrysotile and
324 lizardite) performed so well.

325 These same transformations also imply that the composition of antigorite could be inherited
326 from previous chrysotile and/or lizardite, which may explain the rare misclassification of H_Srp
327 as L_Srp in our model. Furthermore, Evans (2004) suggested that, although serpentinization may
328 force the crystallization of chrysotile, a considerable amount of chrysotile recrystallizes in
329 antigorite-bearing serpentinites surrounded by abundant Si- and Mg-oversaturated fluids.
330 Therefore, chrysotile may also replace antigorite without any associated compositional change
331 (Evans 2004; Ulrich et al. 2020), which may explain the rare misclassification of L_Srp as H_Srp
332 in our model (Figs. 3a, 3b).

333 Our feature importance results indicate that the chemical differences between serpentine
334 minerals mainly arise from their SiO₂ contents (Fig. 3c). Consistent with previous results (e.g.,
335 Deschamps et al. 2013) and the SiO₂ distribution plot (Fig. 2), antigorite samples contain more
336 SiO₂ than lizardite and chrysotile. Indeed, the tetrahedral sheets with periodic reversals along the
337 *a*-axis in antigorite predestine the formation of antigorite to consume more SiO₂ than the formation
338 of chrysotile and lizardite. NiO contents were also important to the performance of our
339 classification (Fig. 3c): chrysotile and lizardite span a fairly wide range of NiO concentrations (0–

340 0.6 wt%), whereas the NiO contents of antigorite are more restricted to the range 0–0.2 wt%. This
341 result is consistent with the conclusions of McCollom and Bach (2009), who identified the
342 formation of Fe-Ni alloy during the replacement of chrysotile and lizardite by antigorite, indicating
343 that Ni migrates from serpentine to the surrounding environment. Al₂O₃ was the third-most
344 important feature for classifying serpentine minerals (Fig. 3c). The distribution plot reveals that,
345 although serpentines have a wide range of Al₂O₃ contents (0.0–6.7 wt%), those of H_Srp are
346 mainly restricted to the range 0.4–3.0 wt%, slightly above the range for L_Srp (Fig. 2c). Because
347 Al generally substitutes for Si in the tetrahedral sheets of serpentine, the higher Al₂O₃ contents of
348 H_Srp are the consequence of the higher SiO₂ contents of H_Srp. Unexpectedly, the substitutions
349 of Fe, Mn, Ti, and Cr into the structures of serpentine minerals had only minimal contributions to
350 the classification (feature importances of 0.09, 0.09, 0.09, and 0.07, respectively; Fig. 3c). The
351 different impacts of these substitutions suggest that Ni and Al equilibria between serpentine
352 minerals and their surrounding environments depend on the minerals' structures and formation
353 conditions, whereas those involving Fe, Mn, Ti, and Cr do not.

354

355 **Application of the *k*-means algorithm to determine the environmental conditions of** 356 **crystallization**

357 Our clustering model can be applied to determine the original geological environments in
358 which serpentine crystallized. The derived silhouette coefficients of the *k*-means model
359 performances for L_Srp (chrysotile and lizardite) indicate that four clusters best classified the data
360 (Fig. 5a). Kodolányi and Pettke (2011) confirmed that the alteration of oceanic crust is dominant
361 in Guatemala, and no high temperature (>300–400 °C) alteration phases have been detected there,
362 indicating that Guatemala chrysotile and lizardite formed at low temperatures (<300 °C).

363 Therefore, we consider that the temperature during the formation of L_Srp in Cluster L_1 is
364 constrained to <300 °C, consistent with previous field observations and experimental results
365 (Evans 2004; Schwartz et al. 2013). Cluster L_1 may also represent L_Srp that were not modified
366 by subduction zones because L_Srp tends to crystallize during water-rock interactions on the
367 seafloor, in the absence of intense tectonic activity and major shear zones (Federico et al. 2005;
368 Putlitz et al. 2005; Obeid et al. 2016; Imayama et al. 2020). Because chrysotile and lizardite tend
369 to crystallize in such geological environments, the classification of ~68.0% of L_Srp into Cluster
370 L_1 is consistent with these observations (Fig. 5a). L_Srp samples in Cluster L_1 occur in all
371 locations listed in our L_Srp dataset, indicating that it is not practical to attempt to distinguish
372 which L_Srp phase is produced upon alteration of oceanic crust on the seafloor at a given sampling
373 location based on chemical composition alone. Furthermore, the hydration of ultramafic or mafic
374 rocks by meteoric water at Earth's surface (Zhang et al. 2019; Ulrich et al. 2020), the equilibration
375 of fluids with shallow subducting sediments (Wu et al. 2018), or upwelling fluids sourced from
376 subducted slabs (Martin et al. 2020) dehydrated under sub-greenschist conditions can also produce
377 both chrysotile and lizardite. Unfortunately, we failed to assign any of this information to Clusters
378 L_2–4. The incorporation of more data, including isotopic results (e.g., $\delta^{11}\text{B}$, $\delta^{18}\text{O}$) should improve
379 our models, allowing us to constrain specific *P-T* conditions and explore the evolution of
380 geological processes and elemental cyclings (Zhang et al. 2019; Martin et al. 2020).

381 We also obtained four clusters by applying the *k*-means algorithm to the H_Srp chemical
382 compositions (Fig. 5b). Ribeiro da Costa et al. (2008) suggested that antigorite at the Mid-Atlantic
383 ridge may result from extensive deformation at low temperature rather than high-temperature
384 prograde metamorphism. Rouméjon et al. (2019) further proposed that the presence of a Si-rich
385 fluid is the only requirement for the crystallization of antigorite. Indeed, antigorite formed under

386 conditions unrelated to subduction was classified into Cluster H_2 in our study (Fig. 5b). In
387 contrast, antigorite produced by blueschist to eclogite facies metamorphism (i.e., 390–650 °C) was
388 classified into Clusters H_1, H_3, and H_4 (Lafay et al. 2013; Vitale Brovarone and Herwartz
389 2013; Martin et al. 2020; Ulrich et al. 2020). Antigorite from Tibet, China, and Iran may have
390 undergone metamorphism at ultra-high pressure (21–35 kbar), but at temperatures not exceeding
391 700 °C (Zhang et al. 2019; Moghadam and Stern 2015), indicating that the stability of antigorite
392 is more sensitive to temperature than pressure. Nonetheless, the prograde recrystallization of
393 chrysotile and/or lizardite to form some of the subduction-related H_Srp (Rouméjon et al. 2019;
394 Martin et al. 2020) may result in misclassifications between L_Srp and H_Srp, consistent with our
395 XGBoost results (Fig. 3).

396 For both L_Srp and H_Srp, samples from different locations can be classified into the same
397 cluster, suggesting that the specific sampling location has a negligible effect on compositional
398 clustering. However, our results confirm that the geological environment determines the serpentine
399 species and chemical composition. We note that our dataset only contains 1,375 analyses, limiting
400 our efforts to assess the *P-T* conditions required to form serpentine of specific chemical
401 compositions. Nevertheless, subduction-related serpentines seem to contain less TiO₂, Al₂O₃, and
402 Cr₂O₃ (Figs. S4 and S5).

403

404

IMPLICATIONS

405 Serpentinization occurs ubiquitously in seafloor ultramafic rocks and subduction zones,
406 exerting a strong influence on the geochemistry of those systems (Schmidt et al. 2011; Schroeder
407 et al. 2015; Seyfried et al. 2015; Scott et al. 2017; Debret et al. 2019). Serpentine-group minerals
408 can structurally accommodate water and elements, and minerals thus play a significant role in the

409 alteration process (Evans 2008; Hacker 2008; Iyer et al. 2008; Jöns et al. 2010; Schrenk et al. 2013;
410 Canales et al. 2017; Mayhew et al. 2018). Therefore, the transformation from L_Srp to H_Srp
411 promotes elemental cycling (Malvoisin 2015; Mayhew et al. 2018). Our feature importance data
412 and the distributions of serpentine chemical compositions (Figs. 2a, 3c) indicate that the
413 enrichment of Si in serpentine is associated with subduction-generated Si-rich fluids. This sheds
414 light on the factors controlling the solubility of hydrous melts and dilute aqueous solutions in
415 which the dissolved silicate components also play significant roles (Manning 2004). Our feature
416 importance data indicate that Ni migrates from serpentine to the surrounding environment during
417 mineral replacement (Figs. 2g, 3c) and may be consumed by Fe-Ni alloy when dominated by
418 ferrous iron (McCollom and Bach 2009), redistributing Fe among the solids. In contrast, Al
419 diffuses into H_Srp during replacement (Figs. 2c, 3c) in association with the increase in SiO₂
420 content (Figs. 2a, 3c). This indicates that Al tends to substitute for Si in the tetrahedral sheets
421 (Evans et al. 2013), which may stabilize antigorite at higher temperatures (Padrón-Navarta et al.
422 2013). Our XGBoost model thus documents the importance of serpentine mineral replacements
423 during mass transfers in subduction zones and relates serpentine chemical compositions to their
424 geological environments, whereas traditional geochemical methods have limited capabilities to
425 provide such information on alteration chemistry (e.g., Barbier et al. 2020).

426 The dataset used to train these ML algorithms only contains 1,375 analyses for which the
427 concentrations of the eight analyzed elements were reported. Due to the lack of *P-T* data, our
428 discussion on serpentine crystallization and geological environments is qualitative. Model
429 performance will improve when more data becomes available. Similar ML algorithms could be
430 applied to track the sources of fluids participating in the formation of serpentine minerals by
431 incorporating ⁸⁷Sr/⁸⁶Sr data to further constrain the evolution of metasomatic processes (Consuma

432 et al. 2020). Additional data on $\text{Fe}^{2+}/\text{Fe}^{3+}$ ratios and B isotopes will be useful for classifying the
433 redox state of serpentine minerals and revealing the influence of mineral replacements on the redox
434 budget during subduction (Zhang et al. 2019, 2021). Furthermore, applying such ML algorithms
435 to a dataset incorporating the P - T conditions of serpentine crystallization, the age of the lithosphere,
436 and convergence rates will improve our understanding of Earth's tectonic history (Blanco-
437 Quintero et al. 2011; Wu et al. 2018; Cannà et al. 2020).

438

439

440

DATA AVAILABILITY STATEMENT

441 Python scripts for our ML classification and clustering models are available on Zenodo at
442 <https://doi.org/10.5281/zenodo.7551437>. After setting up a python environment with the essential
443 packages installed, the models can be run on Windows, Mac OS X, or Linux.

444

445

446

ACKNOWLEDGMENTS

447 This work was supported by Fundamental Research Funds for the Central Universities (Grant
448 No. 226-2022-00176) and startup funding to ZhangZhou J. We acknowledge Robert Dennen for
449 polishing the language of the paper.

450

451

REFERENCES CITED

452 Aitchison, J. (1982) The Statistical Analysis of Compositional Data. Journal of the Royal Statistical Society:
453 Series B (Methodological), 44(2), 139–160.
454 Arthur, D., Vassilvitskii, S. (2007) k means++: The advantages of careful seeding. Proceedings of the
455 Eighteenth Annual ACM-SIL Symposium on Discrete Algorithms. Philadelphia, USA: SIAM,
456 1027-1035.

- 457 Auzende, A.-L., Devouard, B., Guillot, S., Daniel, I., Baronnet, A., and Lardeaux, J.-M. (2002)
458 Serpentinites from Central Cuba: petrology and HRTEM study. *European Journal of Mineralogy*,
459 14(5), 905–914.
- 460 Auzende, A.-L., Daniel, I., Reynard, B., Lemaire, C., and Guyot, F. (2004) High-pressure behaviour of
461 serpentine minerals: a Raman spectroscopic study. *Physics and Chemistry of Minerals*, 31, 269-
462 277.
- 463 Auzende, A.-L., Guillot, S., Devouard, B., and Baronnet, A. (2006) Serpentinites in an Alpine convergent
464 setting: Effects of metamorphic grade and deformation on microstructures. *European Journal of*
465 *Mineralogy*, 18(1), 21–33.
- 466 Banfield, J.F., Bailey, S.W., Barker, W.W., and Smith, R.C. (1995) Complex polytypism; relationships
467 between serpentine structural characteristics and deformation. *American Mineralogist*, 80(11–12),
468 1116–1131.
- 469 Barbier, S., Huang, F., Andreani, M., Tao, R., Hao, J., Eleish, A., Prabhu, A., Minhas, O., Fontaine, K.,
470 Fox, P., and Minhas, O. (2020) A Review of H₂, CH₄, and Hydrocarbon Formation in Experimental
471 Serpentinization Using Network Analysis. *Frontiers in Earth Science*, 8, 209.
- 472 Blanco-Quintero, I.F., Rojas-Agramonte, Y., García-Casco, A., Kröner, A., Mertz, D.F., Lázaro, C.,
473 Blanco-Moreno, J., and Renne, P.R. (2011) Timing of subduction and exhumation in a subduction
474 channel: Evidence from slab melts from La Corea Mélange (eastern Cuba). *Lithos*, 127, 86–100.
- 475 Boskabadi, A., Pitcairn, I.K., Leybourne, M.I., Teagle, D.A.H., Cooper, M.J., Hadizadeh, H., Bezenjani,
476 R.N., and Bagherzadeh, R.M. (2020) Carbonation of ophiolitic ultramafic rocks: Listvenite
477 formation in the Late Cretaceous ophiolites of eastern Iran. *Lithos*, 352–353, 105307.
- 478 Boudier, F., Baronnet, A., and Mainprice, D. (2010) Serpentine Mineral Replacements of Natural Olivine
479 and their Seismic Implications: Oceanic Lizardite versus Subduction-Related Antigorite. *Journal*
480 *of Petrology*, 51(1–2), 495–512.
- 481 Canales, J.P., Carbotte, S.M., Nedimović, M.R., and Carton, H. (2017) Dry Juan de Fuca slab revealed by
482 quantification of water entering Cascadia subduction zone. *Nature Geoscience*, 10(11), 864–870.
- 483 Cannaò, E., Scambelluri, M., Bebout, G.E., Agostini, S., Pettke, T., Godard, M., and Crispini, L. (2020)
484 Ophicarbonates evolution from seafloor to subduction and implications for deep-Earth C cycling.
485 *Chemical Geology*, 546, 119626.
- 486 Capitani, G., and Mellini, M. (2004) The modulated crystal structure of antigorite: The m = 17 polysome.
487 *American Mineralogist*, 89(1), 147-158.
- 488 Caricchi, L., Petrelli, M., Bali, E., Sheldrake, T., Pioli, L., and Simpson, G. (2020) A Data Driven Approach
489 to Investigate the Chemical Variability of Clinopyroxenes From the 2014–2015 Holuhraun–
490 Bárðarbunga Eruption (Iceland). *Frontiers in Earth Science*, 8, 18.

- 491 Chen, T., and Guestrin, C. (2016) XGBoost: A Scalable Tree Boosting System. In Proceedings of the 22nd
492 ACM SIGKDD International Conference on Knowledge Discovery and Data Mining (pp. 785–
493 794). San Francisco California USA: ACM.
- 494 Consuma, G., Braga, R., Giovanardi, T., Bersani, D., Konzett, J., Lugli, F., Mazzucchelli, M., and Tropper,
495 P. (2020) In situ Sr isotope analysis of mantle carbonates: Constraints on the evolution and sources
496 of metasomatic carbon-bearing fluids in a paleo-collisional setting. *Lithos*, 354–355, 105334.
- 497 Debret, B., Albers, E., Walter, B., Price, R., Barnes, J.D., Beunon, H., Facq, S., Gillikin, D.P., Mattielli, N.,
498 and Williams, H. (2019) Shallow forearc mantle dynamics and geochemistry: New insights from
499 IODP Expedition 366. *Lithos*, 326–327, 230–245.
- 500 Deschamps, F., Guillot, S., Godard, M., Chauvel, C., Andreani, M., and Hattori, K. (2010) In situ
501 characterization of serpentinites from forearc mantle wedges: Timing of serpentinization and
502 behavior of fluid-mobile elements in subduction zones. *Chemical Geology*, 269(3), 262–277.
- 503 Deschamps, F., Godard, M., Guillot, S., and Hattori, K. (2013) Geochemistry of subduction zone
504 serpentinites: A review. *Lithos*, 178, 96–127.
- 505 Evans, B.W. (2004) The Serpentine Multisystem Revisited: Chrysotile Is Metastable. *International
506 Geology Review*, 46, 479–506.
- 507 ——— (2008) Control of the Products of Serpentinization by the $\text{Fe}^{2+}\text{Mg}_{-1}$ Exchange Potential of Olivine
508 and Orthopyroxene. *Journal of Petrology*, 49, 1873–1887.
- 509 ——— (2010) Lizardite versus antigorite serpentinite: Magnetite, hydrogen, and life(?). *Geology*, 38, 879–
510 882.
- 511 Evans, B.W., Hattori, K., and Baronnet, A. (2013) Serpentine: What, Why, What? *Element*, 9, 99-106.
- 512 Federico, L., Capponi, G., Crispini, L., Scambelluri, M., and Villa, I. (2005) $^{39}\text{Ar}/^{40}\text{Ar}$ dating of high-
513 pressure rocks from the Ligurian Alps: Evidence for a continuous subduction–exhumation cycle.
514 *Earth and Planetary Science Letters*, 240(3–4), 668–680.
- 515 Ghaderi, N., Zhang, H., and Sun, T. (2015) Relative stability and contrasting elastic properties of serpentine
516 minerals from first-principles calculations. *Journal of Geophysical Research: Solid Earth*, 120(7),
517 4831–4842.
- 518 Gheyas, I.A., and Smith, L.S. (2010) A neural network-based framework for the reconstruction of
519 incomplete data sets. *Neurocomputing*, 73(16–18), 3039–3065.
- 520 Ghezelbash, R., Maghsoudi, A., and Carranza, E.J.M. (2020) Optimization of geochemical anomaly
521 detection using a novel genetic K-means clustering (GKMC) algorithm. *Computers and
522 Geosciences*, 134, 104335.

- 523 Groppo, C., Rinaudo, C., Cairo, S., Gastaldi, D., and Compagnoni, R. (2006) Micro-Raman spectroscopy
524 for a quick and reliable identification of serpentine minerals from ultramafics. *European Journal of*
525 *Mineralogy*, 18(3), 319–329.
- 526 Guillot, S., Hattori, K.H., de Sigoyer, J., Nägler, T., and Auzende, A.-L. (2001) Evidence of hydration of
527 the mantle wedge and its role in the exhumation of eclogites. *Earth and Planetary Science Letters*,
528 193(1–2), 115–127.
- 529 Hacker, B.R. (2008) H₂O subduction beyond arcs. *Geochemistry, Geophysics, Geosystems*, 9(3), Q03001.
- 530 Hattori, K.H., and Guillot, S. (2007) Geochemical character of serpentinites associated with high- to
531 ultrahigh-pressure metamorphic rocks in the Alps, Cuba, and the Himalayas: Recycling of elements
532 in subduction zones: *Geochemistry of serpentinites. Geochemistry, Geophysics, Geosystems*, 8(9),
533 Q09010.
- 534 Hazen, R.M., Downs, R.T., Eleish, A., Fox, P., Gagné, O.C., Golden, J.J., Grew, E.S., Hummer, D.R.,
535 Hystad, G., Krivovichev, S.V., and others. (2019) Data-Driven Discovery in Mineralogy: Recent
536 Advances in Data Resources, Analysis, and Visualization. *Engineering*, 5(3), 397–405.
- 537 Hirauchi, K., and Katayama, I. (2013) Rheological contrast between serpentine species and implications
538 for slab-mantle wedge decoupling. *Tectonophysics*, 608, 545–551.
- 539 Hirose, T., Bystricky, M., Kunze, K., and Stünitz, H. (2006) Semi-brittle flow during dehydration of
540 lizardite–chrysotile serpentinite deformed in torsion: Implications for the rheology of oceanic
541 lithosphere. *Earth and Planetary Science Letters*, 249(3–4), 484–493.
- 542 Hsu, C., Chang, C., and Lin, C. (2003) A practical guide to support vector classification. Technical report,
543 Department of Computer Science and Information Engineering (p. 16), National Taiwan University,
544 Taipei.
- 545 Imayama, T., Ueda, H., Usuki, T., Minami, M., Asahara, Y., and Nagahashi, T. (2020) Variability of
546 protoliths and pressure-temperature conditions of amphibolites from the Ohmachi Seamount (Izu-
547 Bonin-Mariana arc): evidence of a fossil subduction channel in a modern intra-oceanic arc.
548 *Mineralogy and Petrology*, 114, 305–318.
- 549 Iyer, K., Austrheim, H., John, T., and Jamtveit, B. (2008) Serpentinization of the oceanic lithosphere and
550 some geochemical consequences: Constraints from the Leka Ophiolite Complex, Norway.
551 *Chemical Geology*, 249(1–2), 66–90.
- 552 Ji, S., Zhu, J., He, H., Tao, Q., Zhu, R., Ma, L., Chen, M., Li, S., and Zhou, J. (2018) Conversion of
553 serpentine to smectite under hydrothermal condition: Implication for solid-state transformation.
554 *American Mineralogist*, 103(2), 241–251.

- 555 John, T., Scambelluri, M., Frische, M., Barnes, J.D., and Bach, W. (2011) Dehydration of subducting
556 serpentinite: Implications for halogen mobility in subduction zones and the deep halogen cycle.
557 Earth and Planetary Science Letters, 308(1), 65–76.
- 558 Jöns, N., Bach, W., and Klein, F. (2010) Magmatic influence on reaction paths and element transport during
559 serpentinization. Chemical Geology, 274(3), 196–211.
- 560 Kodolányi, J., and Pettke, T. (2011) Loss of trace elements from serpentinites during fluid-assisted
561 transformation of chrysotile to antigorite — An example from Guatemala. Chemical Geology,
562 284(3–4), 351–362.
- 563 Koutsovitis, P. (2017) High-pressure subduction-related serpentinites and metarodingites from East
564 Thessaly (Greece): Implications for their metamorphic, geochemical and geodynamic evolution in
565 the Hellenic–Dinaric ophiolite context. Lithos, 276, 122–145.
- 566 Lafay, R., Deschamps, F., Schwartz, S., Guillot, S., Godard, M., Debret, B., and Nicollet, C. (2013) High-
567 pressure serpentinites, a trap-and-release system controlled by metamorphic conditions: Example
568 from the Piedmont zone of the western Alps. Chemical Geology, 343, 38–54.
- 569 Majumdar, A.S., Ray, D., and Shukla, A.D. (2020) Serpentinization of olivine–gabbro in Central Indian
570 ridge: Insights into H₂ production during alteration in lower oceanic crust and sustenance of life at
571 slow–spreading ridges. Lithos, 374–375, 105730.
- 572 Malvoisin, B. (2015) Mass transfer in the oceanic lithosphere: Serpentinization is not isochemical. Earth
573 and Planetary Science Letters, 430, 75–85.
- 574 Manning, C.E. (2004) The chemistry of subduction-zone fluids. Earth and Planetary Science Letters, 224,
575 1–14.
- 576 Martin, C., Flores, K.E., Vitale-Brovarone, A., Angiboust, S., and Harlow, G.E. (2020) Deep mantle
577 serpentinization in subduction zones: Insight from in situ B isotopes in slab and mantle wedge
578 serpentinites. Chemical Geology, 545, 119637.
- 579 Mayhew, L.E., Ellison, E.T., Miller, H.M., Kelemen, P.B., and Templeton, A.S. (2018) Iron
580 transformations during low temperature alteration of variably serpentinized rocks from the Samail
581 ophiolite, Oman. Geochimica et Cosmochimica Acta, 222, 704–728.
- 582 McCollom, T.M., and Bach, W. (2009) Thermodynamic constraints on hydrogen generation during
583 serpentinization of ultramafic rocks. Geochimica et Cosmochimica Acta, 73(3), 856–875.
- 584 Mellini, M. (1986) Chrysotile and polygonal serpentine from the Balangero serpentinite. Mineralogical
585 Magazine, 50(356), 301–305.
- 586 Moghadam, H.S., and Stern, R.J. (2015) Ophiolites of Iran: Keys to understanding the tectonic evolution
587 of SW Asia: (II) Mesozoic ophiolites. Journal of Asian Earth Sciences, 100, 31–59.

- 588 Obeid, M.A., Khalil, A.E.S., and Azer, M.K. (2016) Mineralogy, geochemistry, and geotectonic
589 significance of the Neoproterozoic ophiolite of Wadi Arais area, south Eastern Desert, Egypt.
590 International Geology Review, 58(6), 687–702.
- 591 O'Hanley, D.S., and Dyar, M.D. (1998) The composition of chrysotile and its relationship with lizardite.
592 The Canadian Mineralogist, 36, 727–739.
- 593 O'Hanley, D.S., and Wicks, F.J. (1995) Conditions of formation of lizardite, chrysotile and antigorite,
594 Cassiar, British Columbia. The Canadian Mineralogist, 33, 753-773.
- 595 O'Sullivan, G., Chew, D., Kenny, G., Henrichs, I., and Mulligan, D. (2020) The trace element composition
596 of apatite and its application to detrital provenance studies. Earth-Science Reviews, 201, 103044.
- 597 Padregosa, E., Varoquaux, G., Gramfort, A., Michel, V., Thirion, B., Grisel, O., Blondel, M., Prettenhofer,
598 P., Weiss, R., Dubourg, V., Vanderpl, J., Passos, A., Cournapeau, D., Brucher, M., Perrot, M., and
599 Duchesnay, É. (2011) Scikit-learn: Machine Learning in Python. Journal of Machine Learning
600 Research, 12, 2825-2830.
- 601 Padrón-Navarta, J.A., Sánchez-Vizcaíno, V.L., Hermann, J., Connolly, J.A.D., Garrido, C.J., Gómez-
602 Pugnaire, M.T., and Marchesi, C. (2013) Tschermak's substitution in antigorite and consequences
603 for phase relations and water liberation in high-grade serpentinites. Lithos, 178, 186–196.
- 604 Petrelli, M., and Perugini, D. (2016) Solving petrological problems through machine learning: the study
605 case of tectonic discrimination using geochemical and isotopic data. Contributions to Mineralogy
606 and Petrology, 171(10), 81.
- 607 Petrelli, M., Caricchi, L., and Perugini, D. (2020) Machine Learning Thermo-Barometry: Application to
608 Clinopyroxene-Bearing Magmas. Journal of Geophysical Research: Solid Earth, 125,
609 e2020JB020130.
- 610 Petriglieri, J.R., Salvioli-Mariani, E., Mantovani, L., Tribaudino, M., Lottici, P.P., Laporte-Magoni, C., and
611 Bersani, D. (2015) Micro-Raman mapping of the minerals of serpentine: Micro-Raman mapping
612 of the minerals of serpentine. Journal of Raman Spectroscopy, 46(10), 953–958.
- 613 Plissart, G., González-Jiménez, J.M., Garrido, L.N.F., Colás, V., Berger, J., Monnier, C., Diot, H., and
614 Padrón-Navarta, J.A. (2019) Tectono-metamorphic evolution of subduction channel serpentinites
615 from South-Central Chile. Lithos, 336–337, 221–241.
- 616 Plümpner, O., John, T., Podladchikov, Y.Y., Vrijmoed, J.C., and Scambelluri, M. (2017) Fluid escape from
617 subduction zones controlled by channel-forming reactive porosity. Nature Geoscience, 10(2), 150–
618 156.
- 619 Putlitz, B., Cosca, M.A., and Schumacher, J.C. (2005) Prograde mica $^{40}\text{Ar}/^{39}\text{Ar}$ growth ages recorded in
620 high pressure rocks (Syros, Cyclades, Greece). Chemical Geology, 214(1–2), 79–98.
- 621 Reynard, B. (2013). Serpentine in active subduction zones. Lithos, 178, 171–185.

- 622 Ribeiro da Costa, I., Barriga, F.J.A.S., Viti, C., Mellini, M., and Wicks, F.J. (2008) Antigorite in deformed
623 serpentinites from the Mid-Atlantic Ridge. *European Journal of Mineralogy*, 20, 563-572.
- 624 Rinaudo, C., Gastaldi, D., and Belluso, E. (2003) Characterization of chrysotile, antigorite and lizardite by
625 FT-Raman spectroscopy. *The Canadian Mineralogist*, 41(4), 883–890.
- 626 Ringnér, M. (2008) What is principal component analysis? *Nature Biotechnology*, 26(3), 303–304.
- 627 Rouméjon, S., Andreani, M., and Früh-Green, G.L. (2019) Antigorite crystallization during oceanic
628 retrograde serpentinization of abyssal peridotites. *Contributions to Mineralogy and Petrology*,
629 174(7), 60.
- 630 Rüpke, L.H., Morgan, J.P., Hort, M., and Connolly, J.A.D. (2004) Serpentine and the subduction zone water
631 cycle. *Earth and Planetary Science Letters*, 223(1), 17–34.
- 632 Sachan, H.K., Mukherjee, B.K., and Bodnar, R.J. (2007) Preservation of methane generated during
633 serpentinization of upper mantle rocks: Evidence from fluid inclusions in the Nidar ophiolite, Indus
634 Suture Zone, Ladakh (India). *Earth and Planetary Science Letters*, 257(1–2), 47–59.
- 635 Schmidt, K., Garbe-Schönberg, D., Koschinsky, A., Strauss, H., Jost, C.L., Klevenz, V., and Königer, P.
636 (2011) Fluid elemental and stable isotope composition of the Nibelungen hydrothermal field
637 (8°18'S, Mid-Atlantic Ridge): Constraints on fluid–rock interaction in heterogeneous lithosphere.
638 *Chemical Geology*, 280(1), 1–18.
- 639 Schrenk, M.O., Brazelton, W.J., and Lang, S.Q. (2013) Serpentinization, Carbon, and Deep Life. *Reviews*
640 *in Mineralogy and Geochemistry*, 75(1), 575–606.
- 641 Schroeder, T., Bach, W., Jöns, N., Jöns, S., Monien, P., and Klügel, A. (2015) Fluid circulation and
642 carbonate vein precipitation in the footwall of an oceanic core complex, Ocean Drilling Program
643 Site 175, Mid-Atlantic Ridge. *Geochemistry, Geophysics, Geosystems*, 16(10), 3716–3732.
- 644 Schwartz, S., Guillot, S., Reynard, B., Lafay, R., Debret, B., Nicollet, C., Lanari, P., and Auzende, A.L.
645 (2013) Pressure–temperature estimates of the lizardite/antigorite transition in high pressure
646 serpentinites. *Lithos*, 178, 197–210.
- 647 Scott, S.R., Sims, K.W.W., Frost, B.R., Kelemen, P.B., Evans, K.A., and Swapp, S.M. (2017) On the
648 hydration of olivine in ultramafic rocks: Implications from Fe isotopes in serpentinites. *Geochimica*
649 *et Cosmochimica Acta*, 215, 105–121.
- 650 Seyfried, W.E., Pester, N.J., Tutolo, B.M., and Ding, K. (2015) The Lost City hydrothermal system:
651 Constraints imposed by vent fluid chemistry and reaction path models on seafloor heat and mass
652 transfer processes. *Geochimica et Cosmochimica Acta*, 163, 59–79.
- 653 Shen, T., Zhang, C., Chen, J., Hermann, J., Zhang, L., Padrón-Navarta, J.A., Chen, L., Xu, J., and Yang, J.
654 (2020) Changes in the cell parameters of antigorite close to its dehydration reaction at subduction
655 zone conditions. *American Mineralogist*, 105(4), 569–582.

- 656 Tamblyn, R., Zack, T., Schmitt, A.K., Hand, M., Kelsey, D., Morrissey, L., Pabst, S., and Savov, I.P. (2019)
657 Blueschist from the Mariana forearc records long-lived residence of material in the subduction
658 channel. *Earth and Planetary Science Letters*, 519, 171–181.
- 659 Ulrich, M., Muñoz, M., Boulvais, P., Cathelineau, M., Cluzel, D., Guillot, S., and Picard, C. (2020)
660 Serpentinization of New Caledonia peridotites: from depth to (sub-)surface. *Contributions to*
661 *Mineralogy and Petrology*, 175(9), 91.
- 662 Valetich, M.J., Le Losq, C., Arculus, R.J., Umino, S., and Mavrogenes, J. (2021) Compositions and
663 Classification of Fractionated Boninite Series Melts from the Izu–Bonin–Mariana Arc: A Machine
664 Learning Approach. *Journal of Petrology*, 62(2), egab013.
- 665 Vitale Brovarone, A., and Herwartz, D. (2013) Timing of HP metamorphism in the Schistes Lustrés of
666 Alpine Corsica: New Lu–Hf garnet and lawsonite ages. *Lithos*, 172–173, 175–191.
- 667 Viti, C., and Mellini, M. (1997) Contrasting chemical compositions in associated lizardite and chrysotile in
668 veins from Elba, Italy. *European Journal of Mineralogy*, 9(3), 585–596.
- 669 Wang, A., Freeman, J.J., and Jolliff, B.L. (2015) Understanding the Raman spectral features of
670 phyllosilicates. *Journal of Raman Spectroscopy*, 46(10), 829–845. <https://doi.org/10.1002/jrs.4680>
- 671 Wang, Y., Qiu, K., Müller, A., Hou, Z., Zhu, Z., and Yu, H. (2021) Machine Learning Prediction of Quartz
672 Forming-Environments. *Journal of Geophysical Research: Solid Earth*, 126(8).
- 673 Wei, D., Chen, S., and Liu, Q. (2015) Review of fluorescence suppression techniques In Raman
674 spectroscopy. *Applied Spectroscopy Reviews*, 50(5), 387–406.
- 675 Wheat, G.C., Seewald, J.S., and Takai, K. (2020) Fluid transport and reaction processes within a
676 serpentinite mud volcano: South Chamorro Seamount. *Geochimica et Cosmochimica Acta*, 269,
677 413–428.
- 678 Whittaker, E.J.W., and Wicks, F.J. (1970) Chemical difference among the serpentine "polymorphs": A
679 discussion. *American Mineralogist*, 55, 1025–1047.
- 680 Wicks, F.J., and Whittaker, E.J.W. (1975) A reappraisal of the structures of the serpentine minerals. *The*
681 *Canadian Mineralogist*, 13, 227–243.
- 682 Wu, K., Ding, X., Ling, M.-X., Sun, W., Zhang, L.-P., Hu, Y.-B., and Huang, R.-F. (2018) Origins of two
683 types of serpentinites from the Qinling orogenic belt, central China and associated fluid/melt-rock
684 interactions. *Lithos*, 302–303, 50–64.
- 685 Zhang, G., Zhang, L., Song, S., and Niu, Y. (2009) UHP metamorphic evolution and SHRIMP
686 geochronology of a coesite-bearing meta-ophiolitic gabbro in the North Qaidam, NW China.
687 *Journal of Asian Earth Sciences*, 35(3–4), 310–322.

- 688 Zhang, L., Sun, W., and Chen, R.-X. (2019) Evolution of serpentinite from seafloor hydration to subduction
689 zone metamorphism: Petrology and geochemistry of serpentinite from the ultrahigh pressure North
690 Qaidam orogen in northern Tibet. *Lithos*, 346–347, 105158.
- 691 Zhang, Y., Gazel, E., Gaetani, G.A., and Klein, F. (2021) Serpentinite-derived slab fluids control the
692 oxidation state of the subarc mantle. *Science Advances*, 7, eabj2515.
- 693 Zhou, S., Zhou, K., Wang, J., Yang, G., and Wang, S. (2018) Application of cluster analysis to geochemical
694 compositional data for identifying ore-related geochemical anomalies. *Frontiers of Earth Science*,
695 12(3), 491–505.
- 696
- 697

698 **Figure 1.** Operational flow chart of the methods used in this study. Step 1: the original dataset was used to
699 train the XGBoost models and evaluate model performance. Step 2: the preprocessed (KNN & CLR)
700 original dataset was used to train the *k*-means clustering models.

701

702 **Figure 2.** The distributions of the eight analyzed elemental concentrations in serpentine minerals. L_Srp
703 represents chrysotile and lizardite, and H_Srp represents antigorite.

704

705 **Figure 3.** The performance of our XGBoost model presented in the form of **(a)** nonnormalized and **(b)**
706 normalized confusion matrices. **(c)** Relative feature importances (summing to 1) during the training of the
707 XGBoost algorithm to classify serpentines as either L_Srp or H_Srp.

708

709 **Figure 4.** Distribution plots of **(a,d)** PC1 vs. PC2, **(b,e)** PC1 vs. PC3, and **(c,f)** PC2 vs. PC3 for **(a–c)** L_Srp
710 and **(d–f)** H_Srp. Symbol colors represent clusters identified by *k*-means analysis and symbol shapes
711 indicate sampling locations. MAR, Mid-Atlantic Ridge; IOR, Indian Oceanic Ridge; IBM, Izu-Bonin
712 Mariana arc; ED, Eastern Desert; BC, British Columbia; NQO, North Qaidam orogen.

713

714 **Figure 5.** The distributions of each cluster of major chemical compositions of **(a)** L_Srp and **(b)** H_Srp
715 identified by *k*-means analysis. BC, British Columbia; NQO, North Qaidam orogen; IBM, Izu-Bonin-
716 Mariana arc; ED, Eastern Desert; MAR, Mid-Atlantic Ridge; IOR, Indian Oceanic Ridge.

717

718

719 **Figure S1.** TEM images of **(a)** chrysotile, **(b)** lizardite, and **(c)** antigorite. Inset images are schematic
720 diagrams of their respective structures, with yellow tetrahedra, purple tetrahedra, light green octahedra, and
721 purple octahedra representing SiO_4 , AlO_4 , $\text{Mg}(\text{O},\text{OH})_6$, and $\text{Al}(\text{O},\text{OH})_6$, respectively.

722

723 **Figure S2.** The XGBoost learning curve when using the eight selected major elements to classify L_Srp
724 and H_Srp.

725

726 **Figure S3.** Silhouette scores used to select the optimum number of clusters k when applying the k -means
727 algorithm to **(a)** L_Srp and **(b)** H_Srp.

728

729 **Figure S4.** Violin plots of the density distributions of oxide concentrations in each L_Srp cluster.

730

731 **Figure S5.** Violin plots of the density distributions of oxide concentrations in each H_Srp cluster.

732

733

734 Table 1. Summary of the performance of the XGBoost model applied to the test serpentine dataset.

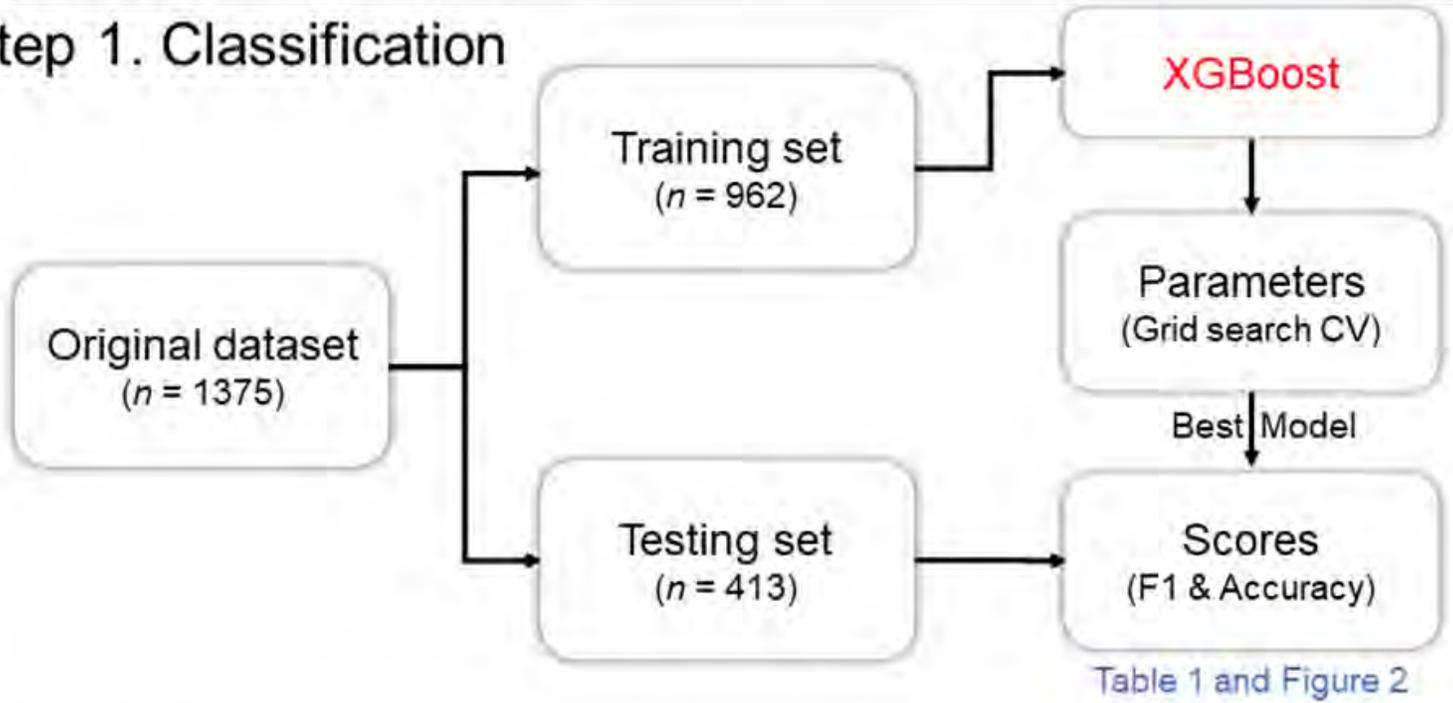
Class	Precision (%)	Recall (%)	F1-score	Accuracy (%)
L_Srp	81.7	86.3	0.839	87.2
H_Srp	91.0	87.7	0.872	

735

736

Figure 1

Step 1. Classification



Step 2. Clustering

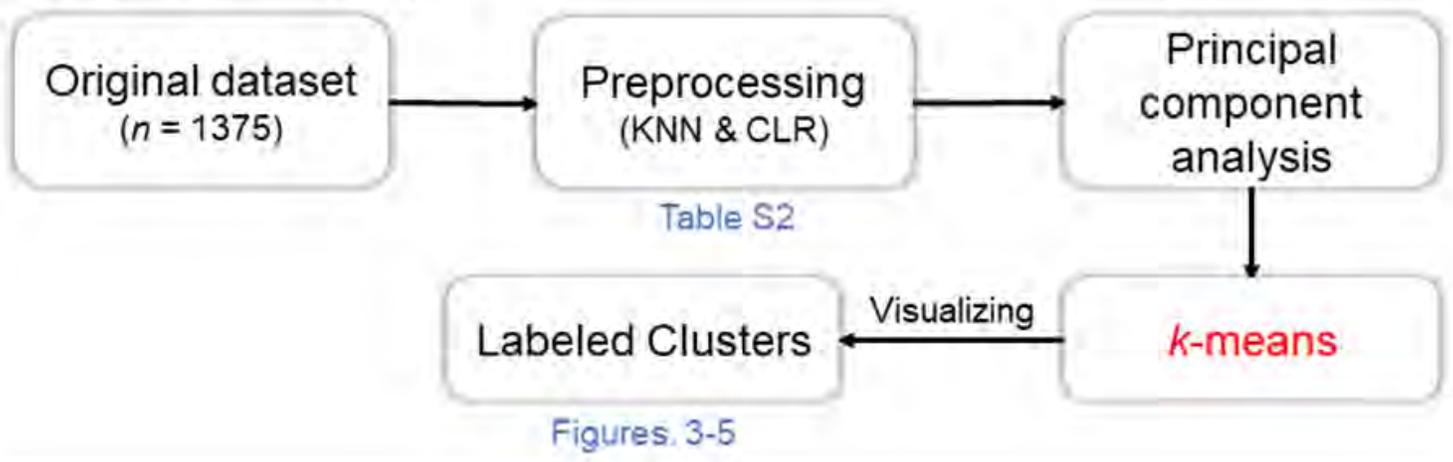


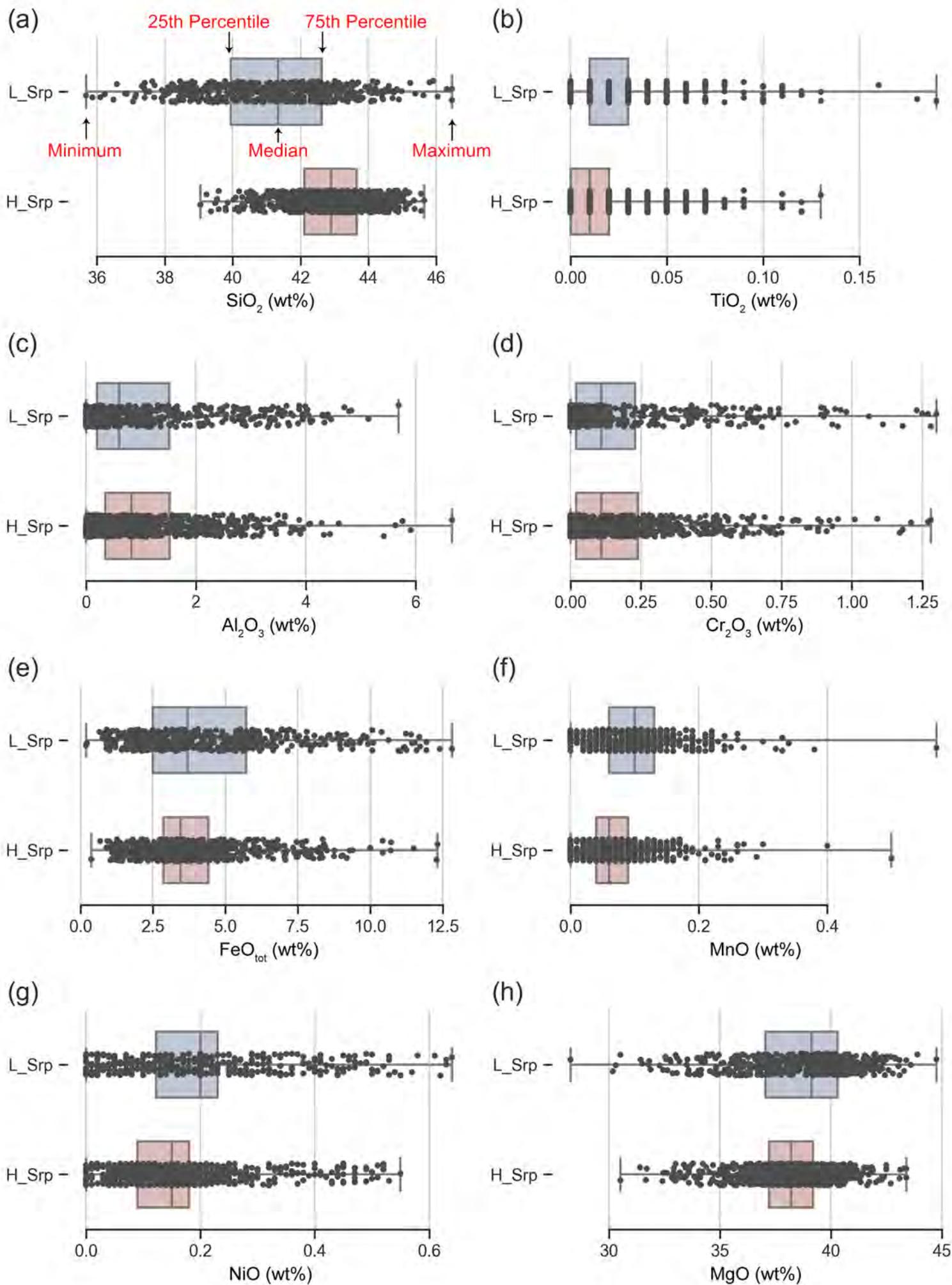
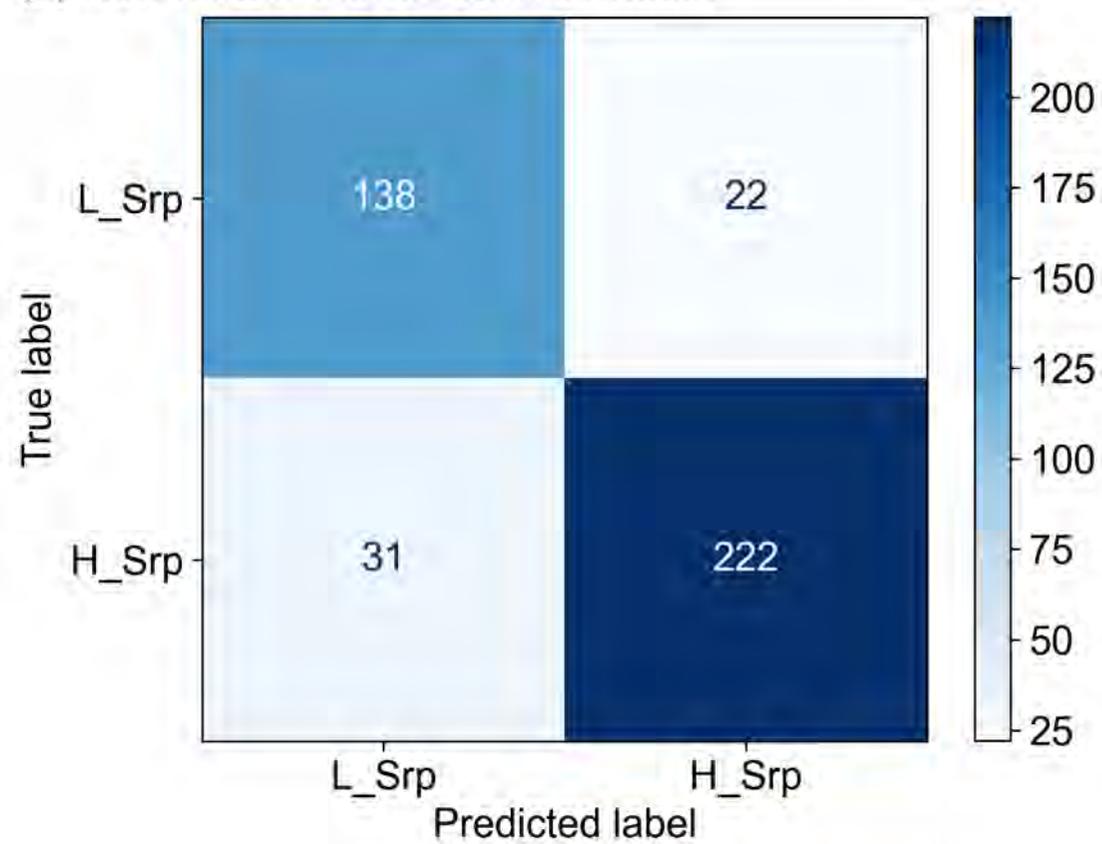
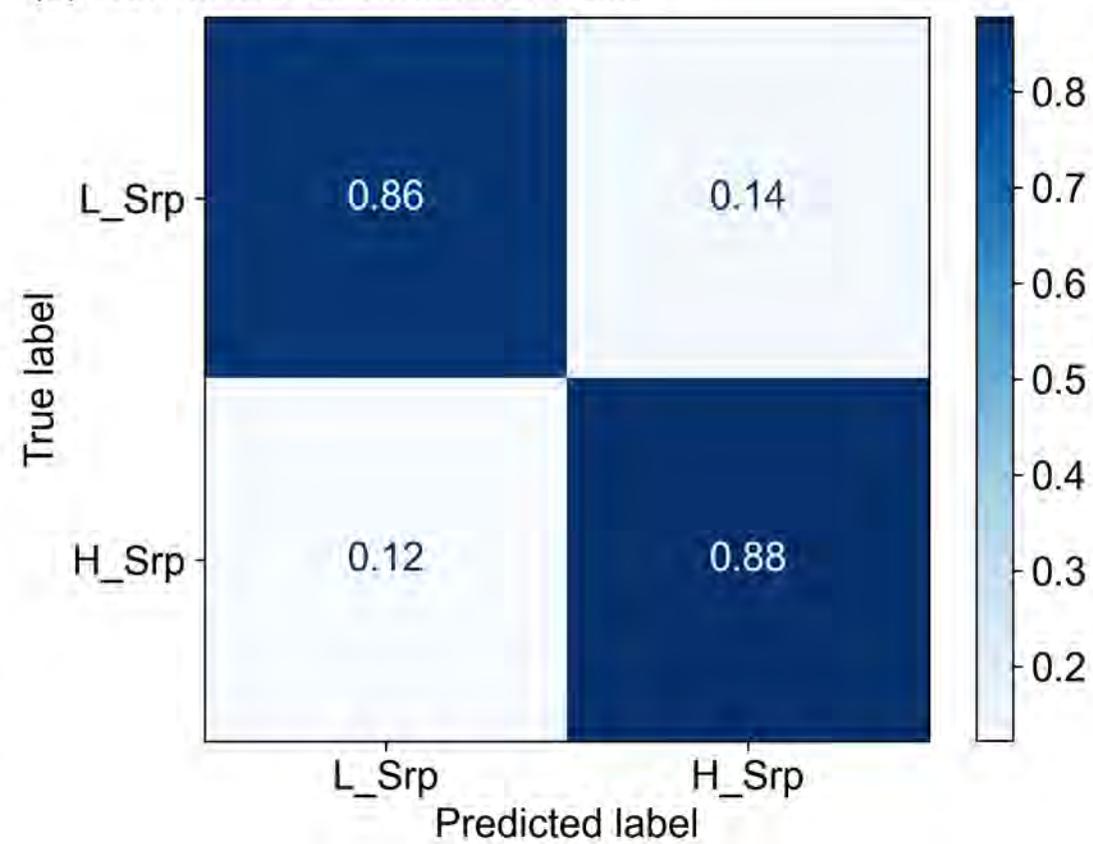
Figure 2

Figure 3

(a) Non-normalized Confusion Matrix



(b) Normalized Confusion Matrix



(c) Feature Importance

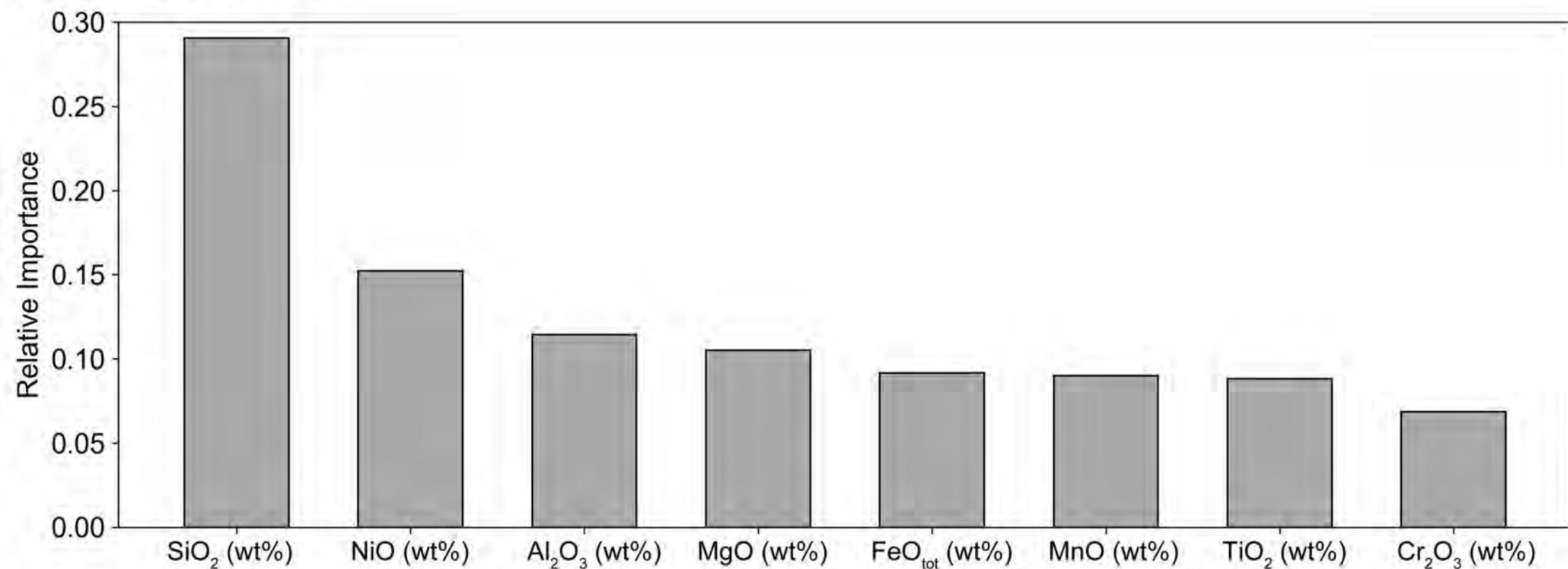


Figure 4

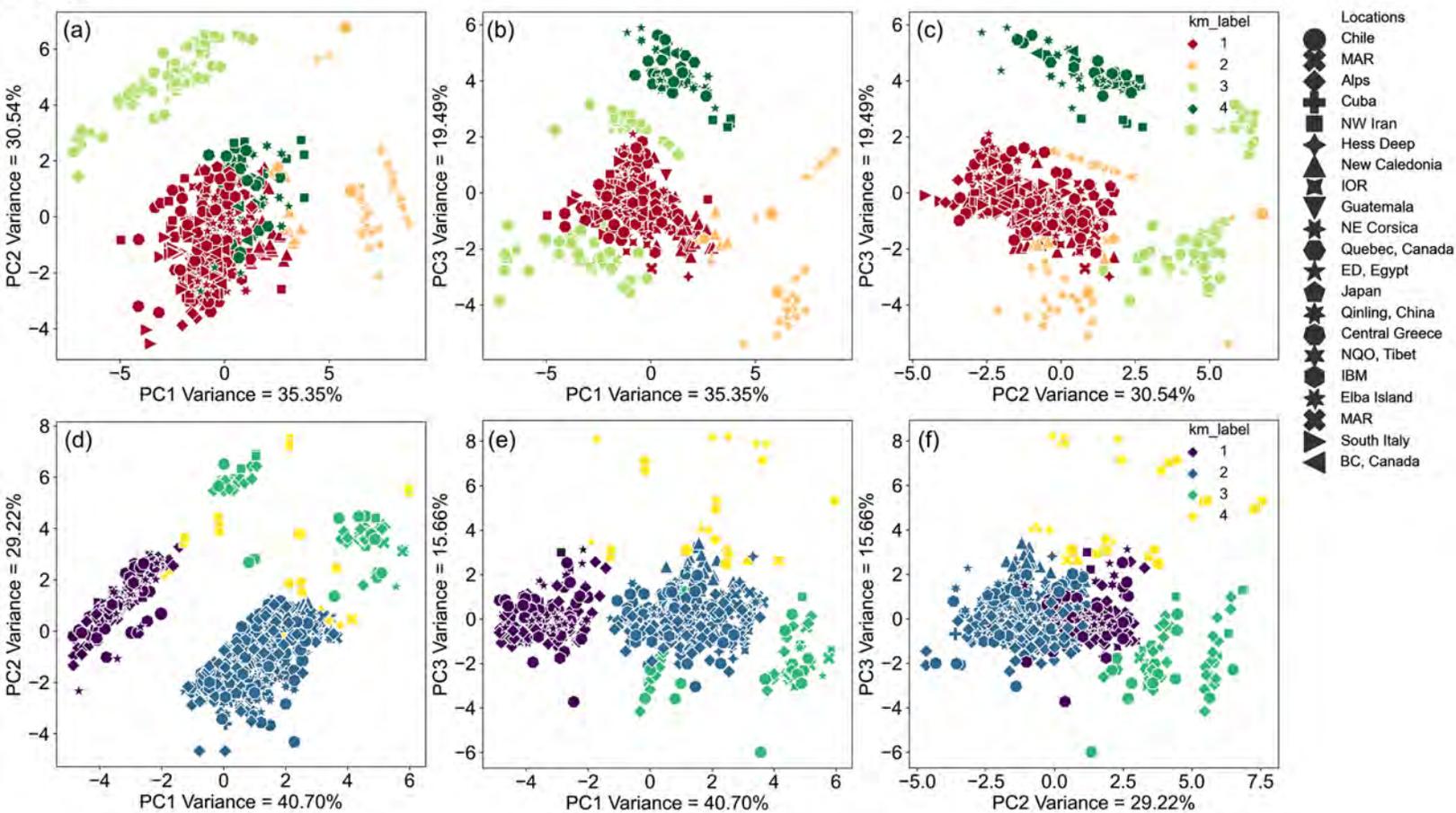
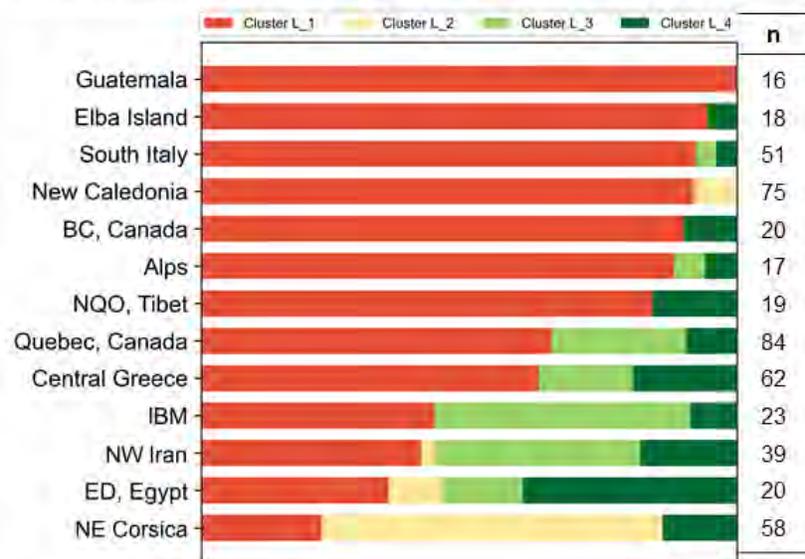


Figure 5

(a) Clusters classified by *k*-means algorithm for L_Srp



(b) Clusters classified by *k*-means algorithm for H_Srp

