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2 **Revision 1**

3 **Thermoelasticity of phase D and implications for low-velocity**
4 **anomalies and local discontinuities at the uppermost lower mantle**

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15 **Abstract**

16 The distribution of water reservoirs in the deep Earth is critical to understanding
17 geochemical evolution and mantle dynamics. Phase D is a potential water carrier in
18 the slab subducted to the uppermost lower mantle (ULM) and its seismic velocity and
19 density characteristics are important for seismological detection on water reservoirs,
20 but these properties remain poorly constrained. Here we calculate the seismic
21 velocities and density of Mg-endmember phase D ($\text{MgSi}_2\text{H}_2\text{O}_6$) under the ULM
22 conditions using first-principles calculations based on the density functional theory.
23 The velocities of phase D are higher than those of periclase and slightly lower than
24 those of bridgmanite by 0.5–3.4% for V_P and by 0–1.9% for V_S between 660- and
25 1000-km depths. Considering its relatively low content, phase D can hardly produce a
26 low-velocity anomaly in the ULM observed by seismological studies. However, due
27 to its strong elastic anisotropy, it may contribute significantly to the observed seismic
28 anisotropy at a similar depth. Additionally, phase D dehydrates into bridgmanite and
29 stishovite at the ULM, producing insignificant velocity changes but a substantial
30 density increase of ~14%. Therefore, the dehydration is probably too weak to generate
31 discontinuities associated with velocity jumps, whereas it may account for seismic
32 discontinuities that are sensitive to impedance changes, and particularly density
33 jumps, near the dehydration depth observed in some subduction zones.

34 **Keywords:** Phase D, dense hydrous magnesium silicate, high-pressure phase
35 transition, dehydration, impedance jump, seismic discontinuity

36

37 **1. Introduction**

38 Water in the Earth's interior exerts significant influences on geochemical
39 evolution and mantle dynamics because a small amount of water can significantly
40 change the rheological properties, melting temperature, diffusion rate of materials,
41 and phase stability (Hirschmann, 2006; Karato and Jung, 2010; Mei and Kohlstedt,
42 2000). For example, water can reduce both viscosity and solidus temperature of
43 mantle rocks, which significantly influences the pattern and velocity of mantle flow. It
44 has been shown that wadsleyite and ringwoodite, candidate nominally anhydrous
45 minerals in the mantle transition zone (MTZ), could incorporate several weight
46 percent of H₂O (Bell and Rossman, 1992; Inoue et al., 2010; Inoue et al., 1995;
47 Jacobsen et al., 2005; Smyth, 1987), whilst the hydrous ringwoodite and ice-VII
48 discovered in superdeep diamonds (Pearson et al., 2014; Tschauer et al., 2018)
49 provide direct evidence for the existence of the non-negligible amount of H₂O in the
50 deep mantle. Water can be likely transported into the MTZ and the lower mantle via
51 subducting slabs through the formation and dehydration of dense hydrous magnesium
52 silicates (DHMSs) (Angel et al., 2001). Some geophysical anomalies, such as high
53 V_p/V_s , high electrical conductivity, and low-velocity anomalies, were interpreted as
54 locally high water content released by the dehydration of DHMSs (Karato, 2011; Li et
55 al., 2013; Savage, 2012) or merely their existence owing to their low-velocity
56 characteristics (Liu et al., 2016; Schmandt et al., 2014; Yang et al., 2017). Meanwhile,
57 the generation/breakdown of DHMSs could also cause considerable impedance
58 contrasts, giving rise to seismic discontinuities. For example, the decomposition of a

59 small amount of superhydrous phase B may contribute to the 800-km discontinuity in
60 western Pacific subduction zones (Yang et al., 2017), and the dehydration of phase H
61 could produce significant seismic impedance increase in the mid-mantle (Song et al.,
62 2022). Combining these seismological observations with the velocity and density
63 characteristics, phase transitions, and corresponding impedance contrasts of DHMSs
64 could help constrain the water content in the deep interior and provide insights into
65 the circulation of water in the whole Earth.

66 Phase D is considered as a significant carrier of water in slabs subducted to the
67 uppermost lower mantle (ULM). The ideal chemical formula of phase D is
68 $\text{MgSi}_2\text{H}_2\text{O}_6$ containing 10 wt.% water, whereas the synthesized phase D shows a wide
69 variation in Mg/Si ratio from 0.53 to 0.71 and the water content varies from 10 wt.%
70 to 18 wt.% (Chang et al., 2013; Frost and Fei, 1999; Hushur et al., 2011; Litasov et
71 al., 2007; Rosa et al., 2013a; Rosa et al., 2012; Shinmei et al., 2008; Wu et al., 2016;
72 Xu et al., 2020; Xu et al., 2021b; Yang et al., 1997). Many hydrous phases exhibit low
73 velocities (Li et al., 2016; Mao et al., 2012; Rosa et al., 2012; Rosa et al., 2015; Yang
74 et al., 2017) and could be identified by seismic observations. The weight fraction of
75 phase D can be as high as 57% in hydrous peridotite (Ohtani et al., 2004), which is
76 also possible to generate seismic velocity anomalies in the MTZ and the ULM. On the
77 other hand, phase D will dehydrate into bridgmanite and stishovite at low
78 temperatures at the ULM (Nishi et al., 2014). Correspondingly, many seismic studies
79 detected discontinuities exhibiting large impedance contrasts at the ULM in
80 subduction zones (Courtier and Revenaugh, 2008; Schumacher and Thomas, 2016;

81 Waszek et al., 2018). The overlap of these depths may suggest a connection between
82 these discontinuities and the dehydration of phase D. However, the velocities and
83 density of phase D under the lower mantle conditions, which are crucial for
84 deciphering its role in such seismic observations, remain unknown.

85 The crystal structure and equation of state of phase D have been widely studied
86 (Frost and Fei, 1998; Frost and Fei, 1999; Hushur et al., 2011; Kudoh et al., 1997;
87 Litasov et al., 2007; Liu, 1987; Liu et al., 1998; Rosa et al., 2013a; Rosa et al., 2012;
88 Shieh et al., 2009; Shinmei et al., 2008; Wu et al., 2016; Xue et al., 2008; Yang et al.,
89 1997) and the elastic properties of Mg-endmember phase D were investigated by first-
90 principles calculations at static conditions (Mainprice et al., 2007; Thompson et al.,
91 2022; Tsuchiya and Tsuchiya, 2008) and Brillouin scattering and ultrasonic
92 measurements at ambient conditions (Liu et al., 2004; Rosa et al., 2012; Xu et al.,
93 2021b). The sound velocities of Al-bearing phase D up to 22 GPa and 1300 K were
94 also determined by ultrasonic measurements (Xu et al., 2020). However, there are no
95 elasticity data of Mg-endmember phase D under both high-temperature and high-
96 pressure conditions, which are crucial to understanding its characteristics and
97 constraining its distribution. In this contribution, we obtained the elastic properties of
98 Mg-endmember phase D ($\text{MgSi}_2\text{H}_2\text{O}_6$) and its velocity and density characteristics
99 under the lower mantle conditions using first-principles calculations within the
100 generalized gradient approximation. Combining our results with available data of
101 other minerals, we calculated the velocity and impedance contrasts caused by its
102 dehydration and discussed its close relationship to seismic observations in the ULM.

103 2. Computational Detail

104 The Mg-endmember phase D ($\text{MgSi}_2\text{H}_2\text{O}_6$) is trigonal and in the $P\bar{3}1m$ space
105 group, with MgO_6 and SiO_6 octahedra in two separate layers stacked along the c-axis
106 (Figure 1). The H–O bonds of phase D are located between adjacent octahedra in the
107 MgO_6 layer and the hydrogens are only 1/3 occupied (Xue et al., 2008; Yang et al.,
108 1997). To maintain the symmetries of phase D and handle the fractional occupancy of
109 H atoms, instead of using a triclinic unit cell as the one in Tsuchiya et al. (2005), we
110 constructed a super cell following Mainprice et al. (2007) (Figure 1). The basic vector
111 (**a**, **b**, **c**) in this super cell is equal to (**a-b**, **a+2b**, **c**) in the unit cell, so it consists of 3
112 unit cells (33 atoms) and has a slightly different space group, $P\bar{3}m1$.

113 All calculations in this study were performed based on the density functional
114 theory (DFT) using the open-source quantum espresso package (Giannozzi et al.,
115 2009) with the Generalized Gradient Approximation (GGA) (Hamann, 1997; Perdew
116 et al., 1996). The energy cutoff for the plane wave was set to 70 Ry. Structural
117 optimizations were performed using the variable cell-shape damped molecular
118 dynamics method (Wentzcovitch et al., 1993) at certain pressures with a k-point mesh
119 of $6 \times 6 \times 6$.

120 Isothermal elastic constants can be expressed as (Barron and Klein, 1965)

$$121 \quad c_{ijkl}^T = \frac{1}{V} \left(\frac{\partial^2 F}{\partial e_{ij} \partial e_{kl}} \right) + \frac{1}{2} P (2\delta_{ij}\delta_{kl} - \delta_{il}\delta_{kj} - \delta_{ik}\delta_{jl}), \quad (1)$$

122 where V , T , P , e_{ij} , δ_{ij} , and F represent the volume, temperature, pressure,
123 infinitesimal strain, Kronecker delta, and Helmholtz free energy, respectively.

124 Adiabatic elastic constants can be further calculated by

125
$$c_{ijkl}^S = c_{ijkl}^T + \frac{T}{VC_V} \frac{\partial S}{\partial e_{ij}} \frac{\partial S}{\partial e_{kl}} \delta_{ij} \delta_{kl} , \quad (2)$$

126 where S and C_V represent the entropy and isochoric heat capacity, respectively.

127 According to equation (1), Helmholtz free energy F is required to obtain these elastic
128 constants, which, in quasi-harmonic approximation, is expressed as

129
$$F(V, T, e_{ij}) = U_0(V, e_{ij}) + \sum_{q,m} \frac{\hbar \omega_{qm}(V, e_{ij})}{2} + k_B T \sum_{q,m} \ln \left\{ 1 - \exp \left[- \frac{\hbar \omega_{qm}(V, e_{ij})}{k_B T} \right] \right\}.$$

130 (3)

131 The three terms on the right side are static internal, zero point, and vibrational energy,
132 where k_B and \hbar are Boltzmann and reduced Planck constants, respectively, and ω_{qm} is
133 the vibrational frequency. Equation (1) and (3) suggest that the determination of all
134 elastic constants requires the computation of vibrational frequencies, ω_{qm} , for many
135 (~ 10) strained configurations, which is computationally demanding. To address this
136 challenge, we employ the semi-analytical method developed by Wu and Wentzcovitch
137 (2011) to calculate the thermal elasticity of phase D, which requires only the
138 computation of vibrational frequencies under the unstrained condition and static
139 elastic constants. The static elastic constants were calculated according to the stress-
140 strain relationship with $\pm 1\%$ strain. The dynamical matrices with a $4 \times 4 \times 4$ q-point
141 mesh were calculated using density functional perturbation theory (Baroni et al.,
142 2001) and further interpolated in a denser mesh to obtain the vibrational density of
143 state. This method accelerates the computational efficiency by approximately tenfold
144 while maintaining high accuracy, and it has been successfully applied to numerous
145 minerals (Hao et al., 2019; Qian et al., 2018; Wang et al., 2020; Wang et al., 2019; Wu
146 and Wang, 2016; Yang et al., 2017; Yao et al., 2018; Zou et al., 2018). At each

147 pressure, we calculate the volume and corresponding vibrational frequencies under
148 the unstrained condition, which are utilized to compute the Helmholtz free energy at
149 different temperatures and volumes based on Equation (3). Finally, we apply the semi-
150 analytical method developed by Wu and Wentzcovitch (2011) to calculate the
151 adiabatic elastic constants.

152

153 **3. Results**

154 **3.1 Absent H-bond symmetrization in optimized structures**

155 Previous theoretical studies (Thompson et al., 2022; Tsuchiya and Tsuchiya,
156 2008; Tsuchiya et al., 2005) predicted that the hydrogen bond symmetrization (HBS)
157 (the H atom is located at the middle point of two neighboring O atoms) takes place in
158 phase D at the pressure of approximately 40 GPa, causing a ~20% increase in the bulk
159 modulus. However, HBS is absent in the optimized structures up to 80 GPa in both
160 Mainprice et al. (2007) and our study (Figure 1c) and neither of the studies shows an
161 abrupt increase in the bulk modulus (see Section 3.3). The discrepancy is probably
162 caused by different structural models of phase D related to the fractional occupancy of
163 hydrogen, as discussed in Tsuchiya and Tsuchiya (2008). The structures of Mg-
164 endmember phase D in their studies are unit cells, where H atoms occupy 2 of 6 sites,
165 thus the structure distorts from trigonal and has a triclinic unit cell, whereas we use a
166 triple cell to keep the symmetry.

167 Despite this discrepancy, infrared spectroscopic studies on phase D
168 ($\text{Mg}_{2.23}\text{Si}_{1.18}\text{H}_{2.80}\text{O}_6$) observe neither significant changes in the frequency or intensity

169 of OH stretching vibrations nor the merging of the separate OH peaks with increasing
170 pressure up to 42 GPa (Shieh et al., 2009). In a recent study, Meier et al. (2022)
171 suggest that the minima in the pressure dependence of the NMR resonance line-
172 widths of Al- and Fe-bearing phase D ($(\text{Mg}_{0.88}, \text{Fe}_{0.12})(\text{Si}_{0.9}, \text{Al}_{0.1})\text{O}_6\text{H}_2$), which
173 represents the maximum in hydrogen mobility, can be regarded as the precursor to
174 HBS. However, based on first principles calculations, Thompson et al. (2022) suggest
175 that phase D with 50% Al-substitution ($\text{AlMg}_{0.5}\text{Si}_{1.5}\text{O}_6\text{H}_2$), containing 8 unit cells in
176 their setting, does not show HBS in the pressure range of 0–75 GPa, whereas both
177 Mg- and Al-endmember phase D, containing one and two unit cells respectively,
178 undergo pressure-induced HBS. The discrepancy in experiments and calculations
179 suggests that whether HBS is present could be affected by both the structural models
180 and compositions of phase D. Our results provide the optimized structures of Mg-
181 endmember phase D based on the super-cell configuration and corresponding elastic
182 properties at high temperatures, which can be combined with prior results to explore
183 the effect of cell setting and Al content on the elasticity of phase D.

184

185 **3.2 Thermodynamic Properties**

186 The calculated equations of state of phase D are shown together with the
187 experimental results in Figure 2. The differences among experiments primarily result
188 from the wide variations in Mg/Si ratio, aluminum and water contents in the
189 synthesized phase D. Except the Al-bearing phase D (Xu et al., 2020), most measured
190 volumes (Frost and Fei, 1999; Hushur et al., 2011; Ohtani et al., 2004; Rosa et al.,

191 2013a; Rosa et al., 2012; Rosa et al., 2013b; Shinmei et al., 2008; Wu et al., 2016; Xu
192 et al., 2021b; Yang et al., 2017) are slightly smaller than our calculated results due to
193 the overestimation of GGA calculation, but there is much better consistency in the
194 relative change of volume (V/V_0) with pressure (Figure 2c). Our calculated volumes
195 lie between the experimental data of Shinmei et al. (2008) and Wu et al. (2016) within
196 a broad pressure range and agree with those of Rosa et al. (2013a) at all pressures
197 (Figure 2c). Furthermore, the calculated relative changes in cell parameters are also
198 consistent with the experimental results. In comparison with the absolute values at
199 different pressures, the relative changes in volume (V/V_0) and lattice constants (a/a_0 ,
200 c/c_0) play a more important role in the determination of the compressional properties.

201 The calculated thermal expansion ($\alpha = \frac{1}{V} \frac{\partial V}{\partial T}$) of phase D as well as the
202 experimental data (Shinmei et al., 2008) are shown in Figure 3a. The blue dashed line
203 represents the experimental results at 0 GPa (Shinmei et al., 2008), which are larger
204 than our calculated results. The thermal expansion of Shinmei et al. (2008) was
205 calculated using high-temperature Birch-Murnaghan equation of state (HTBM EOS)
206 based on the P-V-T data of phase D. It should be noted that there are only two room-
207 temperature volume data points at ambient pressure in their study, whereas most of
208 the data were measured at 17-46 GPa. Therefore, the thermal expansion of phase D at
209 high pressures should be more convincing than that at 0 GPa, and these high-pressure
210 data do agree better with our calculated results (Figure 3a). The calculated thermal
211 Grüneisen parameter, $\gamma_{th} = V \left(\frac{\partial P}{\partial U} \right)_V$ (U is the internal energy), decreases with
212 pressure (Figure 3b). The heat capacity at constant volume ($C_V = \left(\frac{\partial U}{\partial T} \right)_V$) and constant

213 pressure ($C_P = \left(\frac{\partial H}{\partial T}\right)_P$) (H is the enthalpy) increase significantly with temperature but
214 slightly decreases with pressure (Figure 3c-d).

215 **3.3 Thermal Elastic Properties**

216 The elastic tensor of phase D with a trigonal symmetry can be totally determined
217 by six independent elastic constants (C_{11} , C_{33} , C_{12} , C_{13} , C_{44} , and C_{14}). The calculated
218 thermal elastic constants, adiabatic bulk (K_S) and shear (G) moduli, longitudinal (V_P)
219 and shear wave (V_S) velocities as well as the experimental data (Rosa et al., 2012; Xu
220 et al., 2020; Xu et al., 2021b) are shown in Figure 4. The first and second derivatives
221 of elastic moduli M ($M=C_{ij}$, K_S , and G) with respect to pressure and temperature are
222 shown in Table S1 and Table S2, respectively. The calculated compressional elastic
223 constants (C_{11} and C_{33}) and shear elastic constant (C_{44}) are slightly larger than the
224 experimental results (Rosa et al., 2012), whereas the off-diagonal elastic constants
225 (C_{12} , C_{13} , and C_{14}) are slightly smaller (Figure 4). Since there are no high-pressure
226 experimental data of elastic constants of phase D, we compared the linear
227 compressibility calculated from our elastic constants with that from the lattice
228 constants of experimental data and they exhibit good consistency (Text S1; Figure
229 S1).

230 The adiabatic bulk moduli (K_S) are in good agreement with the experimental
231 results of Rosa et al. (2012) and slightly larger than that of Xu et al. (2020) and Xu et
232 al. (2021b), but our shear moduli (G) are significantly larger than all experiments,
233 particularly Xu et al. (2021b) (Figure 4c). The deviations probably result from
234 different Al and H₂O contents in phase D. Phase D in this study contains 10 wt.%

235 H₂O, whereas the synthesized samples in Rosa et al. (2012) and Xu et al. (2021b)
236 include 12.1 and 16.1 wt.% H₂O, respectively, and the sample in Xu et al. (2020)
237 contains 18.8 wt.% Al₂O₃ and ~16.0 wt.% H₂O. The negative correlation between the
238 H₂O content and elastic moduli is consistent with many minerals, such as wadsleyite
239 and ringwoodite (Wang et al., 2020; Wang et al., 2019), but the effect of aluminum
240 content on the elastic moduli of phase D requires further investigation. Since the
241 density of phase D in this study is close to these experiments and the compressional
242 and shear wave velocities are expressed by $V_P = \sqrt{(K_S + \frac{4}{3}G) / \rho}$ and $V_S = \sqrt{G / \rho}$, the
243 larger shear moduli further lead to higher velocities (Figure 4d). The V_P and V_S of
244 phase D in this study are 3.0% and 5.8% larger than those in Rosa et al. (2012) at
245 ambient conditions and approximately 4–6% and 8–9% larger than that in Xu et al.
246 (2021b) at pressures of 3–13.6 GPa, respectively.

247 Our results do not show the abrupt increase in the bulk modulus of phase D up to
248 80 GPa, which disagrees with the result of Tsuchiya et al. (2005). They attribute the
249 jump of bulk modulus to HBS at approximately 40 GPa, whereas the absent HBS in
250 this study, which are probably caused by different cell settings as discussed in Section
251 3.1, does not cause such a jump. Among experimental results, the dramatic increase in
252 bulk modulus observed by Hushur et al. (2011) probably results from the assumption
253 of a fixed K'_0 (the pressure derivative of the bulk modulus at 0 GPa) of 4, whereas
254 Rosa et al. (2012), Xu et al. (2020) and this study obtain a value close to or larger than
255 5 (Table S1). The large K'_0 can fit the equation of states well without an abrupt
256 increase in bulk modulus.

257 3.4 Anisotropy

258 The elastic wave velocities of single crystal usually exhibit variations along
259 different crystallographic orientations and the single-crystal anisotropy can be defined
260 as in Karki et al. (2001):

$$\begin{aligned} 261 \quad A_P &= 2 \times \frac{(V_{P,\max} - V_{P,\min})}{(V_{P,\max} + V_{P,\min})} \\ 262 \quad A_S &= 2 \times \frac{(V_{S,\max} - V_{S,\min})}{(V_{S,\max} + V_{S,\min})} \\ 263 \quad A_S^{Po} &= 2 \times \frac{(V_{S1} - V_{S2})_{\max}}{V_{S1} + V_{S2}}, \end{aligned} \quad (4)$$

264 where A_P , A_S and A_S^{Po} represent the V_P , V_S , and V_S polarization anisotropies,
265 respectively. V_P , V_{S1} , and V_{S2} represent the wave velocities along a given
266 crystallographic orientation, which can be calculated using the Christoffel equation
267 (Musgrave, 1970):

$$268 \quad |C_{ijkl}n_jn_l - \rho V^2\delta_{ik}| = 0. \quad (5)$$

269 Here C_{ijkl} refers to the fourth-ranked elastic tensor and the unit vector \mathbf{n} ((n_1, n_2, n_3))
270 is the propagation direction of the elastic wave. V and ρ represent velocity and
271 density, respectively.

272 The S wave polarization anisotropy of phase D is significantly large
273 (approximately 18%) under the conditions of the MTZ and ULM (Figure 5). Rosa et
274 al. (2013b) indicate that phase D exhibits a relatively low strength under uniaxial
275 compression and tends to develop lattice preferred orientations under plastic flow.
276 They also estimated that 16 vol.% of phase D in hydrous subducted peridotite could
277 explain the shear wave splitting ($0.9 \pm 0.3\%$) and the shear wave ray polarization
278 geometry observed in a detached fragment of the Tonga slab below the transition zone

279 (Chen and Brudzinski, 2003). Our calculated anisotropies of phase D are as strong as
280 those in Rosa et al. (2012) (Figure 5), further corroborating this interpretation.

281

282 **4. Discussions**

283 **4.1 The Velocities and Density Characteristics of Phase D in the MTZ and ULM**

284 Phase D is stable at the lowermost MTZ and the ULM at low temperatures (Nishi
285 et al., 2014). The seismic velocities and density of phase D, ringwoodite, periclase,
286 bridgmanite, and stishovite along a cold geotherm 500 K lower than the normal
287 geotherm (Brown and Shankland, 1981) are shown in Figure 6. Similar to other
288 hydrous phases, phase D has a significantly lower density than all other minerals,
289 especially at the ULM. The density contrast between phase D and bridgmanite is as
290 large as ~13%, thus it could contribute to the stagnation of slabs at the depth of ~600–
291 1000 km (Fukao et al., 2009) to a large extent. The seismic velocities of phase D,
292 however, surpass or are comparable to those of candidate minerals. Phase D shows
293 relatively high velocities in the MTZ, and the velocities of phase D increase faster
294 with pressure than those of candidate minerals in the MTZ and ULM (Figure 6) due to
295 its large pressure derivatives of K_S and G (Table S1). Its V_P and V_S are 1.5–3.0% and
296 6.5–8.7% higher than those of ringwoodite at depths of 500–660 km, respectively.
297 Although phase D exhibits significantly lower velocities than bridgmanite at ambient
298 conditions as in Rosa et al. (2012), the velocity contrasts between phase D and
299 bridgmanite are not prominent for both V_P (0.5%–3.4%) and V_S (0%–1.9%) within the
300 depth range of 660–1000 km (Figure 6). Moreover, the velocities of phase D are

301 larger than periclase in the same depth range (Figure 6). Stishovite, one of the
302 dehydration products of phase D (Nishi et al., 2014) and an important component in
303 the oceanic crust, has significantly high velocities in the MTZ (Karki et al., 2001;
304 Yang and Wu, 2014; Zhang et al., 2021). The velocity contrasts between phase D and
305 stishovite are 11.1–13.1% for V_P and 10.2–12.7% for V_S at depths of 500–660 km.
306 However, due to the softening of shear modulus of stishovite (Karki et al., 2001; Yang
307 and Wu, 2014; Zhang et al., 2021) and the large pressure dependence of velocities of
308 phase D, the V_P and V_S of stishovite are only 5.6% and 1.6% higher than those of
309 phase D at 1000-km depth, respectively, which become even smaller at larger depths
310 (Figure 6).

311 The comparable velocities of phase D to candidate minerals in the ULM suggest
312 that the accumulation of phase D can hardly produce prominent low-velocity
313 anomalies in the ULM observed by some seismological studies (Brudzinski and Chen,
314 2003; Liu et al., 2016), which is inconsistent with the conclusion drawn at ambient
315 conditions (Rosa et al., 2012). H_2O and Al have a significant effect on the velocities
316 and density of phase D. The velocities of phase D are negatively correlated with the
317 H_2O content (Figure 4), so a higher H_2O content may increase the possibility of
318 generating low-velocity anomalies, but it still requires more quantitative
319 investigations on the effect. Xu et al. (2020) used the data of Al-bearing phase D,
320 which include 18.8 wt.% Al_2O_3 and ~16.0 wt.% H_2O , to calculate the velocities and
321 density contrasts between the dry and hydrous harzburgite. Their calculation indicates
322 that the hydrous harzburgite with ~1.2 wt.% H_2O exhibits slightly lower velocities at

323 the ULM, -0.5% and -1.0% for V_P and V_S respectively, hardly accounting for the -3%
324 velocity anomalies for both V_P and V_S in Tonga slab (Brudzinski and Chen, 2003),
325 although their Al-bearing phase D has lower velocities than the Mg-endmember phase
326 D in our study under such conditions. This calculation provides an approximate
327 estimation on the H₂O effect, that is, such a water content is not enough to cause
328 obvious low-velocity anomalies, but the contribution of Al to the velocities remains to
329 be explored.

330 In contrast, superhydrous phase B, another stable hydrous mineral in cold slabs in
331 the ULM, may explain these low-velocity observations. It has much lower velocities
332 than bridgmanite and periclase, and the released water by its dehydration at the depth
333 of ~800 km should migrate upwards, causing the partial melt to reduce the velocity at
334 a shallower depth (Yang et al., 2017). However, superhydrous phase B has negligible
335 anisotropy compared with the strong anisotropy of phase D (Figure 5) under such
336 conditions, which cannot explain the observed seismic anisotropy in the same region
337 (Chen and Brudzinski, 2003). Previous studies suggest that phase D and superhydrous
338 phase B are likely to coexist in the ULM at low temperatures (Nishi et al., 2014; Xu et
339 al., 2021a). Therefore, the low-velocity anomaly could be mainly caused by
340 superhydrous phase B, whereas phase D may primarily contribute to the seismic
341 anisotropy.

342 **4.2 The Dehydration of Phase D and Implications on Discontinuities in the ULM**

343 At the ULM, with increasing pressure and temperature, phase D should dehydrate
344 into bridgmanite and stishovite:

345 $\text{MgSi}_2\text{O}_6(\text{phase D}) = \text{MgSiO}_3(\text{bridgmanite}) + \text{SiO}_2(\text{stishovite}) + \text{H}_2\text{O}$,

346 but the transition depth spans a wide range due to its significantly large negative

347 Clapeyron slope (dP/dT) (Nishi et al., 2014). Assuming that the slab is around 400–

348 500 K lower than the normal geotherm (Brown and Shankland, 1981), the dehydration

349 could take place across a broad depth range of approximately 700–1000 km, with

350 potential deviations influenced by uncertainties in the phase boundary (Nishi et al.,

351 2014). The V_P , V_S , and density contrasts between phase D and the aggregate of

352 bridgmanite plus stishovite along a cold isotherm 500 K lower than the normal

353 geotherm are shown in Figure 7. It is expected that the velocity jumps caused by the

354 dehydration of phase D decrease with increasing pressure because the velocities of

355 phase D increase faster with pressure than those of bridgmanite and stishovite and the

356 shear modulus of stishovite soften at high pressures (Karki et al., 2001; Yang and Wu,

357 2014; Zhang et al., 2021) (Figure 6). The velocity jumps caused by the dehydration of

358 phase D are 5.8% for V_P and 4.4% for V_S at 700-km depth but reduce to 2.3% and

359 0.4% at 1000-km depth, respectively (Figure 7). At deeper depths, the dehydration of

360 phase D even results in a decrease in V_S . In contrast, the density jump caused by the

361 dehydration of phase D is as large as ~14% at the depth of ~700–1000 km. The

362 impedance contrasts ($\Delta(\rho V)$, where ρ and V represent the density and wave velocity,

363 respectively) caused by the dehydration of phase D are 20% and 16% for

364 compressional wave and 19% and 14% for shear wave at 700- and 1000-km depths,

365 respectively, which are comparable to the transformation from olivine to wadsleyite

366 (Núñez Valdez et al., 2013) accounting for the 410-km discontinuity. Such large

367 impedance contrasts indicate that a small amount of phase D could produce
368 seismically detectable discontinuities at the ULM in subduction zones.

369 It should be noted that the effects of iron on the elasticity of phase D are not
370 considered above. Previous studies indicate that the iron in Fe-Al-bearing phase D
371 undergoes a high-spin to low-spin transition, which significantly reduce the bulk
372 modulus of phase D (Chang et al., 2013; Wu et al., 2016). The pressure range of the
373 spin transition in Fe-Al-bearing phase D is related to the valence state of iron. The
374 spin transition of Fe^{2+} occurs at 37-41 GPa and the spin transition of Fe^{3+} occurs at
375 40-65 GPa for $\sum \text{Fe}^{3+}/\text{Fe} = 0.94$ and at 64-68 GPa for $\sum \text{Fe}^{3+}/\text{Fe} = 0.40$ (Chang et
376 al., 2013; Wu et al., 2016). Therefore, the spin transition of Fe^{2+} in phase D and the
377 dehydration of phase D are likely to occur simultaneously near the 1000-km depth if
378 phase D contains a certain amount of Fe^{2+} . The spin transition of Fe^{2+} in phase D
379 ($\text{Mg}_{0.89}\text{Fe}_{0.11}\text{Al}_{0.37}\text{Si}_{1.55}\text{H}_{2.65}\text{O}_6$ $\sum \text{Fe}^{2+}/\text{Fe} = 0.60$) will cause a reduction of 28% on
380 bulk sound velocity and a reduction of 1.7% on volume (Wu et al., 2016). Iron in
381 bridgmanite occupies mainly in the Mg site as Fe^{2+} or Fe^{3+} , and the iron does not
382 experience any spin transition over the entire pressure range of the lower mantle. Si
383 site in bridgmanite may contain a small amount of Fe^{3+} , and the Fe^{3+} undergoes a spin
384 transition at approximately 15–50 GPa (Lin et al., 2013), whose effect on bulk
385 modulus of bridgmanite is relatively small at relevant mantle conditions (Badro, 2014;
386 Catalli et al., 2011; Catalli et al., 2010; Shukla and Wentzcovitch, 2016). Therefore,
387 the spin transition of Fe^{2+} may significantly increase the V_P jump and slightly
388 decrease the density jump caused by the dehydration of phase D at the depth of ~1000

389 km. Thus, the compressional impedance contrast caused by the dehydration of Fe-
390 bearing phase D may significantly increase if the Fe²⁺ in phase D undergoes a spin
391 transition.

392 Besides the global 410-km and 660-km discontinuities, seismological studies
393 detected many local discontinuities in the ULM, especially in subduction zones, and
394 their origins have been widely discussed (Courtier and Revenaugh, 2008; Schumacher
395 and Thomas, 2016; Waszek et al., 2018). These detections are sensitive to the
396 impedance contrasts across discontinuities and the most robust ones are at the depth
397 of ~800 km and ~1000 km, respectively. The dehydration of superhydrous phase B
398 may account for the discontinuities at the depth of ~800 km in subduction zones (Liu
399 et al., 2016; Porritt and Yoshioka, 2016; Yang et al., 2017), but the ones at the depth of
400 ~1000 km were ascribed to various mechanisms including viscosity jump (Marquardt
401 and Miyagi, 2015; Rudolph et al., 2015), mineral phase transitions within subducted
402 slab (King et al., 2015; Kingma et al., 1995), and the impedance contrasts between
403 oceanic crust and other parts of a slab (Niu, 2014; Rost et al., 2008). The dehydration
404 of phase D causes substantial impedance contrasts at the ULM, providing another
405 mechanism for these discontinuities. Most of these discontinuities are roughly located
406 within the fast anomalies in tomography models, that is, subducted slabs, where the
407 dehydration of phase D takes place. The dehydration of phase D at the ULM mainly
408 accounts for discontinuities caused by large impedance contrasts, specifically, density
409 contrast. Therefore, seismic observations which are mainly sensitive to the velocity
410 contrast, such as the S-to-P scatterers beneath the circum-Pacific regions near 1000-

411 km depth, may not result from the dehydration of phase D, where the presence of
412 oceanic crust is more preferred (Kaneshima, 2019).

413

414 **5. Implications**

415 In this study, we obtain the elasticity of Mg-endmember phase D at high
416 pressures and high temperatures using first-principles calculations based on the
417 density functional theory with the generalized gradient approximation. Compared
418 with other candidate minerals, the low-density feature of phase D could contribute to
419 the stagnation of slabs at the ULM. On the other hand, phase D has larger pressure
420 derivatives of K and G than those of major minerals in the MTZ and ULM, thus its
421 velocities increase much faster with depth. As a result, unlike other hydrous phases,
422 phase D exhibits higher velocities than ringwoodite at the transition zone and
423 comparable velocities to bridgmanite, periclase, and stishovite in the ULM,
424 respectively. Therefore, the accumulation of phase D is not likely to cause prominent
425 low-velocity anomalies at the ULM. In contrast, superhydrous phase B, another
426 hydrous phase coexisting with phase D at the ULM, may account for the low-velocity
427 anomalies, whereas phase D could explain the shear wave splitting in the same region.

428 The velocity contrasts caused by the dehydration of phase D into stishovite and
429 bridgmanite are negligible at the ULM, but the impedance contrasts are significantly
430 large because of the large density jump (~14%). Such large impedance contrasts may
431 provide an alternative explanation for the discontinuities at the ULM in subduction
432 regions. The equation of states and elasticity of Mg-endmember phase D obtained in

433 this study can be combined with other studies with different Al and water contents to
434 explore the effect of composition on these physical properties.

435

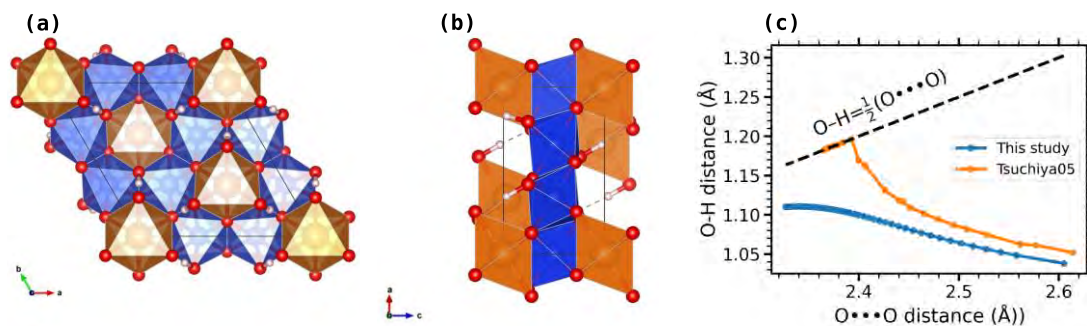
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440 the University of Science and Technology of China. The original data for all the
441 figures are available at the Zenodo database
442 (<https://doi.org/10.5281/zenodo.10452880>).

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446

447 **Figure 1.** (a) (b) Crystal structure of phase D at 0 GPa. The yellow, blue, red, and

448 white balls are Si, Mg, O, and H atoms, respectively. Images were generated in

449 VESTA (Momma and Izumi, 2008). (c) The relationship between the O...O and the

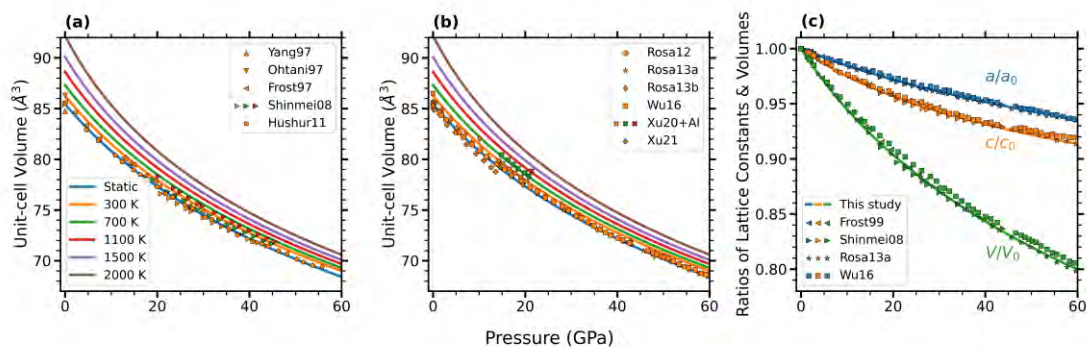
450 O-H distances of phase D in our study, shown in blue, and Tsuchiya et al. (2005),

451 shown in orange. The dashed line represents the relationship of hydrogen bond

452 symmetrization.

453

454



455

456 **Figure 2.** (a) The equation of states of phase D, (b) the relative change in volume

457 (V/V_0), and (c) lattice constant (a/a_0 , c/c_0) as a function of pressure (this study: static

458 results; experiments: data at 300 K). Solid lines represent our calculation results and

459 the experimental results are shown with scatters. Chemical formulas: $\text{MgSi}_2\text{H}_2\text{O}_6$

460 (this study), $\text{Mg}_{1.11}\text{Si}_{1.89}\text{H}_{2.22}\text{O}_6$ (Yang et al., 1997), $\text{Mg}_{1.14}\text{Si}_{1.73}\text{H}_{2.81}\text{O}_6$ (Ohtani et al.,

461 1997), $\text{Mg}_{1.11}\text{Si}_{1.6}\text{H}_{3.6}\text{O}_6$ (Frost and Fei, 1999), $\text{Mg}_{1.02}\text{Si}_{1.73}\text{H}_{3.03}\text{O}_6$ (Shinmei et al.,

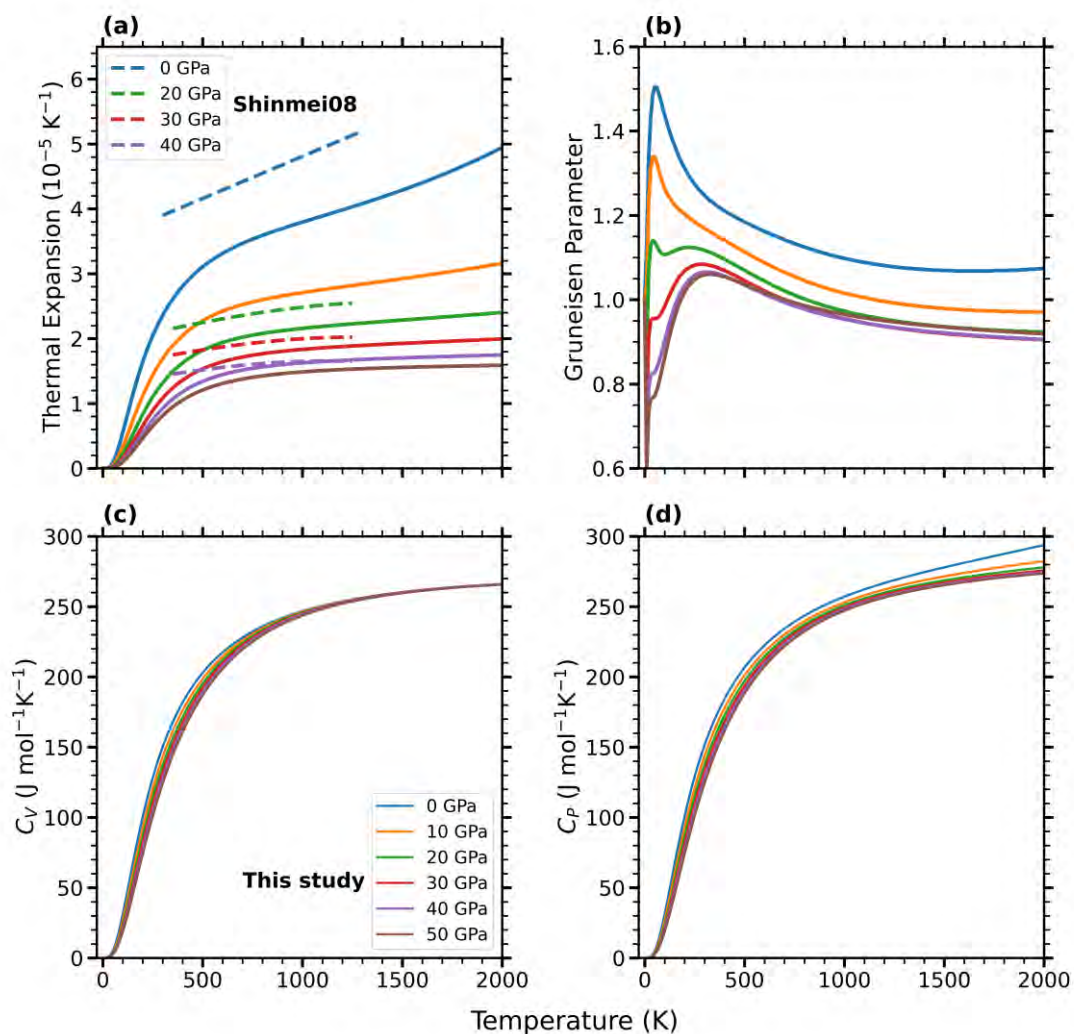
462 2008), $\text{Mg}_{1.0}\text{Si}_{1.7}\text{H}_{3.0}\text{O}_6$ (Hushur et al., 2011), $\text{Mg}_{1.1}\text{Si}_{1.9}\text{H}_{2.4}\text{O}_6$ (Rosa et al., 2012),

463 $\text{Mg}_{1.1}\text{Si}_{1.8}\text{H}_{2.5}\text{O}_6$ (Rosa et al., 2013a; Rosa et al., 2013b), $\text{Mg}_{1.14}\text{Si}_{1.73}\text{H}_{2.81}\text{O}_6$ (Wu et

464 al., 2016), $\text{Mg}_{0.89}\text{Si}_{1.30}\text{Al}_{0.64}\text{H}_{3.10}\text{O}_6$ (Al-bearing phase D) (Xu et al., 2020),

465 $\text{Mg}_{1.03}\text{Si}_{1.71}\text{H}_{3.05}\text{O}_6$ (Xu et al., 2021b).

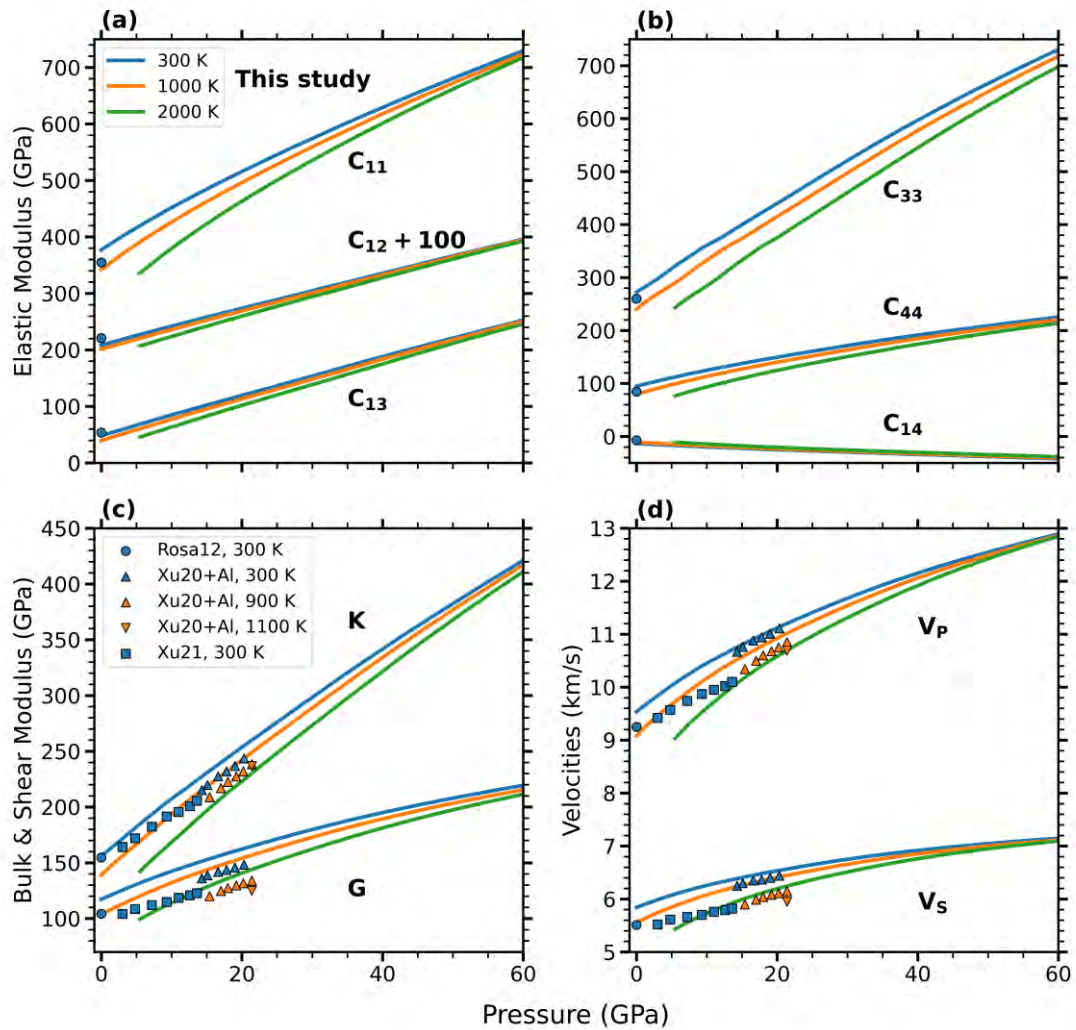
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467

468 **Figure 3.** (a) thermal expansion, (b) thermal Grüneisen parameter, (c) heat capacity at
469 constant volume, and (d) heat capacity at constant pressures of phase D. Solid lines
470 represent our calculation results at various pressures and dashed lines represent the
471 experimental results from HTBM EOS of phase D reported by Shinmei et al. (2008).

472



473

474 **Figure 4.** (a and b) elastic constants, (c) bulk modulus and shear modulus, (d)
475 compressional wave velocity, and shear wave velocity of phase D at various pressures
476 and temperatures. Solid lines represent our calculation results and the experimental
477 results are shown with scatters. Chemical formulas: This study, $\text{MgSi}_2\text{H}_2\text{O}_6$; Rosa12,
478 $\text{Mg}_{1.1}\text{Si}_{1.9}\text{H}_{2.4}\text{O}_6$ (Rosa et al., 2012); Xu20+Al, $\text{Mg}_{0.89}\text{Si}_{1.30}\text{Al}_{0.64}\text{H}_{3.10}\text{O}_6$ (Xu et al.,
479 2020); Xu21, $\text{Mg}_{1.03}\text{Si}_{1.71}\text{H}_{3.05}\text{O}_6$ (Xu et al., 2021b).

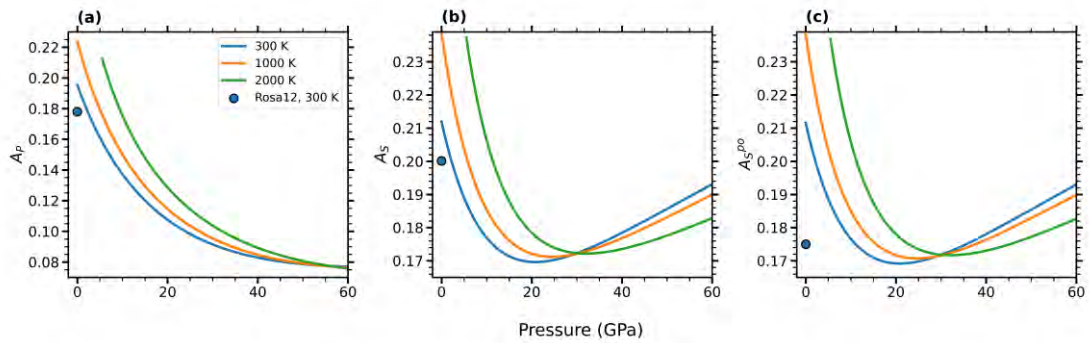
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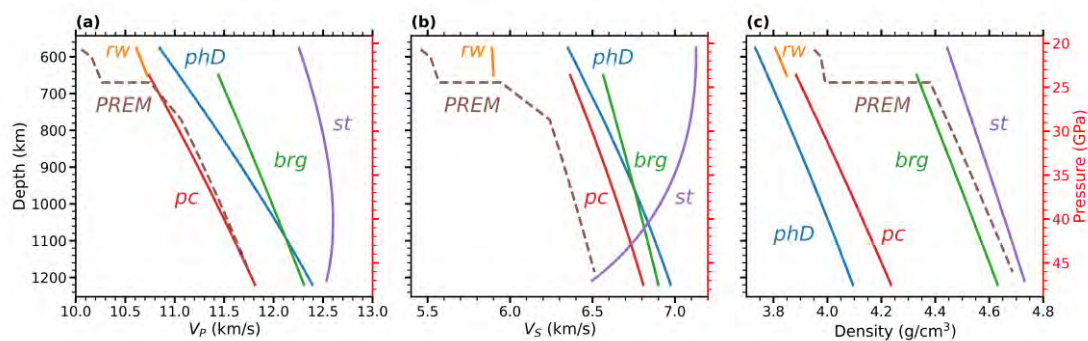
485

486 **Figure 5.** Anisotropy of phase D as a function of pressure. (a) A_p , (b) A_s , and (c) A_s^{p0}

487 of phase D at various pressures and temperatures. Blue circles represent the

488 experimental results at ambient conditions obtained by Rosa et al. (2012).

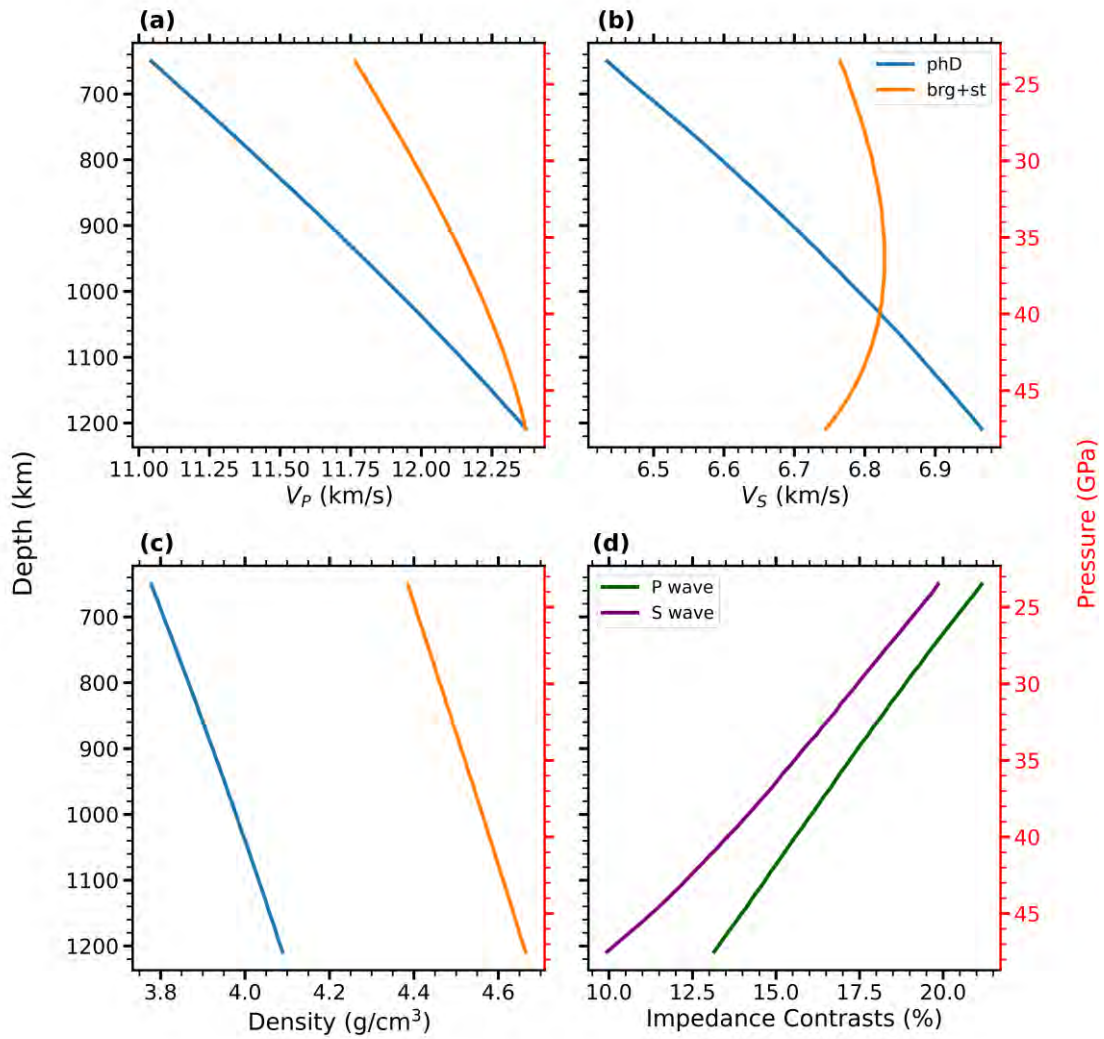
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490

491 **Figure 6.** (a) Compressional wave velocity V_p , (b) shear wave velocity V_s , and (c)
492 densities of phase D (phD) compared to those of iron-free ringwoodite (rw) (Núñez
493 Valdez et al., 2012), stishovite (st) (Yang and Wu, 2014), iron-free bridgmanite (brg)
494 (Shukla et al., 2015) and periclase (pc) (Wu and Wentzcovitch, 2011) along a slab
495 geotherm 500 K lower than the normal mantle geotherm (Brown and Shankland,
496 1981). The dashed lines show the profiles of the 1D seismic reference model PREM
497 (Dziewonski and Anderson, 1981).

498



499

500 **Figure 7.** (a) Compressional wave velocity V_P , (b) shear wave velocity V_S , and (c)
501 densities of phase D (phD) and the aggregate of bridgmanite (brg) (Shukla et al.,
502 2015) and stishovite (st) (Yang and Wu, 2014) along the cold isotherm. (d) Impedance
503 contrasts of P and S waves caused by the dehydration of phase D into bridgmanite and
504 stishovite.

505

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