

The fate of ammonium in phengite at high temperature

YAN YANG^{1,*}, VINCENT BUSIGNY², ZHONGPING WANG³, AND QUNKE XIA¹

¹Institute of Geology and Geophysics, School of Earth Sciences, Zhejiang University, Hangzhou 310027, China

²Institut de Physique du Globe de Paris, Sorbonne Paris Cité, University Paris Diderot, UMR 7154 CNRS, F-75005 Paris, France

³Physics Experiment Teaching Center, University of Science and Technology of China, Hefei 230026, China

ABSTRACT

Nitrogen (N) is the main component of the atmosphere and is largely considered as a volatile element. However, most researchers now agree that a significant amount of N, in the form of ammonium (NH_4^+) substituting for K^+ in some K-bearing minerals such as clays, micas, and feldspars, can be transferred to the deep Earth through subduction. The fate of ammonium in those minerals during subduction is still poorly known but is likely controlled by temperature and pressure pathways. In an attempt to contribute to understanding the fate of N during high-temperature processes, we carried out in situ high-temperature IR and Raman spectra measurements to investigate the rate and mechanism of NH_4^+ loss in phengite. We observed that a new OH band at 3425 cm^{-1} became prominent above $400\text{ }^\circ\text{C}$, and did not change with times during isothermal annealing at 500 and $700\text{ }^\circ\text{C}$. The N-H stretching band shifted to higher wavenumbers in the temperature interval from -150 to $20\text{ }^\circ\text{C}$, while linearly shifted to lower wavenumbers in the temperature interval from 20 to $500\text{ }^\circ\text{C}$ and remained stable above $500\text{ }^\circ\text{C}$. The N-H bending band linearly shifted to lower wavenumbers in the temperature interval from -150 to $400\text{ }^\circ\text{C}$ and remained stable. The K-O stretching frequency decreased with increasing temperature to $600\text{ }^\circ\text{C}$, and then remained stable. These processes were reversible until dehydration and ammonium loss from phengite starting at $800\text{ }^\circ\text{C}$. The results suggest that (1) at low temperatures, ammonium is ordered and hydrogen bonding between ammonium and the framework evolves during cooling; (2) at high temperatures, the N-H interatomic distance of NH_4^+ lengthens with increasing temperature until $500\text{ }^\circ\text{C}$. N-H bond subsequently no longer lengthens, accompanied by H transferring from N to neighboring O and forming a new OH band at 3425 cm^{-1} . At $800\text{ }^\circ\text{C}$, H^+ starts breaking from N and leaving others to form NH_3 and OH^- . This study has implications for evaluating the extent to which these minerals can preserve information regarding nitrogen behavior during high-temperature processes.

Keywords: Phengite, nitrogen, ammonium, high temperature, IR, Raman