

## **Jasonsmithite, a new phosphate mineral with a complex microporous framework, from the Foote mine, North Carolina, U.S.A.**

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### **ABSTRACT**

Jasonsmithite (IMA2019-121),  $\text{Mn}_4^{2+}\text{ZnAl}(\text{PO}_4)_4(\text{OH})(\text{H}_2\text{O})_7 \cdot 3.5\text{H}_2\text{O}$ , is a pegmatite-phosphate mineral from the Foote Lithium Company mine, Kings Mountain district, Cleveland County, North Carolina, U.S.A. It is interpreted as having formed by late-stage, low-temperature hydrothermal alteration. Crystals are colorless to light brown, slightly flattened prisms to about 1 mm in length with wedge-shaped terminations. The mineral is transparent with vitreous luster, white streak, Mohs hardness 2, brittle tenacity, irregular fracture, and perfect {001} cleavage. The density is 2.63(2) g/cm<sup>3</sup>. Jasonsmithite is biaxial (–), with  $\alpha = 1.561(2)$ ,  $\beta = 1.580(2)$ ,  $\gamma = 1.581(2)$ , measured in white light. The  $2V$  is 25(5)° and dispersion is  $r < v$  moderate. The optical orientation is  $Y = \mathbf{b}$ ,  $X \wedge \mathbf{c} = 18^\circ$  in obtuse  $\beta$ . The Raman spectrum is dominated by vibrational modes of  $\text{PO}_4$  and  $\text{ZnO}_4$  tetrahedra,  $\text{AlO}_6$  and  $\text{MnO}_6$  octahedra, and OH groups. Electron microprobe analyses gave the empirical formula  $(\text{Mn}_{3.09}\text{Fe}_{0.87})_{\Sigma 3.96}\text{Zn}_{1.05}\text{Al}_{0.98}(\text{PO}_4)_4(\text{OH})(\text{H}_2\text{O})_7 \cdot 3.5\text{H}_2\text{O}$ . The mineral is monoclinic,  $P2_1/c$ ,  $a = 8.5822(3)$ ,  $b = 13.1770(6)$ ,  $c = 20.3040(14)$  Å,  $\beta = 98.485(7)^\circ$ ,  $V = 2271.0(2)$  Å<sup>3</sup>, and  $Z = 4$ . The structure ( $R_1 = 0.0443$  for 3685  $I > 2\sigma I$  reflections) contains zigzag chains of edge-sharing  $\text{MnO}_6$  octahedra that corner-link to adjacent chains and to  $\text{PO}_4$  tetrahedra to form sheets, which are decorated by  $\text{ZnO}_4$  tetrahedra. The sheets are linked to one another via dimers of  $\text{AlO}_6$  octahedra, forming a framework with large channels containing  $\text{H}_2\text{O}$  groups. With  $\text{H}_2\text{O}$  groups removed, the framework has a void space of 70.2% per unit cell, and a framework density of 14.5 polyhedral atoms/1000 Å<sup>3</sup>, which would place jasonsmithite among the most porous minerals.

**Keywords:** Jasonsmithite, new mineral, phosphate, microporous framework structure, Raman spectroscopy, Foote mine, Kings Mountain, North Carolina, U.S.A.; Microporous Materials: Crystal-Chemistry, Properties, and Utilizations