

## Natural cubic perovskite, $\text{Ca}(\text{Ti,Si,Cr})\text{O}_{3-\delta}$ , a versatile potential host for rock-forming and less-common elements up to Earth's mantle pressure

SERGEY N. BRITVIN<sup>1,2,\*</sup>, NATALIA S. VLASENKO<sup>1</sup>, ANDREY ASLANDUKOV<sup>3</sup>, ALENA ASLANDUKOVA<sup>4</sup>,  
LEONID DUBROVINSKY<sup>5</sup>, LIUDMILA A. GORELOVA<sup>1</sup>, MARIA G. KRZHIZHANOVSKAYA<sup>1</sup>,  
OLEG S. VERESHCHAGIN<sup>1,†</sup>, VLADIMIR N. BOCHAROV<sup>1</sup>, YULIA S. SHELUKHINA<sup>1</sup>, MAKSIM S. LOZHKIN<sup>1</sup>,  
ANATOLY N. ZAITSEV<sup>1</sup>, AND FABRIZIO NESTOLA<sup>6,‡</sup>

<sup>1</sup>Saint-Petersburg State University, Universitetskaya Nab. 7/9, 199034 St. Petersburg, Russia

<sup>2</sup>Kola Science Center of Russian Academy of Sciences, Fersman Str. 14, 184209 Apatity, Russia

<sup>3</sup>Material Physics and Technology at Extreme Conditions, Laboratory of Crystallography, University of Bayreuth, 95440 Bayreuth, Germany

<sup>4</sup>Bavarian Research Institute of Experimental Geochemistry and Geophysics, University of Bayreuth, 95440 Bayreuth, Germany

<sup>5</sup>Bayerisches Geoinstitut, University of Bayreuth, Universitätsstraße 30, 95447 Bayreuth, Germany

<sup>6</sup>Dipartimento di Geoscienze, Università degli Studi di Padova, Via G. Gradenigo 6, I-35131 Padova, 21 Italy

### ABSTRACT

Perovskite,  $\text{CaTiO}_3$ , originally described as a cubic mineral, is known to have a distorted (orthorhombic) crystal structure. We herein report on the discovery of natural cubic perovskite. This was identified in gehlenite-bearing rocks occurring in a pyrometamorphic complex of the Hatrurim Formation (the Mottled Zone), in the vicinity of the Dead Sea, Negev Desert, Israel. The mineral is associated with native  $\alpha$ -(Fe,Ni) metal, schreibersite ( $\text{Fe}_3\text{P}$ ), and Si-rich fluorapatite. The crystals of this perovskite reach 50  $\mu\text{m}$  in size and contain many micrometer-sized inclusions of melilitic glass. The mineral contains significant amounts of Si substituting for Ti (up to 9.6 wt%  $\text{SiO}_2$ ), corresponding to 21 mol% of the davemaoite component (cubic perovskite-type  $\text{CaSiO}_3$ ), in addition to up to 6.6 wt%  $\text{Cr}_2\text{O}_3$ . Incorporation of trivalent elements results in the occurrence of oxygen vacancies in the crystal structure; this is the first example of natural oxygen-vacant  $\text{ABO}_3$  perovskite with the chemical formula  $\text{Ca}(\text{Ti,Si,Cr})\text{O}_{3-\delta}$  ( $\delta \sim 0.1$ ). Stabilization of cubic symmetry (space group  $Pm\bar{3}m$ ) is achieved via the mechanism not reported so far for  $\text{CaTiO}_3$ , namely displacement of an O atom from its ideal structural position (site splitting). The mineral is stable at atmospheric pressure to  $1250 \pm 50$  °C; above this temperature, its crystals fuse with the embedded melilitic glass, yielding a mixture of titanite and anorthite upon melt solidification. The mineral is stable upon compression to at least 50 GPa. The  $a$  lattice parameter exhibits continuous contraction from 3.808(1) Å at atmospheric pressure to 3.551(6) Å at 50 GPa. The second-order truncation of the Birch-Murnaghan equation of state gives the initial volume  $V_0$  equal to 55.5(2) Å<sup>3</sup> and room temperature isothermal bulk modulus  $K_0$  of 153(11) GPa. The discovery of oxygen-deficient single perovskite suggests previously unaccounted ways for incorporation of almost any element into the perovskite framework up to pressures corresponding to those of the Earth's mantle.

**Keywords:** Cubic perovskite, site splitting, disordered oxygen vacancies, davemaoite, mantle, high pressure, pyrometamorphism, Dead Sea Transform