

Effects of electronegativities and charge delocalization on Q² Raman shifts of alkaline- and alkaline earth-bearing glasses and metasilicate crystals

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ABSTRACT

Raman shifts of the symmetric stretch of silicate Q² species vary over a range of ~90 cm⁻¹ in crystals and glasses containing alkali and alkaline earth oxides. The shifts display a striking, sympathetic relationship with the electronegativity of the alkali and alkaline earth metals (M), with the highest frequency observed for Mg-silicate glasses and crystals and the lowest frequency for Cs-bearing glasses. Frequencies are determined primarily by the electron density on constituent Si and O atoms of the Q² tetrahedra, as measured by Si 2p and O 1s X-ray photoelectron spectra (XPS). The electron density is, in turn, determined by the extent to which electronic charge is transferred from the modifier metal “M” to the NBO of the Q² tetrahedron. The charge transferred to NBO is redistributed (delocalized) over all atoms of the tetrahedron by the four equivalent Si sp³ orbitals. Although negative charge accumulates on all atoms of the tetrahedron, it accumulates preferentially on Si. Coulombic interactions among Si and all O atoms are thus weakened, resulting in decreased force constants and lowered symmetric stretch frequencies of Q² species.

Density functional theoretical (DFT) calculations on six staggered and eclipsed M₆Si₂O₇ (M = Li, Na, K) molecules corroborate the findings. Charge is transferred from the metal atoms to NBO and delocalized over tetrahedra in accordance with Li, Na, and K electronegativities. Calculated Si-O force constants and Raman shifts decrease with decreasing electronegativity of the cation but surprisingly, calculated Si-NBO bond lengths are largely unaffected, with all being similar at 1.665 ± 0.003 Å.

Keywords: Raman spectroscopy, electronegativity charge delocalization in silicates, tetrahedral symmetric stretch, silicate glasses