

## Crystal structure of nyerereite: A possible messenger from the deep Earth

AZZURRA ZUCCHINI<sup>1,\*</sup>, PAVEL N. GAVRYUSHKIN<sup>2,3</sup>, ALEXANDER V. GOLOVIN<sup>2</sup>,  
NADEZHDA B. BOLOTINA<sup>4</sup>, PAOLA STABILE<sup>5,†</sup>, MICHAEL R. CARROLL<sup>5</sup>, PAOLA COMODI<sup>1</sup>,  
FRANCESCO FRONDINI<sup>1</sup>, DANIELE MORGAVI<sup>6</sup>, DIEGO PERUGINI<sup>1</sup>, FABIO ARZILLI<sup>5,7</sup>, MARCO CHERIN<sup>1</sup>,  
EMMANUEL KAZIMOTO<sup>8</sup>, KONSTANTIN KOKH<sup>2,3,9</sup>, ARTEM KUZNETSOV<sup>2</sup>, AND INNA V. MEDRISH<sup>10</sup>

<sup>1</sup>Department of Physics and Geology, University of Perugia, 06123 Perugia, Italy

<sup>2</sup>Sobolev Institute of Geology and Mineralogy, Siberian Branch of Russian Academy of Sciences, 630090 Novosibirsk, Russia

<sup>3</sup>Novosibirsk State University, Novosibirsk 630090, Russia

<sup>4</sup>Shubnikov Institute of Crystallography, Federal Scientific Research Centre ‘Crystallography and Photonics’ of Russian Academy of Sciences, Leninskii prosp. 59, 119333 Moscow, Russia

<sup>5</sup>School of Science and Technology, Geology Division, University of Camerino, 62032 Camerino, Italy

<sup>6</sup>Department of Earth, Environmental and Resource Sciences (DiSTAR), University of Naples Federico II, Via Vicinale Cupa Cintia 21, 80126 Naples, Italy

<sup>7</sup>School of Earth and Environmental Sciences, University of Manchester, M139PL, Manchester, U.K.

<sup>8</sup>Department of Geosciences, School of Mines and Geosciences, University of Dar es Salaam 16103, Uvumbuzi Rd, Dar es Salaam Tanzania

<sup>9</sup>Kemerovo State University, 6 Krasnaya Str., Kemerovo, 650000, Russia

<sup>10</sup>Samara Center for Theoretical Material Science (SCTMS), Samara State Technical University, Molodogvardeyskaya St. 244, Samara, 443100, Russia

### ABSTRACT

Carbonates in the system  $\text{Na}_2\text{CO}_3\text{-CaCO}_3$  are nowadays suggested as having a wide stability field at conditions of the mantle transition zone. Our structural analysis of nyerereite, which has limited stability fields at ambient conditions, and its similarities with already known carbonates that are stable at high-pressure conditions, allowed us to propose that nyerereite likely undergoes phase transitions at both high-pressure and high-temperature conditions. This supports the hypothesis that nyerereite takes part in carbon transportation from the mantle/deep crust toward the surface, with important implications for the deep carbon cycle associated with carbonatites.

K-free nyerereite  $[\text{Na}_2\text{Ca}(\text{CO}_3)_2]$  was synthesized both at hydrothermal conditions and from the melt. The structure of nyerereite was refined as a three-component twinned structure in the centrosymmetric *Pbca* space group with ratios of the three twinning components 0.221(3):0.287(3):0.492(3). Twinning at micro- and nano-level can introduce some minor structural deformations that influence the likely occurrence of the inversion center as one of the symmetry elements in the nyerereite structure. Based on the automated topological algorithms, we show that nyerereite has a unique crystal structure, not having analogs among the known structures, except for the structure with a similar composition  $\text{K}_2\text{Ca}(\text{CO}_3)_2$  fairchildite.

A comparison between the centrosymmetric *Pbca* nyerereite structure and that of aragonite ( $\text{CaCO}_3$ , *Pmcn* space group) reveals two main scenarios for the high-pressure form of  $\text{Na}_2\text{Ca}(\text{CO}_3)_2$ : (1) poly-somatic relations as the interlayering of the high-pressure polymorph  $\text{Na}_2\text{CO}_3$  and  $\text{CaCO}_3$ -aragonite, and (2) high-pressure structure with ninefold-coordinated Na and Ca sites resembling that of aragonite. Our discussion heightens the interest in the high-pressure behavior of the nyerereite structure and strengthens the hypothesis about the possibility for nyerereite to be stable at high-pressure/high-temperature conditions.

**Keywords:** Nyerereite, single-crystal X-ray diffraction, hydrothermal synthesis, melt crystallization, Raman spectroscopy, alkali-carbonates