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Correlations between cathodoluminescence intensity and aluminum concentration in low-temperature hydrothermal quartz

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ABSTRACT

Quartz cathodoluminescence (CL) images are commonly combined with trace element concentrations to decipher complex histories of hydrothermal systems. However, the correlations between aluminum content and CL zoning of low-temperature hydrothermal quartz and their genesis remain controversial. In this contribution, a multiparametric study was carried out on CL-aluminum zoning of low-temperature hydrothermal quartz (<350 °C) from the Shihu and Rushan quartz-vein type Au deposits in the North China Craton. The results show that aluminum concentration correlates negatively with CL intensity in quartz from the Shihu Au deposit. CL-dark quartz zoning has significant Al concentrations as well as detectable Al-H bonds. However, in the Rushan Au deposit, the correlation is positive, and aluminum is enriched in the CL-bright quartz zoning. The Al content is positively correlated with K content with $r^2 = 0.769$. Combined with the electron backscatter diffraction (EBSD), X-ray single crystal diffraction (XRD), and transmission electron microscope (TEM) data, we infer that the genesis of CL zoning in the low-temperature hydrothermal quartz is closely related to $\text{Al}^{3+}\text{-H}^+$ and $\text{Al}^{3+}\text{-K}^+$ concentrations. The $\text{Al}^{3+}\text{-K}^+$ may act as the CL-activator, while the $\text{Al}^{3+}\text{-H}^+$ may act as the CL-dampener. Where $\text{Al}^{3+}\text{-Si}^{4+}$ substitution is charge balanced by hydrogen, the intensity of CL response decreases; where $\text{Al}^{3+}\text{-Si}^{4+}$ substitution is charge balanced by potassium, the intensity of CL response increases. The correlations between CL intensity and aluminum concentration in the low-temperature hydrothermal quartz reflect pH fluctuations of hydrothermal system.

Keywords: Shihu and Rushan Au deposits, low-temperature hydrothermal quartz, cathodoluminescence, $\text{Al}^{3+}\text{-H}^+$ and $\text{Al}^{3+}\text{-K}^+$ concentrations, pH