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K isotopic fractionation in K-feldspar: Effects of mineral chemistry

SHANKE LIU^{1,*}, WENJUN LI¹, BENXUN SU^{1,2}, QIQI PAN^{2,3}, MENG YUAN^{2,3}, AND PATRICK ASAMOAH SAKYI⁴

¹Key Laboratory of Mineral Resources, Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing 100029, China ²University of Chinese Academy of Sciences, Beijing 100049, China

³State Key Laboratory of Lithospheric Evolution, Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing 100029, China ⁴Department of Earth Science, College of Basic and Applied Sciences, University of Ghana, P.O. Box LG 58, Legon-Accra, Ghana

ABSTRACT

Controlling factors of potassium (K) isotopic fractionation in K-feldspar remain poorly constrained. In this study, we analyzed the K isotopic compositions of 11 K-feldspar samples from diverse lithological compositions. The degree of Al/Si order ranged from 0.22 to 0.94 (1.0 = completely ordered). Analyzed samples are mixtures of K-feldspar (>70 wt%) and coexisting albite. The relative contribution of K₂O from the K-feldspar phase of the sample was over 98%, indicating that the K isotopic composition (δ^{41} K) derives mainly from K-feldspar and hence reflects its behavior. The δ^{41} K values of these samples range from -0.710 to -0.075‰, which are slightly correlated with the degree of Al/Si order. The correlations of δ^{41} K with SiO₂ and Al₂O₃ contents and the corresponding Al/Si mole ratios reveal that Al and Si play a significant role in the K isotopic behavior of K-feldspar. The correlations of δ^{41} K with SiO₂ and Al₂O₃ contents and the corresponding Al/Si mole ratios reveal that Al and Si play a significant role in the K isotopic behavior of K-feldspar. The correlations of δ^{41} K with SiO₂ and Al₂O₃ contents are attributed to the difference in K-O bond strengths. Compared to K-feldspar, the K content could be a better proxy for constraining the δ^{41} K of plagioclase. Our results demonstrate that the δ^{41} K of K-feldspar is dependent on its mineral chemistry, and its K isotopic composition may be insensitive to other factors, such as the source heterogeneity. The inference is further confirmed by comparing the δ^{41} K values of K-feldspar from different sources.

Keywords: K isotopes, K-feldspar, Al/Si order, bond length, Al/Si mole ratio