

## **Sublattice disorder and Fe-Mg substitution in brucite: Implications for the subduction-zone water cycle**

**WEIBIN GUI<sup>1,†</sup>, JIN LIU<sup>2,\*</sup>, AO DENG<sup>2</sup>, JUN HU<sup>2</sup>, PENGHUI SUN<sup>2</sup>, CHANGZENG FAN<sup>2</sup>, AND FAHUI XIONG<sup>1,\*</sup>**

<sup>1</sup>Key Laboratory of Continental Dynamics of Ministry of Natural Resources, State Key Laboratory of Deep Earth and Mineral Exploration, Institute of Geology, Chinese Academy of Geological Sciences, Beijing 100037, China

<sup>2</sup>State Key Laboratory of Metastable Materials Science and Technology, Yanshan University, Qinhuangdao 066004, China

### **ABSTRACT**

The vibrational properties of hydroxyl groups in pure and Fe-bearing brucite have been investigated under high-pressure and -temperature conditions relevant to subduction zones. Using externally heated diamond-anvil cell techniques combined with in situ Raman spectroscopy, we observed that the hydroxyl stretching modes in pure brucite change from blueshift to redshift at  $\sim 600$  K and 1.2 GPa. This indicates a sublattice proton order-disorder transition associated with decreasing interlayer spacing in brucite. Moreover, the Fe-Mg substitution in brucite leads to a secondary hydroxyl stretching band at  $3644.6\text{ cm}^{-1}$  for  $\text{Mg}_{0.9}\text{Fe}_{0.1}(\text{OH})_2$ , while the primary Raman band is changed to  $3636.7\text{ cm}^{-1}$ . This new Raman mode persists with increasing temperature at atmospheric pressure but disappears at about 3.0 GPa. The decomposition temperature of this Fe-bearing brucite sample is about 130 K lower compared to pure brucite under high oxygen fugacity conditions. These results suggest that the sublattice transition and Fe-Mg substitution may influence the stability of brucite, providing insight into the deep-water cycle at shallow subduction zones.

**Keywords:** Brucite, sublattice disorder, Fe-Mg substitution, subduction zone, oxygen fugacity