

## “Cs-tetra-ferri-annite:” High-pressure and high-temperature behavior of a potential nuclear waste disposal phase

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### ABSTRACT

Structure deformations induced by pressure and temperature in synthetic “Cs-tetra-ferri-annite” 1M [Cs<sub>1.78</sub>(Fe<sup>2+</sup><sub>5.93</sub>Fe<sup>3+</sup><sub>0.07</sub>)(Si<sub>6.15</sub>Fe<sup>3+</sup><sub>1.80</sub>Al<sub>0.05</sub>)O<sub>20</sub>(OH)<sub>4</sub>], space group *C2/m*, were analyzed to investigate the capability of the mica structure to store the radiogenic isotopes <sup>135</sup>Cs and <sup>137</sup>Cs. “Cs-tetra-ferri-annite” is not a mineral name, but for the sake of brevity is used here to designate a synthetic analog of the mineral tetra-ferri-annite. The bulk modulus and its pressure derivative determined by fitting the unit-cell volumes between 0 and 47 kbar to a third-order Birch-Murnaghan equation of state are  $K_0 = 257(8)$  kbar and  $K_0' = 21(1)$ , respectively. Between 23 °C and 582 °C, the *a* and *b* lattice parameters remain essentially unchanged, but the thermal expansion coefficient of the *c* axis is  $\alpha_c = 3.12(9) \times 10^{-5}$  °C<sup>-1</sup>. High pressure (*P*) and high temperature (*T*) produce limited internal strain in the structure. The tetrahedral rotation angle,  $\alpha$ , is very small and does not change significantly throughout the *P* and *T* range investigated. Above 450 °C in air, “Cs-tetra-ferri-annite” underwent an oxidation of octahedral iron in the M2*cis* site, balanced by the loss of H and shown by a decrease of the unit-cell volume.

Independent isobaric data on thermal expansion and isothermal compressibility data define the “geometric” equation of state for “Cs-tetra-ferri-annite”:  $V/V_0 = 1 + 3.0(1) 10^{-5} T - 2.68(9) 10^{-3} P + 2.0(2) \times 10^{-5} P^2$  where *T* is in degrees Celsius, *P* is in kilobars. The  $\alpha/\beta$  ratio of about 12 bar/°C indicate that the cell volume of “Cs-tetra-ferri-annite” remains unchanged under geothermal gradients of ~23 °C/km. On the whole, the data confirm that the structure of “Cs-tetra-ferri-annite” may be a suitable candidate for the storage of large ions, such as Cs in the interlayer and should be considered as a potential Synroc component.