## HIGHLIGHTS AND BREAKTHROUGHS

## Seeking the most hydrous, primitive arc melts: The glass is half full

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Experimental studies and petrologic constraints suggest that H<sub>2</sub>O contents of deep, primitive melts in subduction settings may reach up to >15 wt% H<sub>2</sub>O (e.g., Krawczynski et al. 2012). But curiously, mafic glasses preserved in melt inclusions-commonly the best available tool to analyze H2O contents of melts-seem to be limited to much lower values, mostly <6 wt% (Plank et al. 2013). This apparent conundrum suggests that empirical results defy predictions and challenges our view of H2O in subductionrelated magmatism. To address this issue, Gavrilenko et al. (2019) experimentally tested the quenching behavior of hydrous, mafic melts. Their results demonstrate that quenching to glass becomes difficult at high H<sub>2</sub>O concentrations and that mafic melts exceeding ~9 wt% H<sub>2</sub>O are essentially unquenchable at realistic cooling rates. This implies that glasses preserved in melt inclusions provide only a partial record of the volatile contents of deep-seated melts and are incapable of recording the deepest, most hydrous melts. This work thus elegantly reconciles what previously appeared to be a stark contradiction between prediction and observation, and adds a key piece to our evolving understanding of how to analyze and interpret melt inclusions.

The H<sub>2</sub>O concentrations of melts exert a strong control on properties such as buoyancy (Ochs and Lange 1999), viscosity (Schulze et al. 1996), chemical diffusivity (Watson 1994), and explosivity (Sparks 1978), as well as the ore-forming potential of arc magmas (Hedenquist and Lowenstern 1994). The H<sub>2</sub>O contents of arc magmas are also central to quantifying and interpreting global geochemical cycling between Earth's surface and deep interior (Bodnar et al. 2013). Moreover, H<sub>2</sub>O contents of melts are widely used to evaluate depths of magmatic plumbing systems, based on the thermodynamic relationship between pressure and solubility of volatiles (Audétat and Lowenstern 2014, and references therein). However, the H<sub>2</sub>O contents of pre-eruptive melts are also elusive parameters. Experimentally calibrated proxies have been developed to estimate H2O contents of melts based on mineral equilibria (e.g., Krawczynski et al. 2012), but commonly, the only available tool to directly quantify the H<sub>2</sub>O (and other volatile) contents of pre-eruptive melts is by analysis of melt inclusions (Audétat and Lowenstern 2014).

In recent years, a growing body of theoretical, experimental, and analytical studies has contributed new insights into the systematics of volatiles in melt inclusions and how to best analyze and interpret them. It is now widely recognized that bubbles within melt inclusions can host a preponderance of the bulk  $CO_2$  (Moore et al. 2015) and  $H_2O$  (Esposito et al. 2016), and that  $H_2O$  concentrations can be rapidly modified by diffusive re-equilibration (Portnyagin et al. 2008; Gaetani et al. 2012). Careful attention to these phenomena has helped elucidate the record of pre-eruptive volatiles and degassing. Yet even in light of these developments, still the growing body of analytical data presents some enigmatic results.

One of the crucial and fundamental questions that has confounded our view of volatiles in subduction-related melt inclusions arises from the growing recognition that H<sub>2</sub>O (as well as CO<sub>2</sub>) contents of glasses preserved in melt inclusions seem to show an unexpectedly restricted range. Specifically, mafic glasses in melt inclusions from arc settings seem to be limited to H<sub>2</sub>O contents mostly less than ~6 wt% and never exceeding ~9 wt% (Plank et al. 2013). In contrast, experimental phase equilibria consistently predict much higher H<sub>2</sub>O contents, up to >15 wt% (Krawczynski et al. 2012). This apparent contradiction fundamentally challenges our view of either the fidelity of melt inclusions, or how well our experiments reproduce nature, or both.

Although some H<sub>2</sub>O is likely partitioned into bubbles (Esposito et al. 2016), such partitioning is unlikely to have such a dramatic effect on the measured H<sub>2</sub>O concentration in the glass (Steele-MacInnis et al. 2011). Diffusive re-equilibration also likely plays a role in reducing water contents in melt inclusions (Portnyagin et al. 2008; Gaetani et al. 2012). But neither process is expected to yield such a consistent threshold of H<sub>2</sub>O across the breadth of thousands of reported analyses, which is moreover so far below experimental predictions. What then limits melt inclusion H<sub>2</sub>O contents? Could it be that magmas related to subduction have only half the amount of water implied by experimental studies?

On page 936 of the July issue, Gavrilenko et al. (2019) test an alternative hypothesis that the upper limit of H<sub>2</sub>O contents of glasses preserved in melt inclusions reflects a quench control. Specifically, Gavrilenko et al. (2019) hypothesize that wetter melts are more difficult to quench, and that the wettest melts simply cannot be quenched. This hypothesis is rooted in the well-known relationships between H2O concentration, viscosity, and the glass transition (Mysen and Richet 2005): wetter melts are less viscous, and less viscous melts are less easily quenched, requiring either greater degrees of undercooling or faster cooling rates to be quenched as glass. Gavrilenko et al. (2019) test this hypothesis by conducting rapid-quench experiments on mafic melts over a wide range of H<sub>2</sub>O contents. Importantly, the cooling rates achieved in their experiments (20-90 K/s) are consistent with best estimates for cooling rates during the eruption (maximum ~22 K/s; Lloyd et al. 2013). The results are remarkable. Melts that contain modest H2O concentrations up to ~6 wt% consistently quench to form optically clear glass. Melts containing from ~6 to ~9 wt% H2O are somewhat difficult to quench, and consistently form crystallites in addition to glass. Melts exceeding 9 wt% H2O do not quench to glass and instead

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form friable aggregates of crystallites, vapor bubbles, and material resembling devitrified glass. And compellingly, the limiting values of H<sub>2</sub>O concentrations align perfectly with the empirical results from melt inclusions.

The results by Gavrilenko et al. (2019) strongly indicate that an apparent upper limit on the H<sub>2</sub>O contents of mafic melts is a consequence of the inability of wetter melts to form glassy inclusions. Recently, Maclennan (2017) used numerical modeling to investigate an apparent upper limit of CO<sub>2</sub> contents of melt inclusions from low-H2O settings, and concluded that high CO2 concentrations (resulting from high trapping pressures) give rise to intense overpressure and inevitable decrepitation during magma ascent. These two studies, Maclennan (2017) and Gavrilenko et al. (2019), indicate complementary phenomena that control and restrict the observed ranges of both H<sub>2</sub>O and CO2 in melt inclusions. Similarly, Esposito et al. (2016) argued that H<sub>2</sub>O exsolved into vapor bubbles in melt inclusions rapidly reacts with the surrounding glass causing devitrification, which would further obscure bulk volatile concentrations. It seems likely that the quench control could work in tandem with exsolution of vapor, devitrification, overpressure, and decrepitation, as well as diffusive re-equilibration, all conspiring to prevent the preservation of the most volatile-rich glasses in melt inclusions.

The major implication of the results by Gavrilenko et al. (2019) is that wetter melts are unlikely to form glassy melt inclusions, which in turn implies that glassy inclusions are incapable of preserving a complete record of the deepest, wettest melts. On the one hand, this is a sobering message. But on the other hand, this work illuminates a fundamental control, and by understanding this phenomenon, we stand a better chance of obtaining robust interpretations of water contents of melts. This conclusion underscores the need to analyze melt inclusions in the context of robust petrographic and geochemical constraints on the timing of trapping and post-entrapment processes (Esposito et al. 2014) and provides the impetus for development and application of complementary techniques that are not reliant on glassy inclusions. In any case, previous empirical observations showing a restricted range of H2O contents can be viewed as a natural consequence of "quenchability." This result neatly reconciles the observations with experimental predictions, without undermining the fidelity of glassy melt inclusions or predicted H<sub>2</sub>O contents of arc magmas—instead, indicating that the elusive, H<sub>2</sub>O-rich primitive arc melts may simply be missing from the record of glassy melt inclusions. Interestingly, this does not necessarily preclude the possibility of more H2O-rich, non-glassy melt inclusions, which would appear as being partially crystallized or devitrified. Such inclusions may be easily overlooked or discarded, but future workers seeking wet, primitive melts may be wise to search for and specifically target such inclusions.

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